

# Engineering tolerance in yeast for cellulosic biofuels



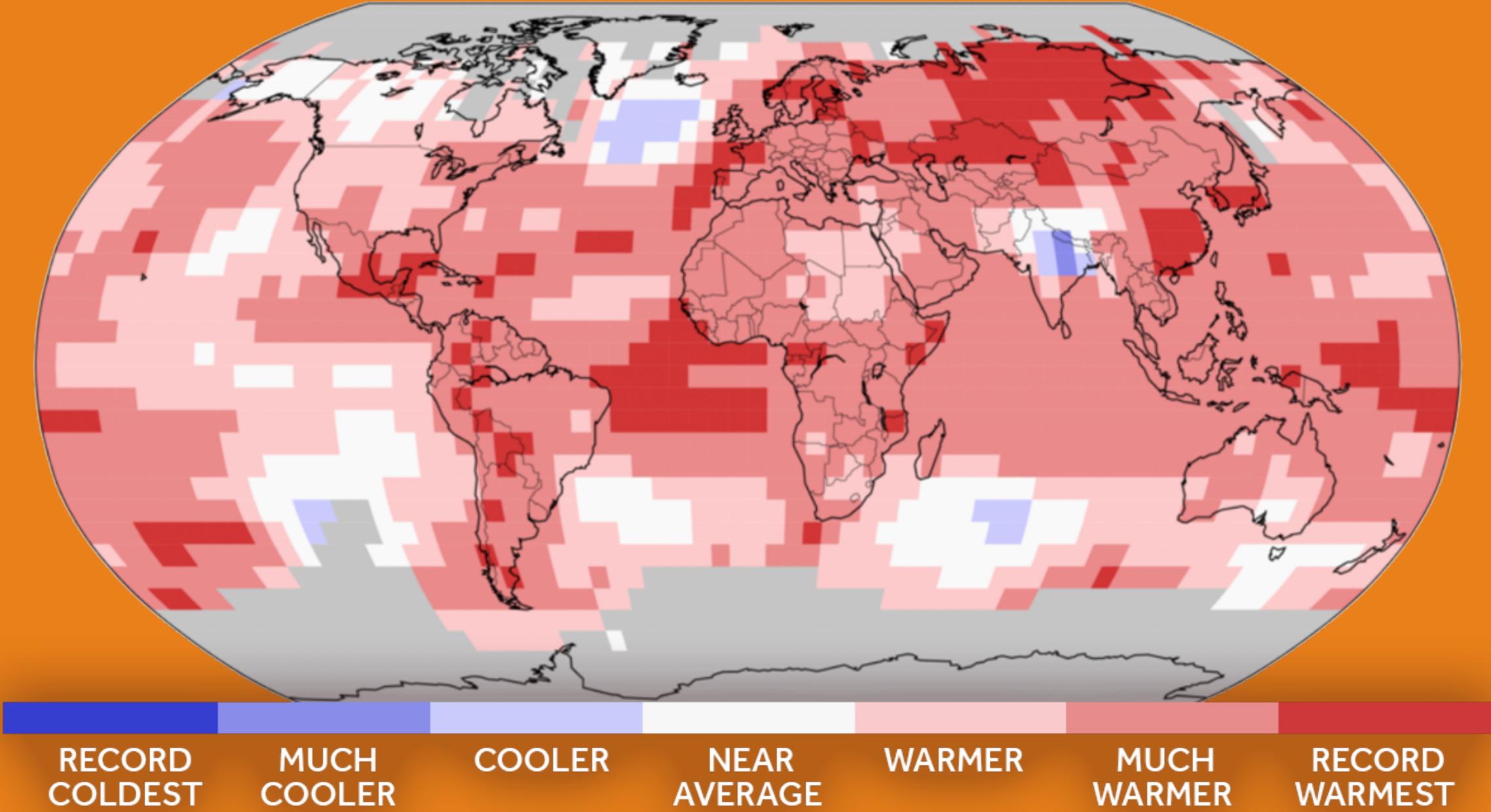
Seminar Series for  
High School  
Teachers

Fall 2021

**Felix Lam**  
Whitehead Institute  
MIT Chemical Engineering

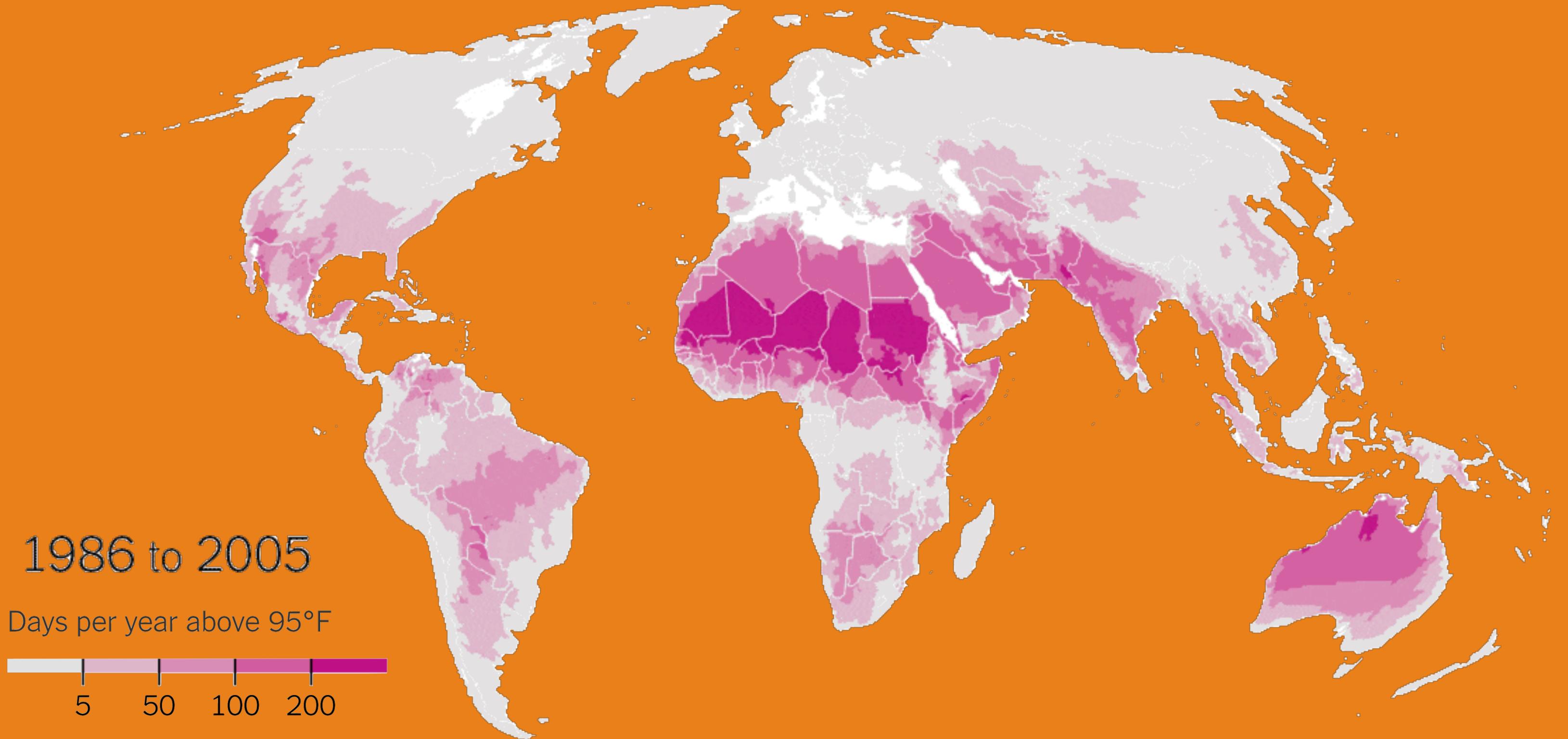


# Global temperatures continue setting records

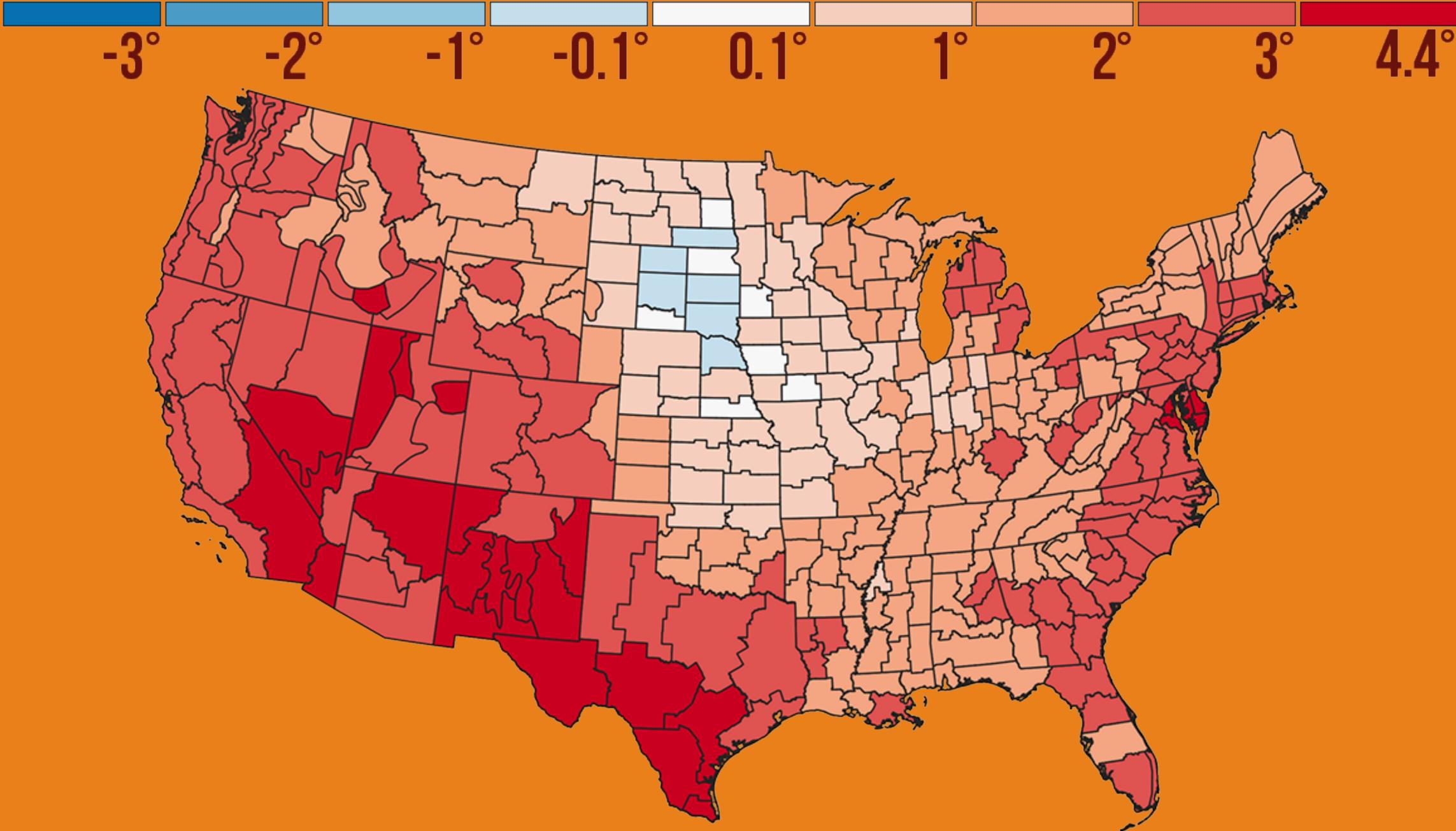


Jan – June 2020 Temperatures

# Global temperatures continue setting records

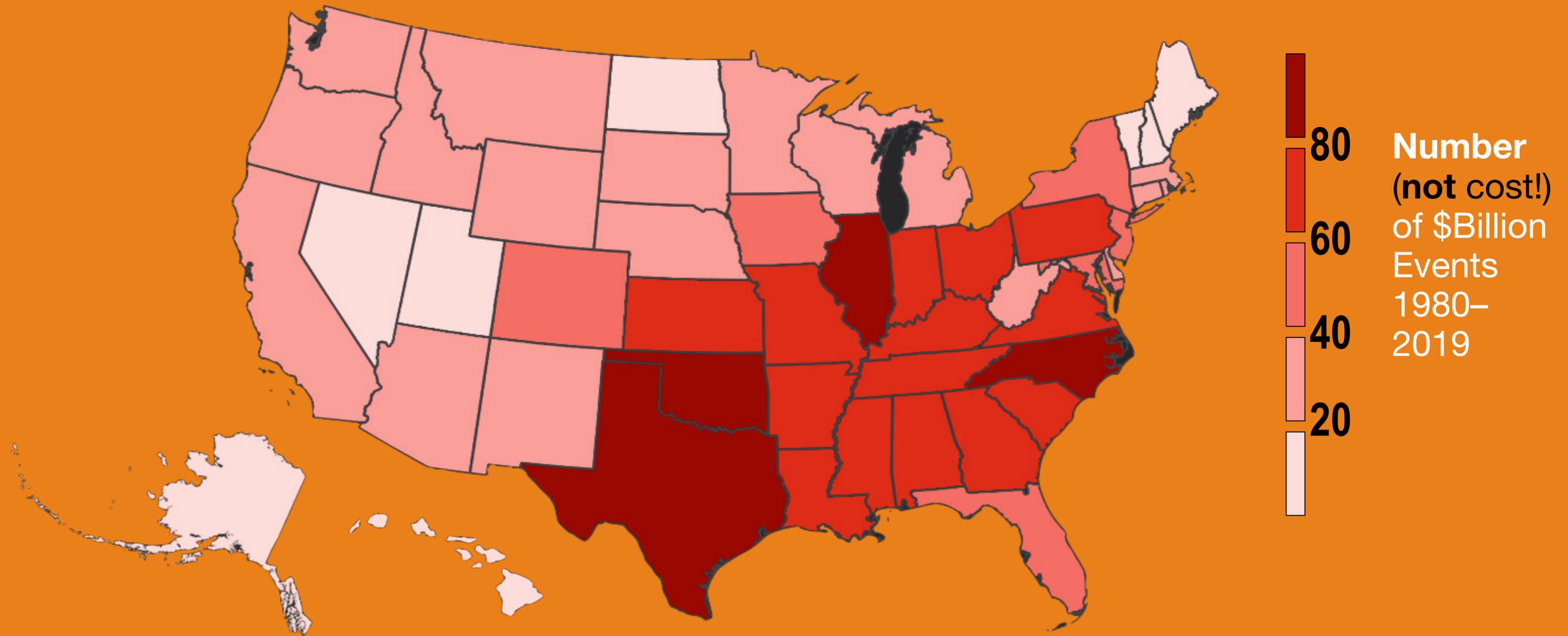


# Global temperatures continue setting records



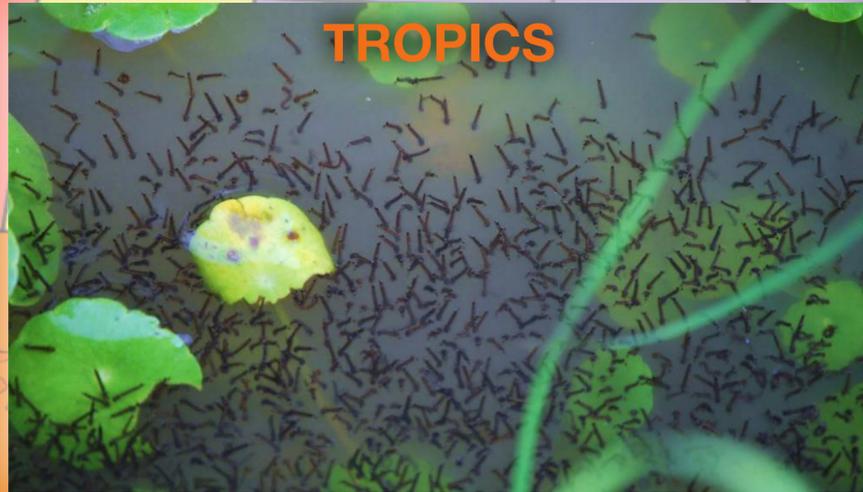
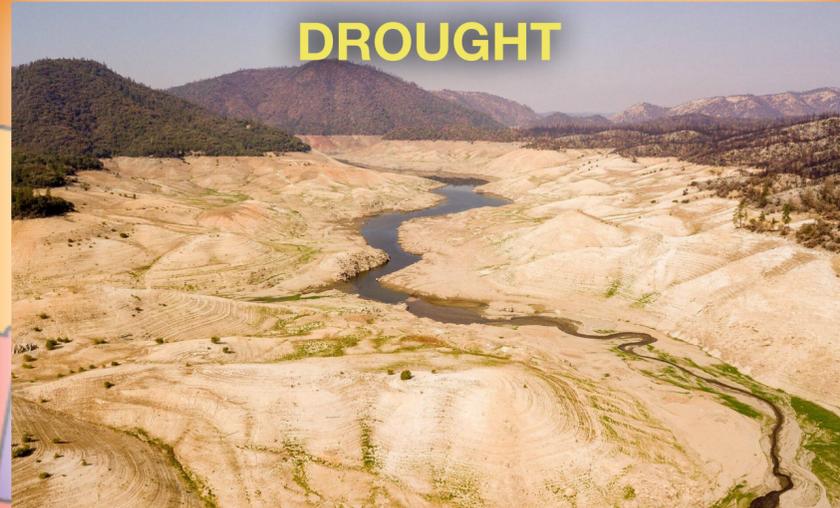
Summer Warmings Since 1970 (°F)

# Related \$billion disasters continue setting records



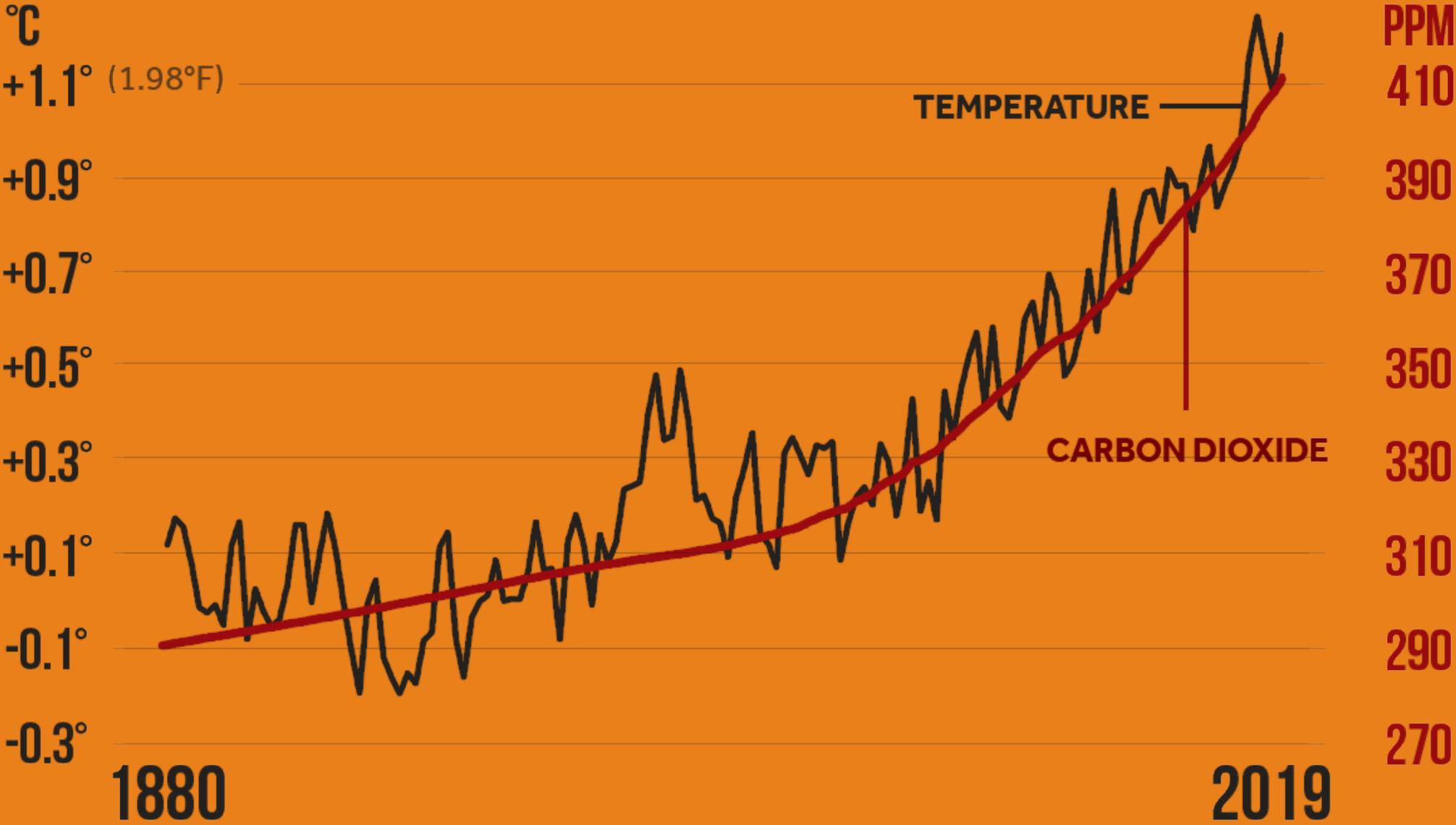


# Related \$billion disasters continue setting records

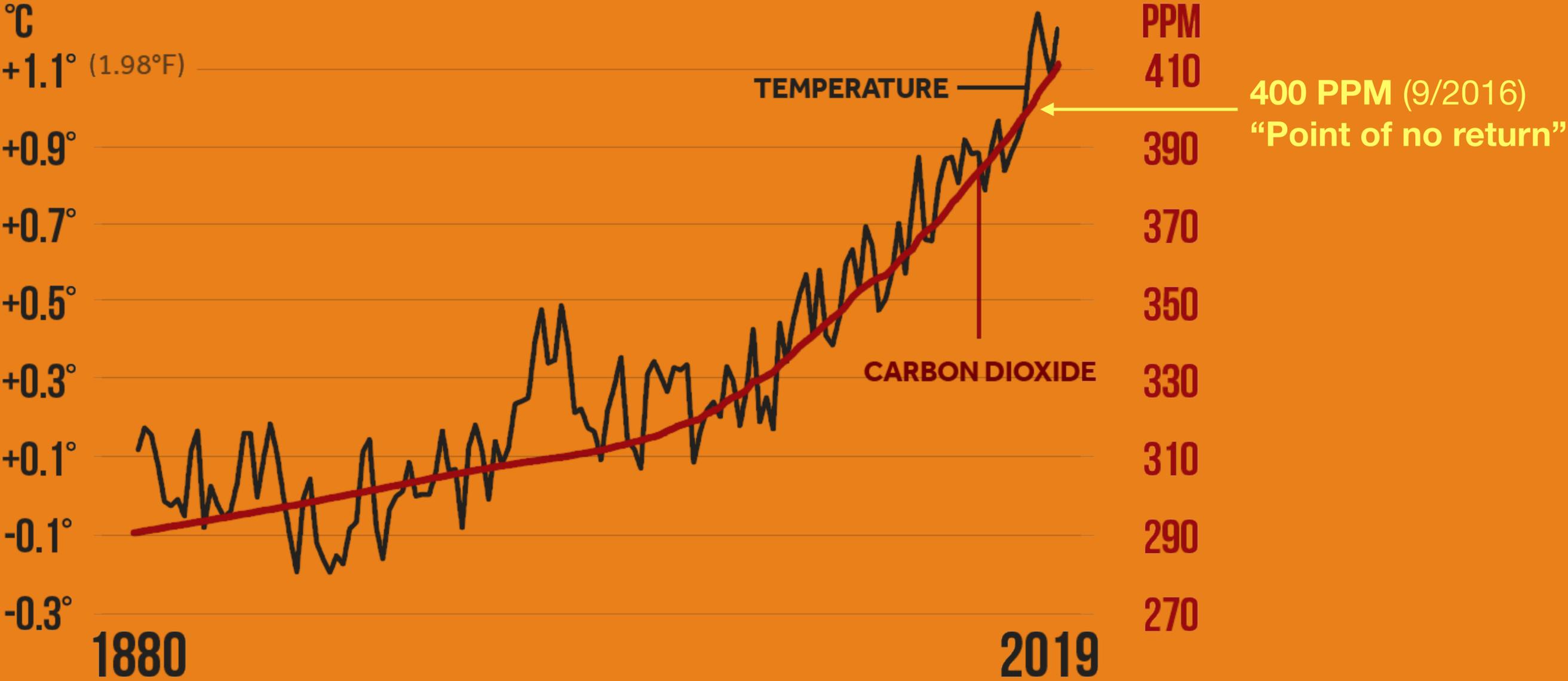


Most Frequent Disaster Type 1980–2019

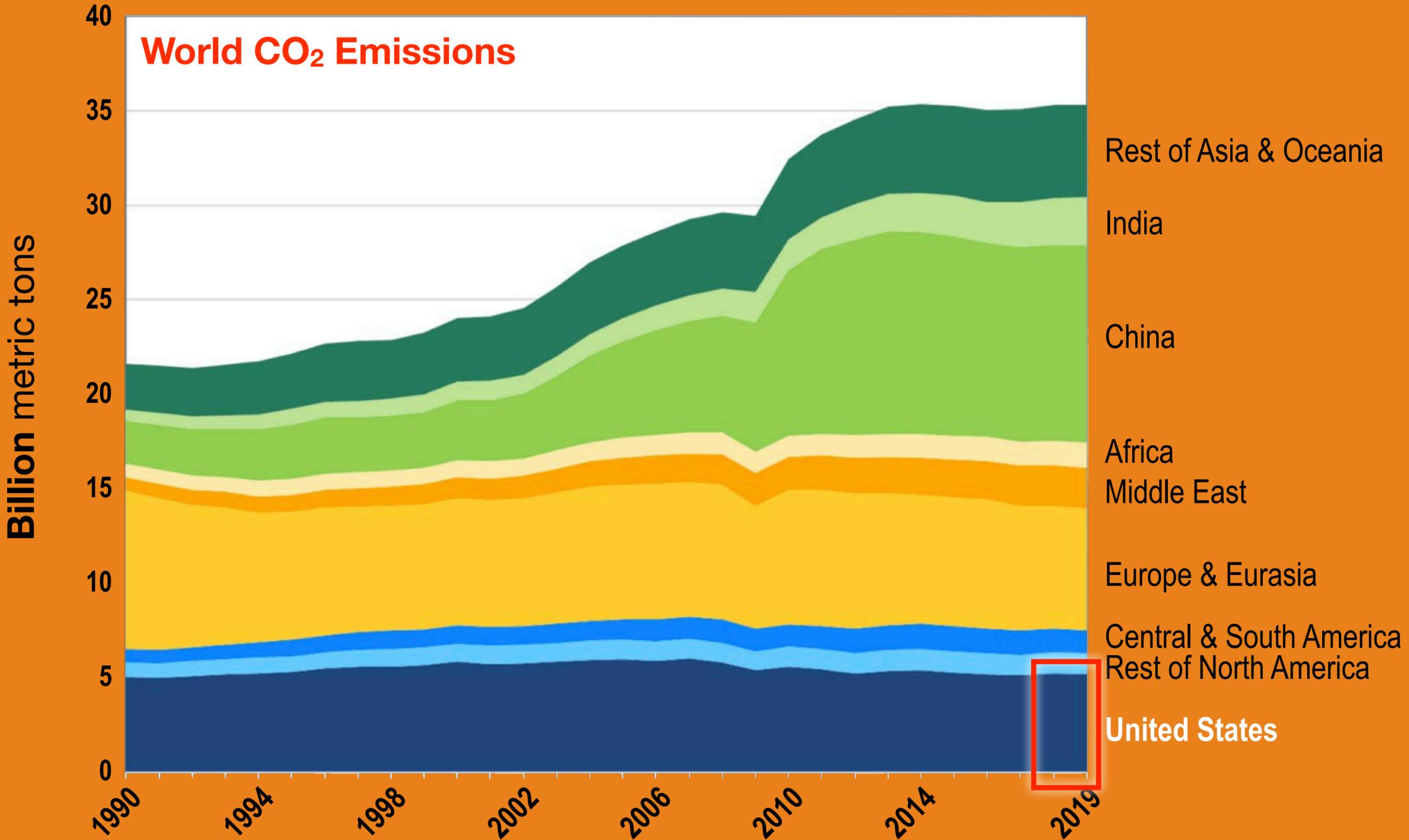
# Global temperatures directly linked to atmospheric CO<sub>2</sub>



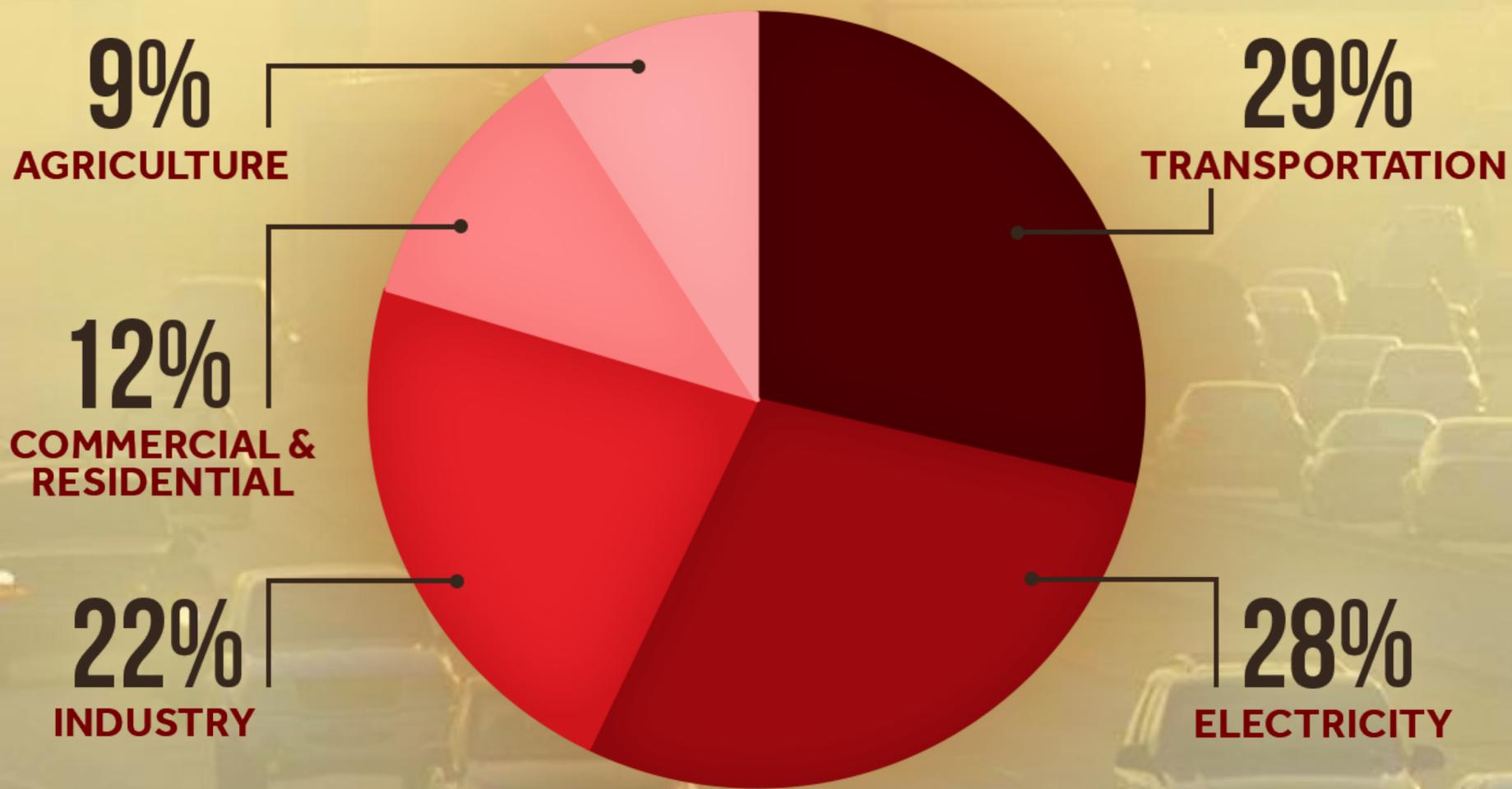
# Global temperatures directly linked to atmospheric CO<sub>2</sub>



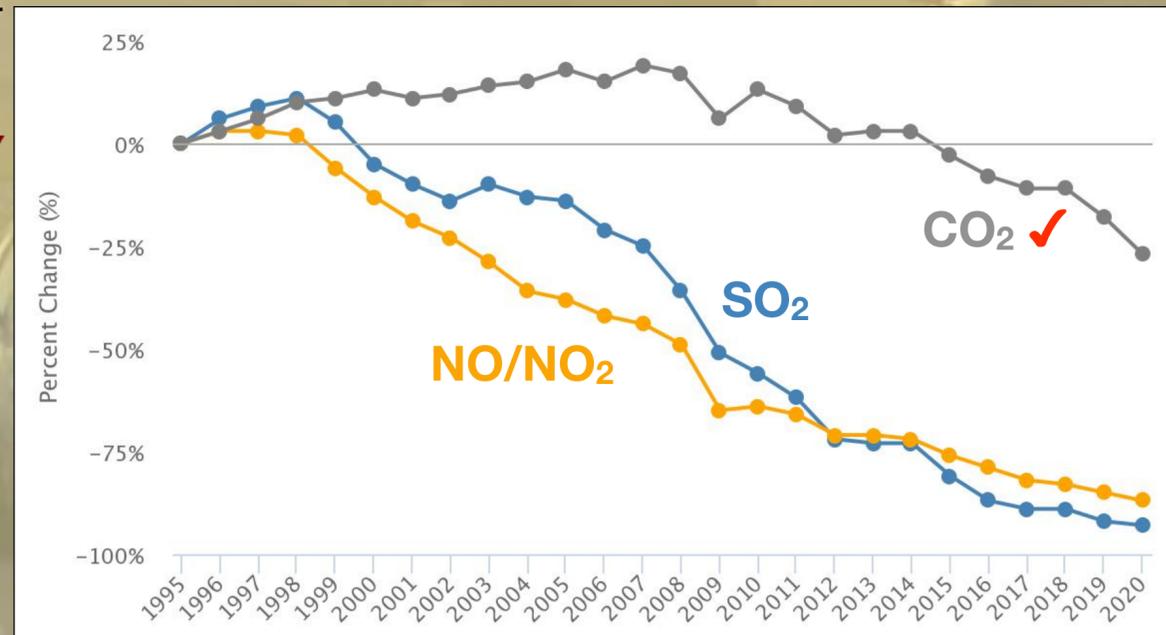
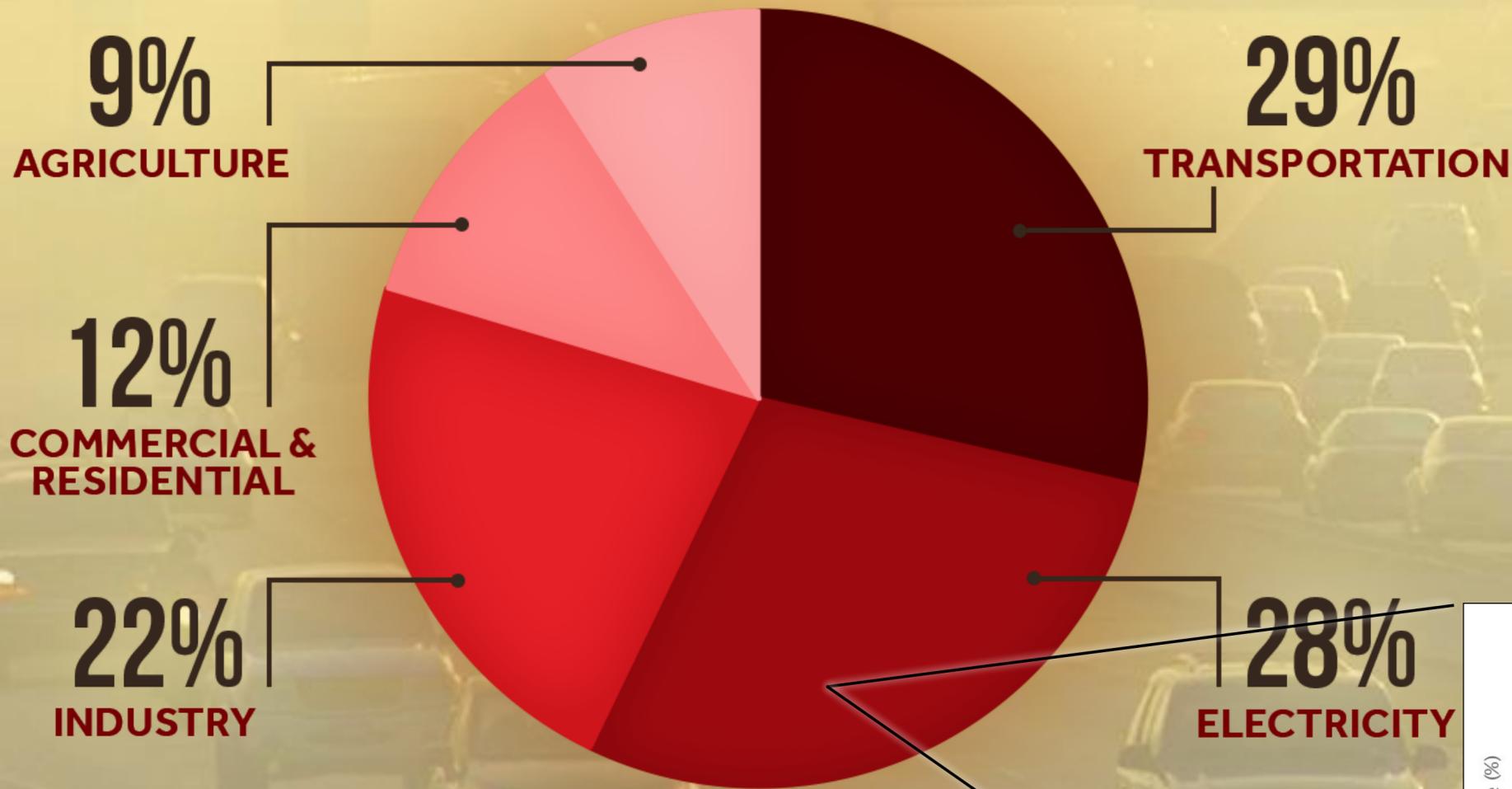
# Global temperatures directly linked to atmospheric CO<sub>2</sub>



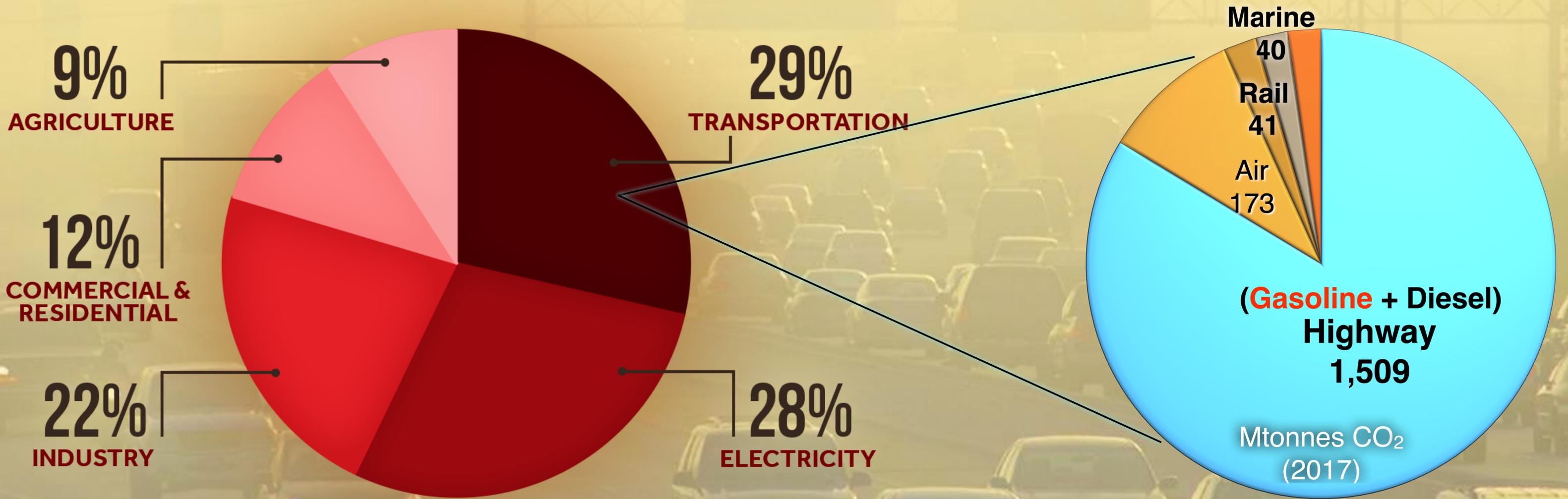
# Highest US emissions from transportation



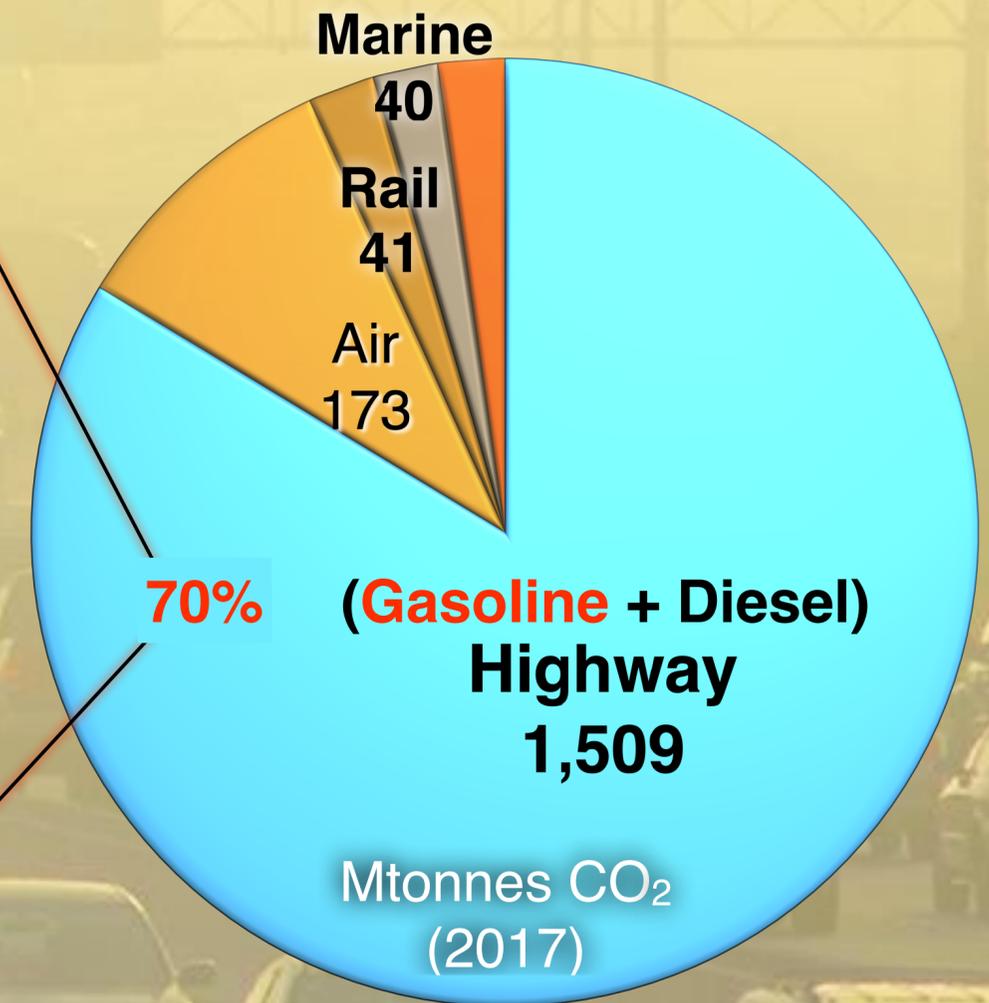
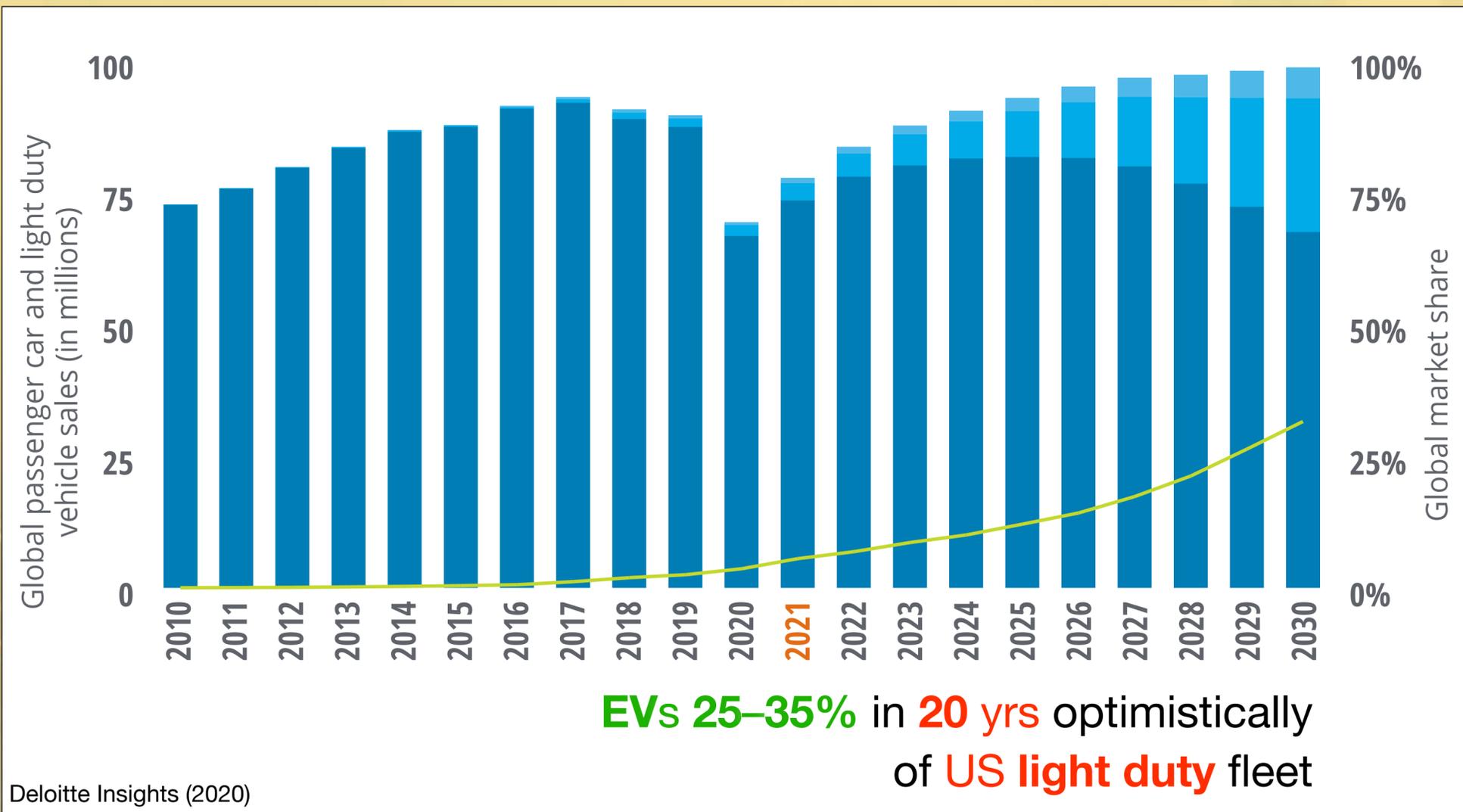
# Highest US emissions from transportation



# Highest US emissions from transportation

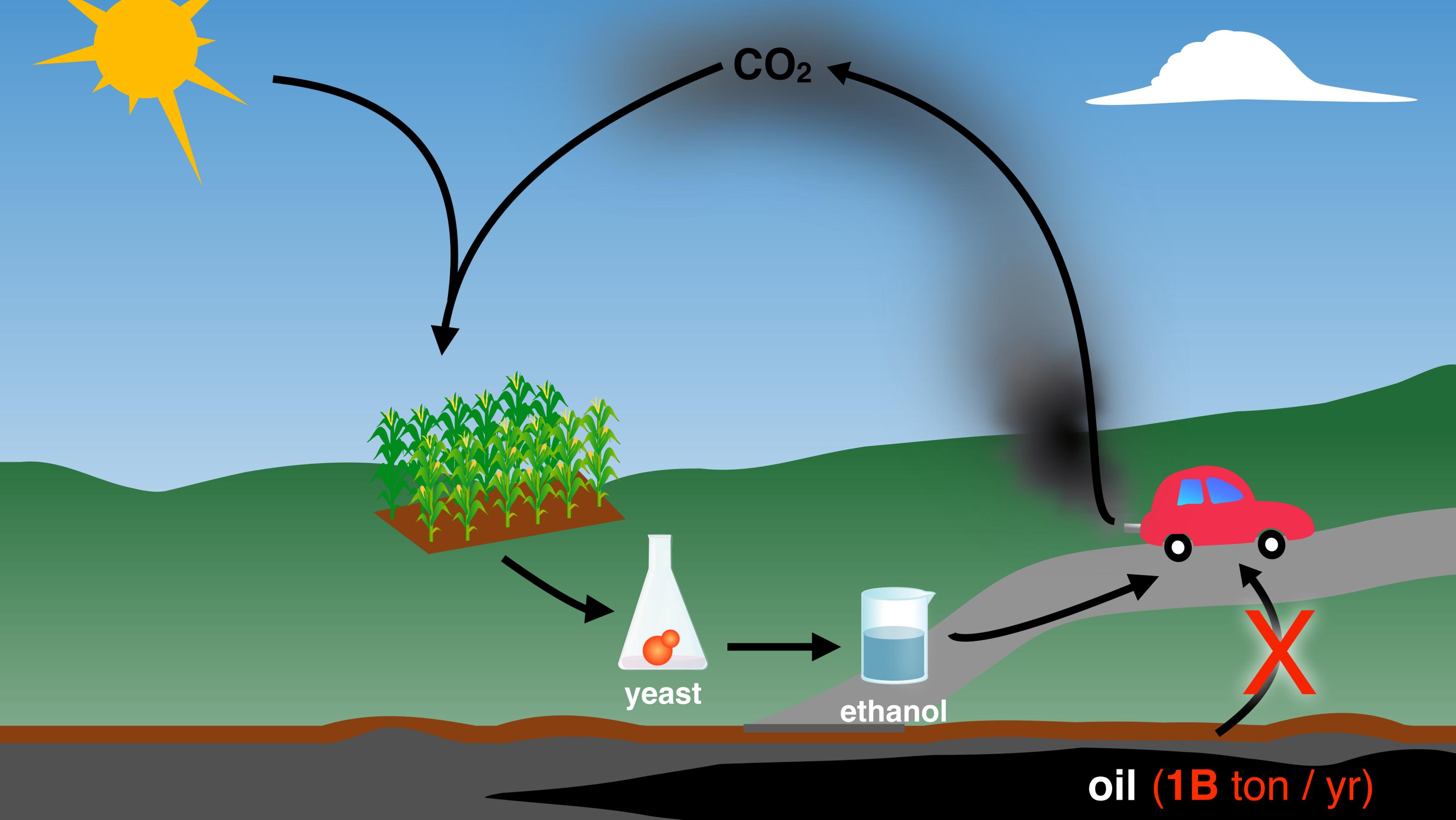


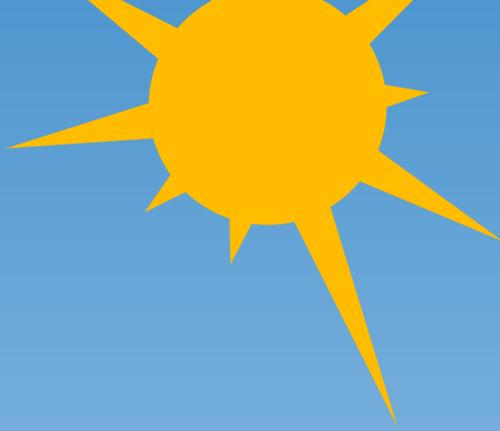
# Internal combustion fleet will predominate indefinitely



→ **Renewable liquid combustion biofuels**  
**critical to global CO<sub>2</sub> reduction**

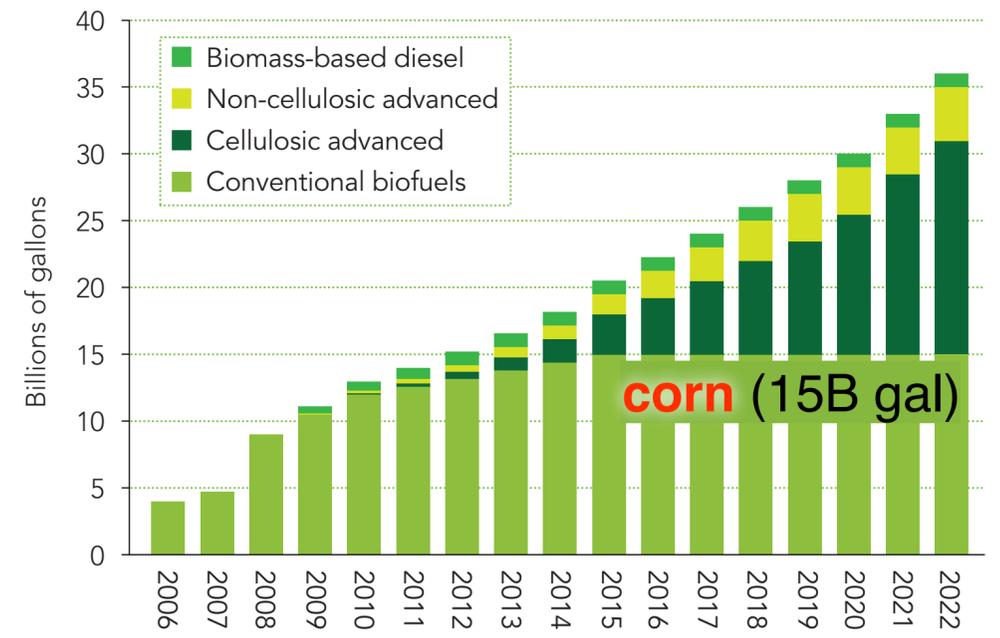






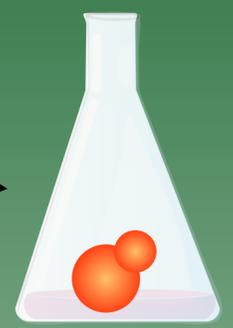
# EPA Renewable Fuel Standard (2007)

## U.S. BIOFUELS PLAN

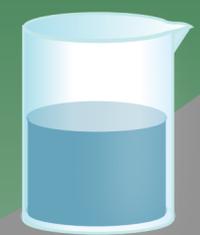


corn (15B gal)

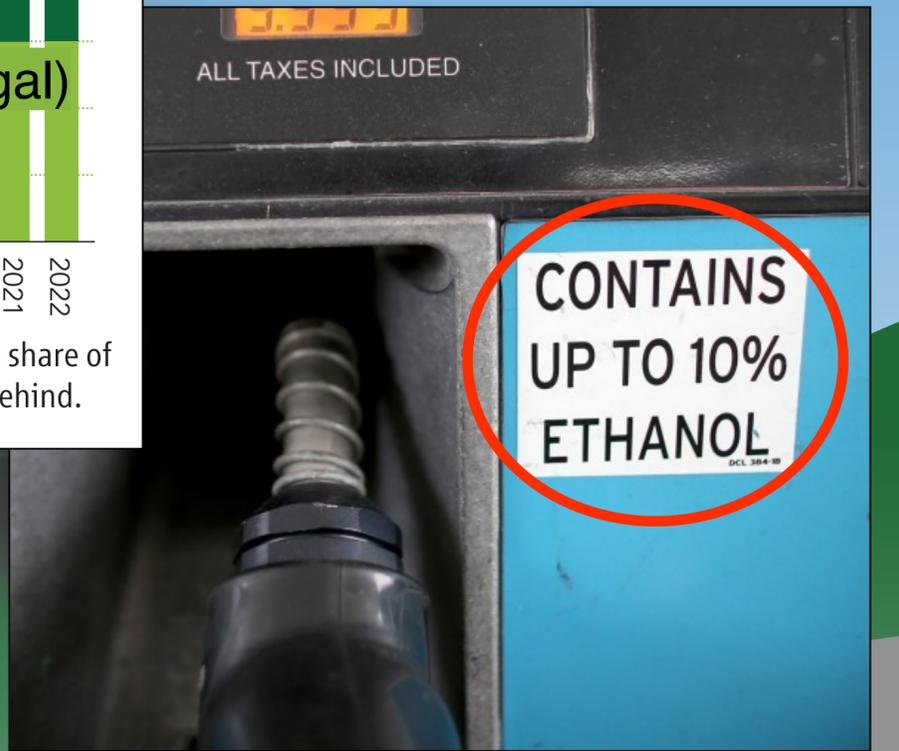
**Growing gap.** Energy legislation from 2007 mandates an increasing share of cellulosic ethanol (dark green). But the industry is already falling behind.



yeast



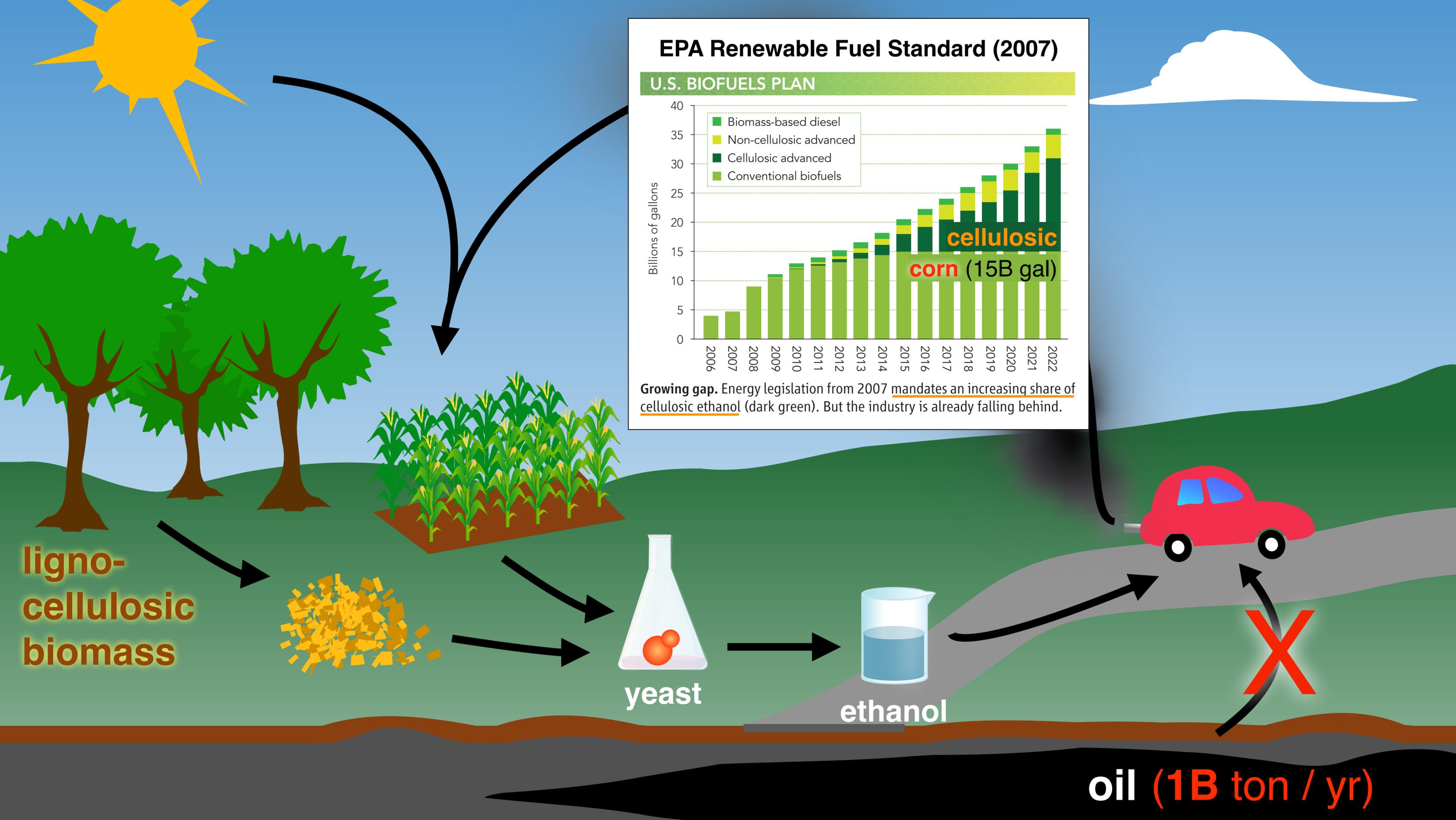
ethanol



10%

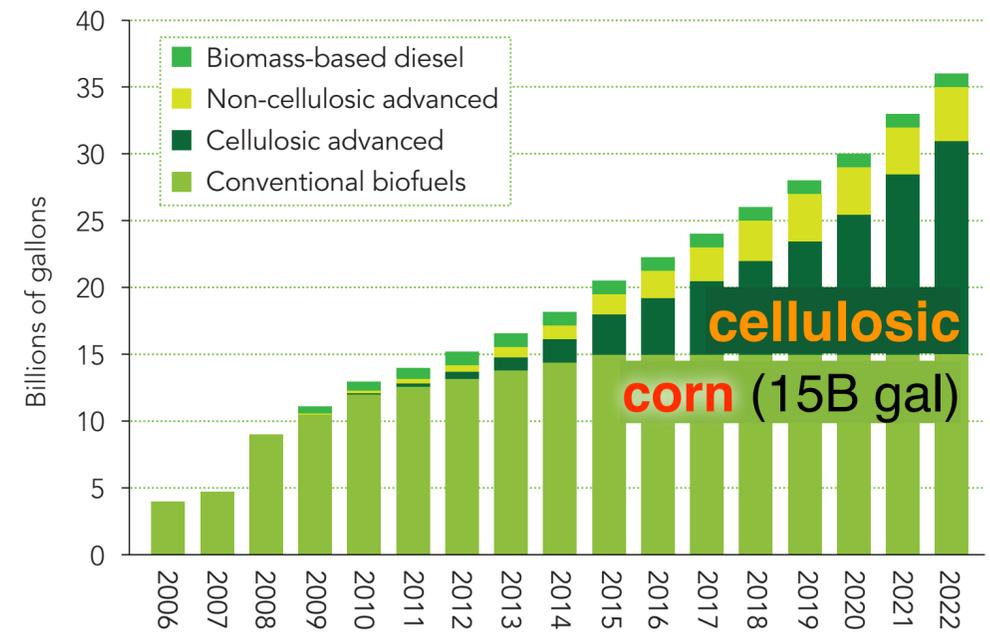


oil (1B ton / yr)



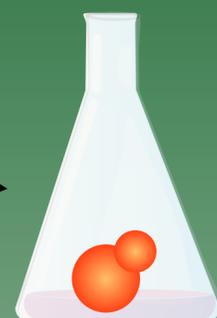
## EPA Renewable Fuel Standard (2007)

### U.S. BIOFUELS PLAN

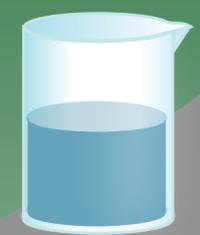


**Growing gap.** Energy legislation from 2007 mandates an increasing share of cellulosic ethanol (dark green). But the industry is already falling behind.

**ligno-cellulosic biomass**



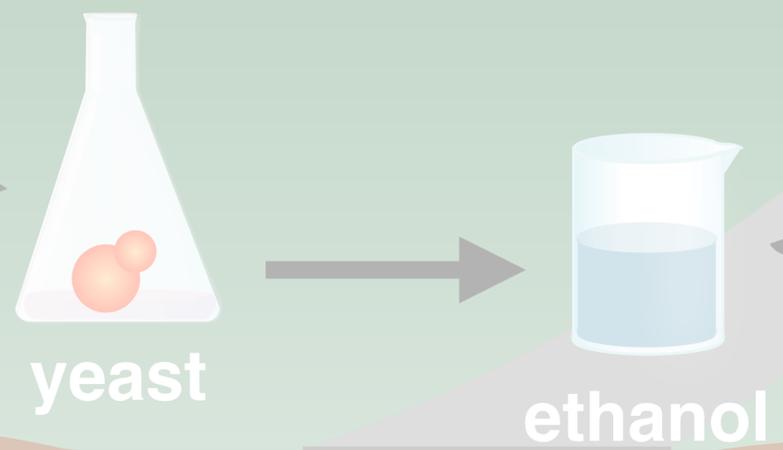
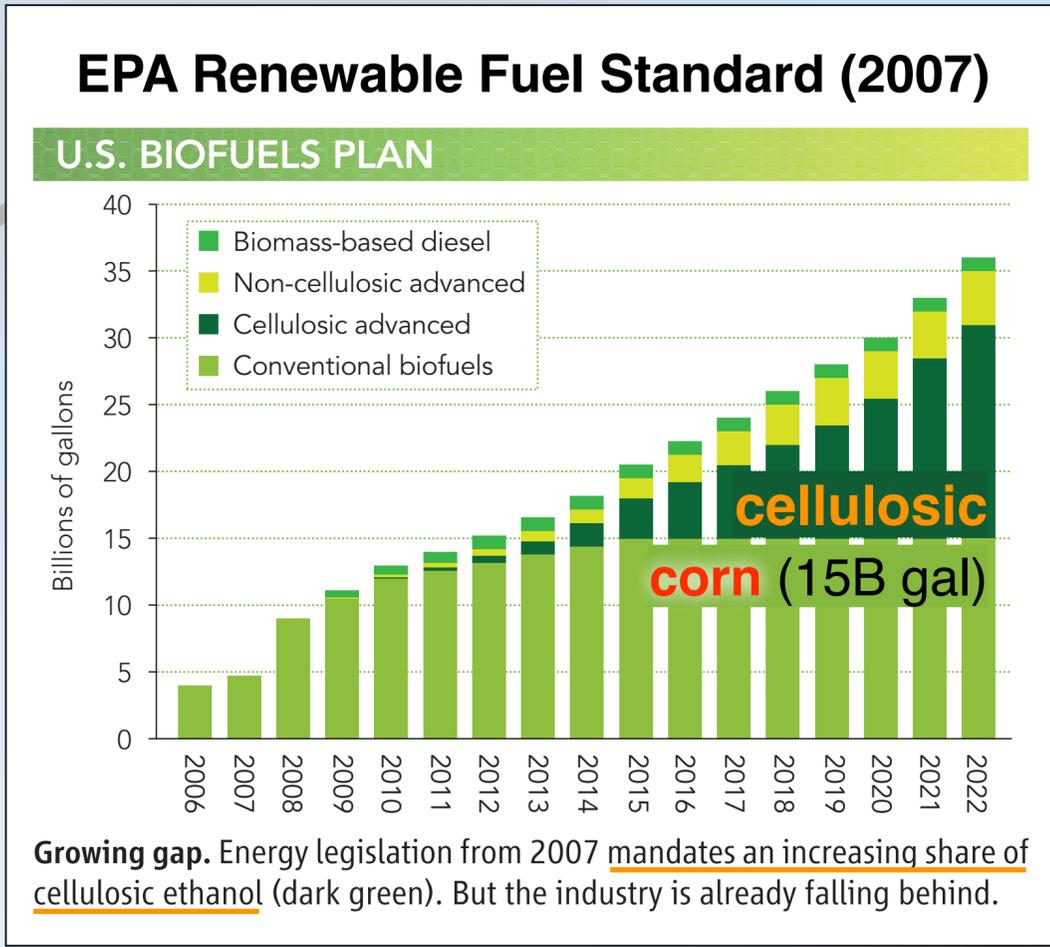
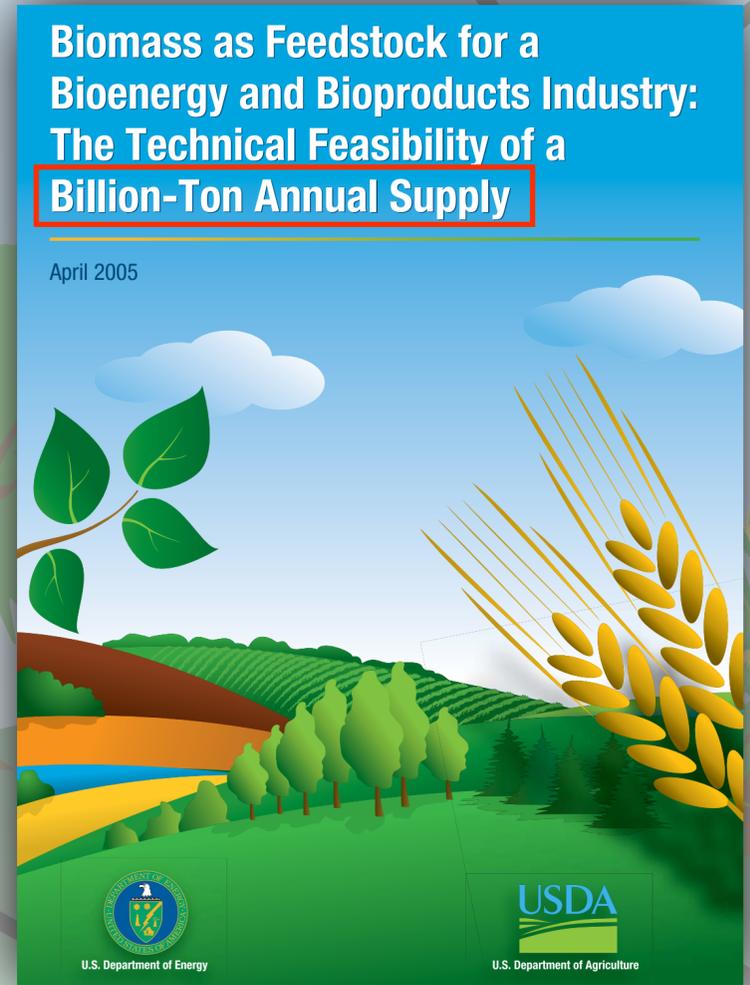
**yeast**



**ethanol**

**oil (1B ton / yr)**

**ligno-cellulosic biomass**  
**(~1B ton / yr)**



**oil (1B ton / yr)**

ENERGY

# Cellulosic ethanol fights for life

Pioneering biofuel producers hope that US government largesse will ease their way into a tough market.

BY MARK PELOW

On the flat plains of Kansas, a stack of gleaming steel towers and pipes stretches 16 storeys into the sky. More than 1,000 construction workers toiled to complete the ethanol plant near the town of

by DuPont). The industry has long promised that this second-generation biofuel will reduce greenhouse-gas emissions, reduce US reliance on imported oil and boost rural economies just as the fuel is on the cusp of making it market forces and government policies choke its progress. "This is going to be a

included in the country's fuel supply. In its early years, the law emphasized the production of corn ethanol, considered ripe for early commercialization.

Yet corn ethanol comes with problems. It offers only modest savings in greenhouse-gas emissions compared to petrol (see *Nature* 499, 13–14; 2013). Production is vulnerable to poor harvests and can contribute to increased food prices because the maize must be grown on land that would otherwise be used for food. Tapping the storehouse of biomass left after the harvest is much less controversial. Ethanol made from corn stover produces at least 60% less greenhouse-gas emissions than petrol, and making it does not require any extra farmland.

## MIT Technology Review

### The Cellulosic Ethanol Industry Faces Big Challenges

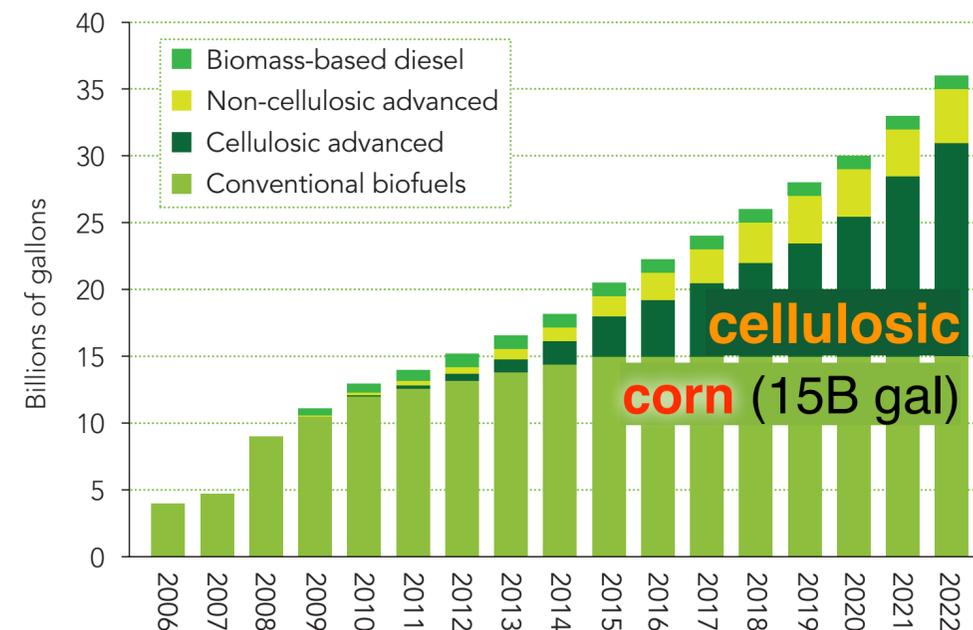
The advanced-biofuels industry is in danger of withering away.

By [Kevin Bullis](#) on August 12, 2013

A series of cellulosic-biofuel plants are finally starting to come on line after years of delay. But the wave of plant openings, good news as it is for the emerging industry, also shows just how far it still has to go.

## EPA Renewable Fuel Standard (2007)

### U.S. BIOFUELS PLAN



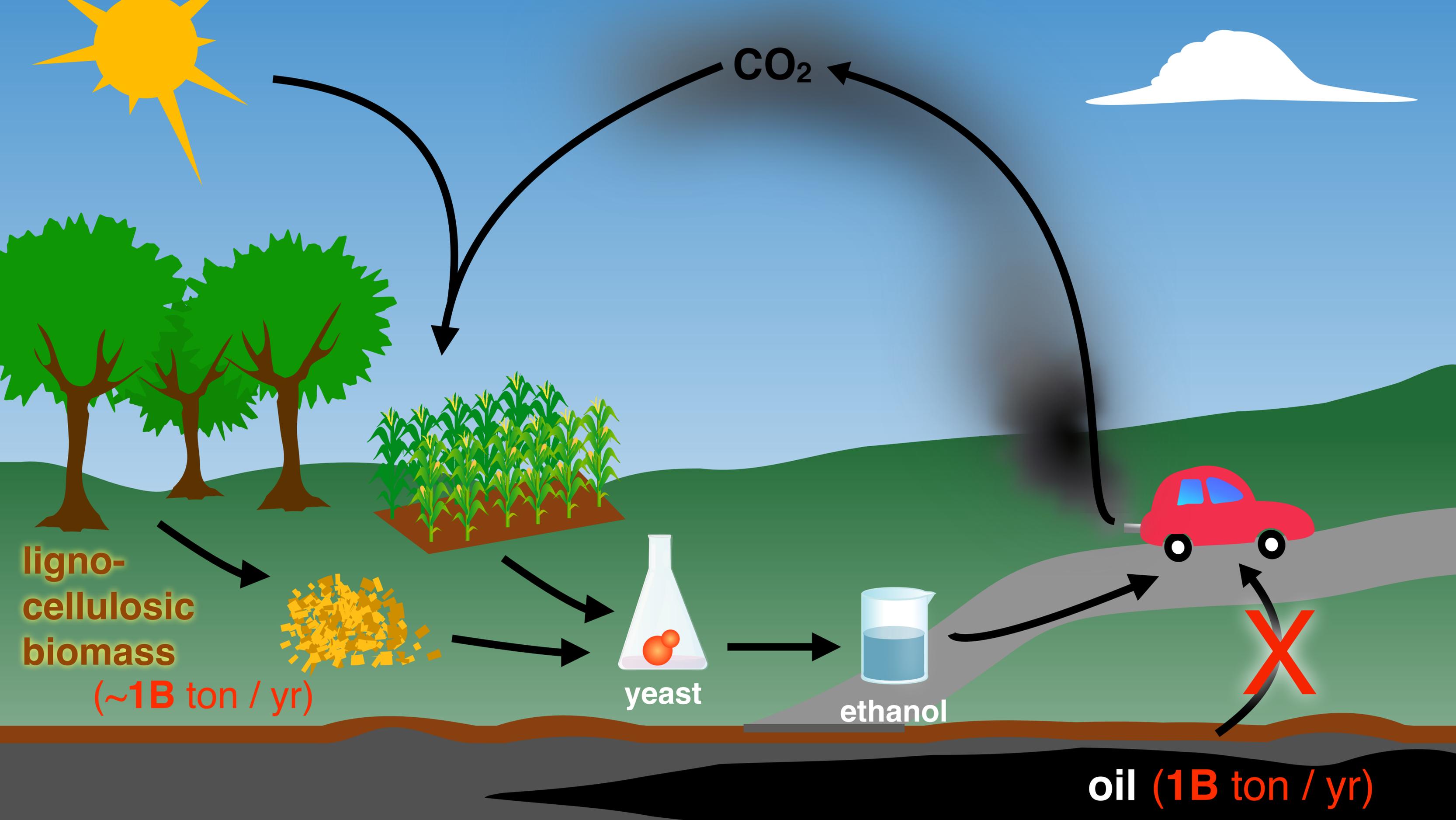
**Growing gap.** Energy legislation from 2007 mandates an increasing share of cellulosic ethanol (dark green). **But the industry is already falling behind.**

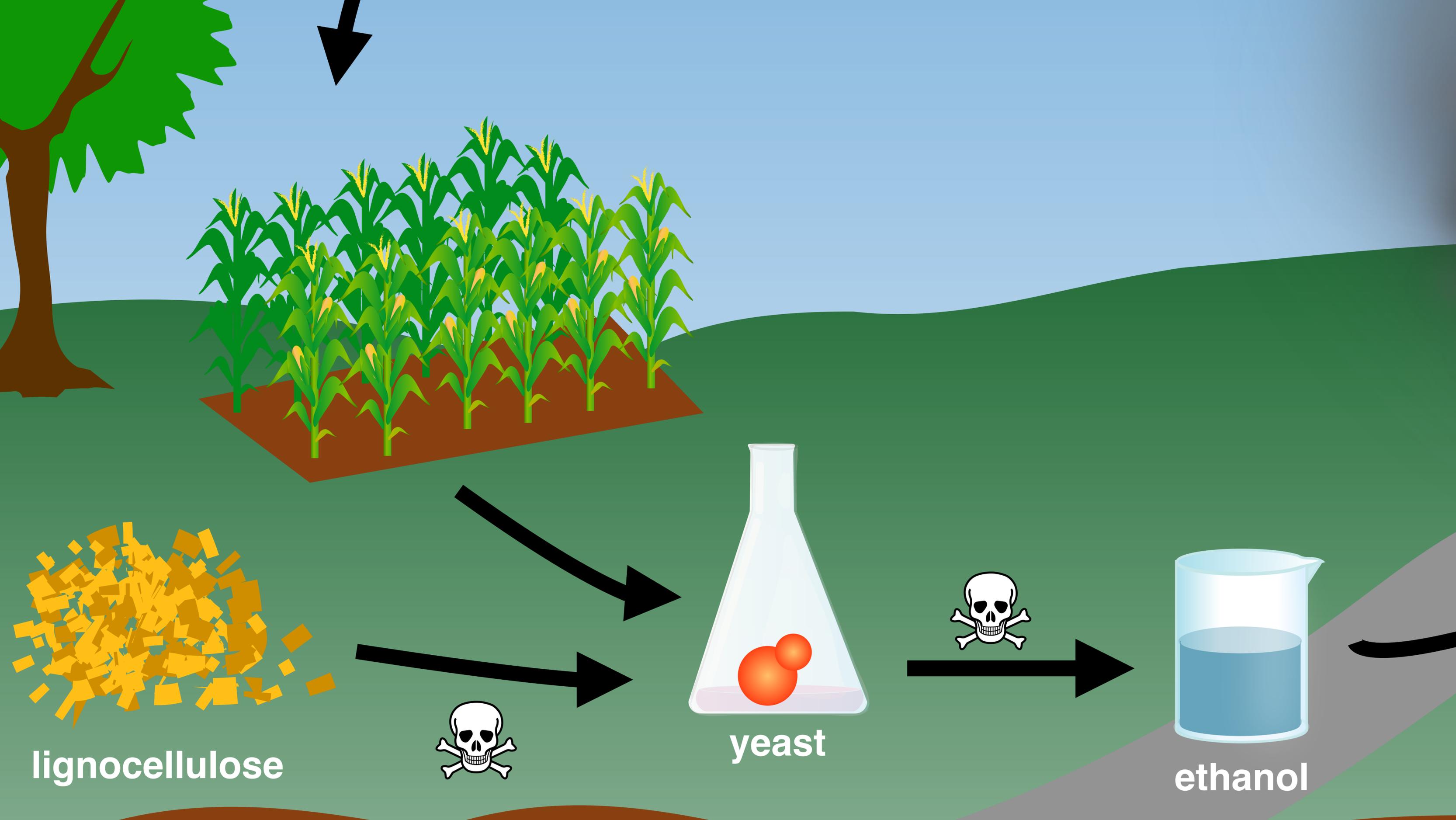
ligno-cellulosic biomass

(~1B ton / yr)

oil (1B ton / yr)

anol

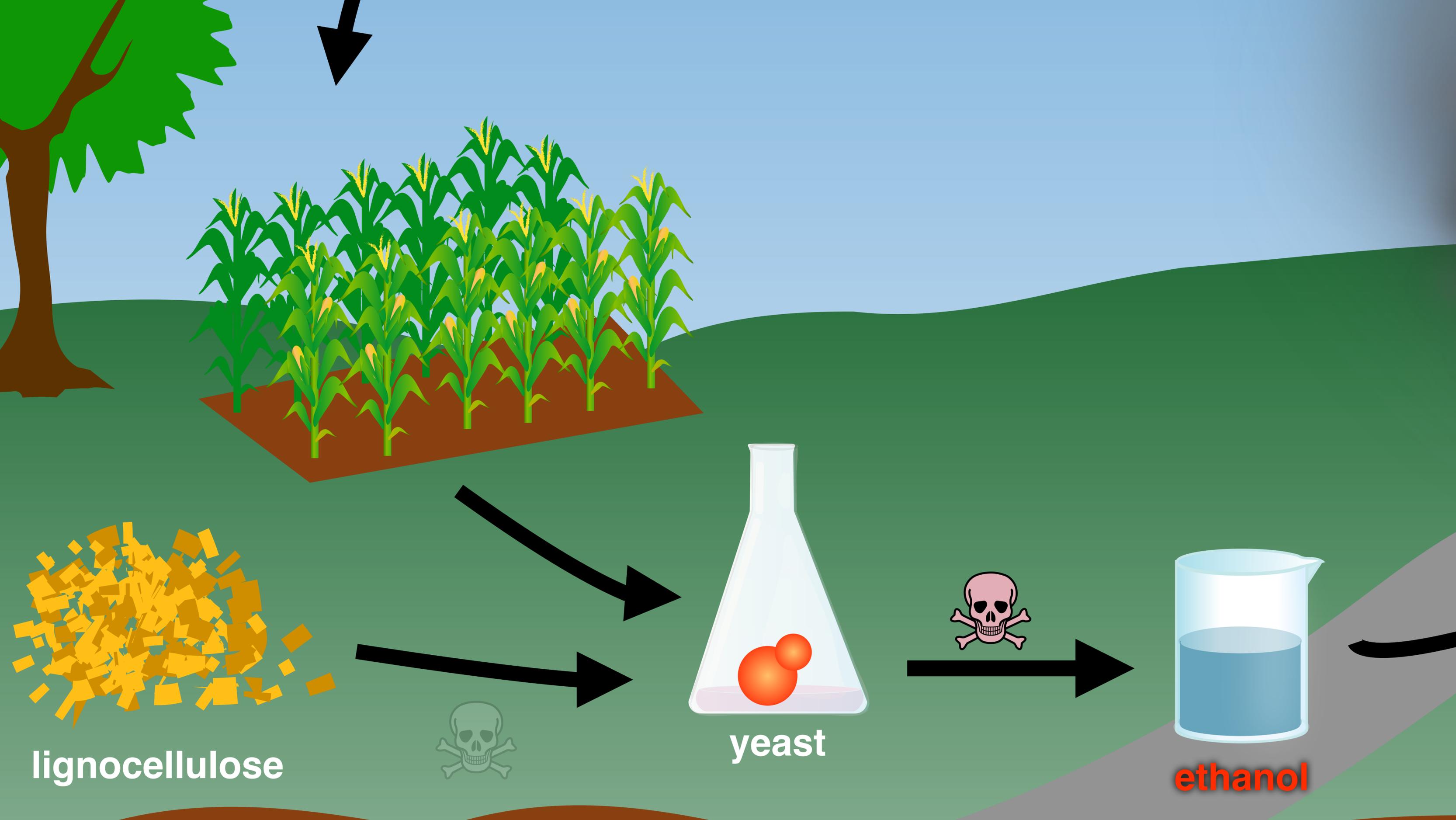




**lignocellulose**

**yeast**

**ethanol**



**lignocellulose**

**yeast**

**ethanol**

# Origin of ethanol tolerance **unknown**

## Yeast

Yeast 2006; **23**: 351–359.

Published online in Wiley InterScience (www.interscience.wiley.com). DOI: 10.1002/yea.1359

## Research Article

### Genome-wide identification of genes required for growth of *Saccharomyces cerevisiae* under ethanol stress

Frank van Voorst, Jens Houghton-Larsen, Lars Jønson<sup>#</sup>, Morten C. Kielland-Brandt and Anders Brandt\*  
Carlsberg Laboratory, Gamle Carlsberg Vej 10, DK-2500 Copenhagen Valby, Denmark

Copyright © 2010 by the Genetics Society of America  
DOI: 10.1534/genetics.110.121871

### Exploiting Natural Variation in *Saccharomyces cerevisiae* to Identify Genes for Increased Ethanol Resistance

Jeffrey A. Lewis,<sup>\*,†</sup> Isaac M. Elkon,<sup>\*,†</sup> Mick A. McGee,<sup>†,‡</sup> Alan J. Higbee<sup>†,‡</sup>  
and Audrey P. Gasch<sup>\*,†,§,1</sup>

<sup>\*</sup>Laboratory of Genetics, <sup>†</sup>Great Lakes Bioenergy Research Center, <sup>‡</sup>Biotechnology Center, and <sup>§</sup>Genome Center of Wisconsin, University of Wisconsin, Madison, Wisconsin 53706

Manuscript received August 5, 2010

APPLIED AND ENVIRONMENTAL MICROBIOLOGY, Mar. 2003, p. 1499–1503  
0099-2240/03/\$08.00+0 DOI: 10.1128/AEM.69.3.1499–1503.2003  
Copyright © 2003, American Society for Microbiology. All Rights Reserved.

Vol. 69, No. 3

### Ethanol Tolerance in the Yeast *Saccharomyces cerevisiae* Is Dependent on Cellular Oleic Acid Content

Kyung Man You,<sup>†</sup> Claire-Lise Rosenfield, and Douglas C. Knipple\*

Department of Entomology, Cornell University, New York State Agricultural Experiment Station,  
Geneva, New York 14456



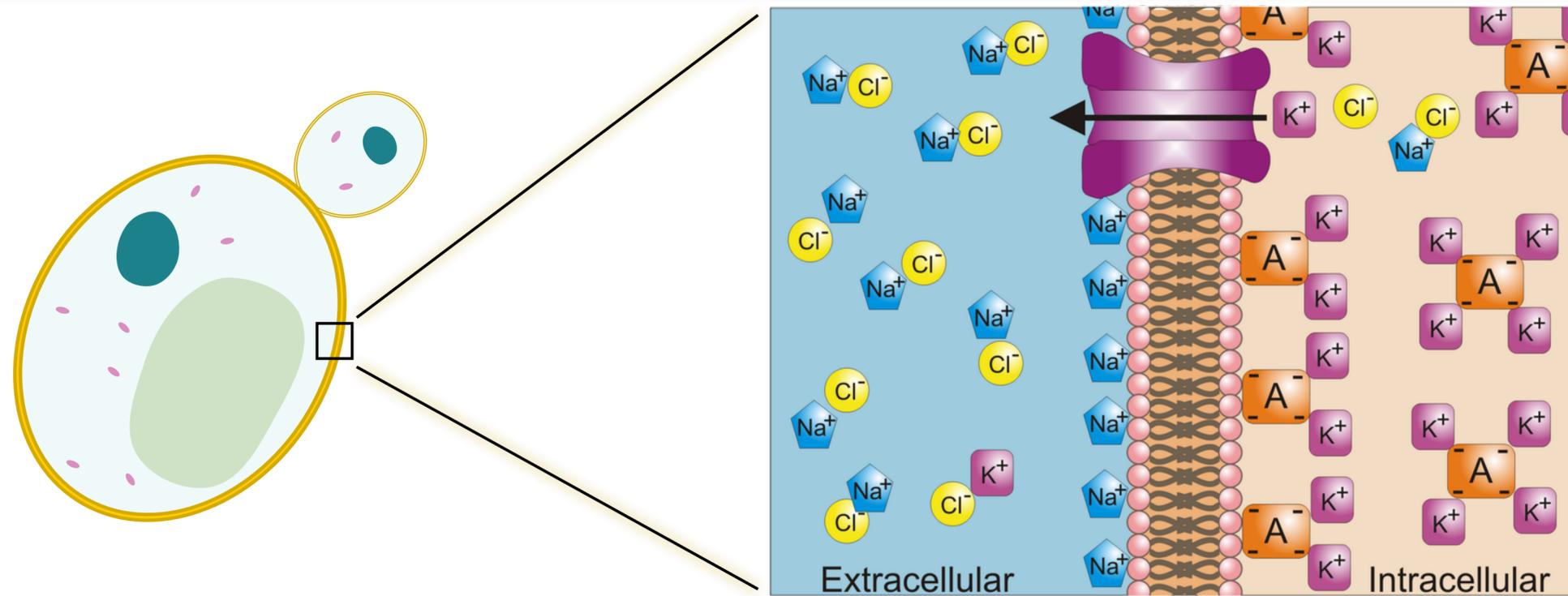
## RESEARCH ARTICLE

### Trehalose promotes the survival of *Saccharomyces cerevisiae* during lethal ethanol stress, but does not influence growth under sublethal ethanol stress

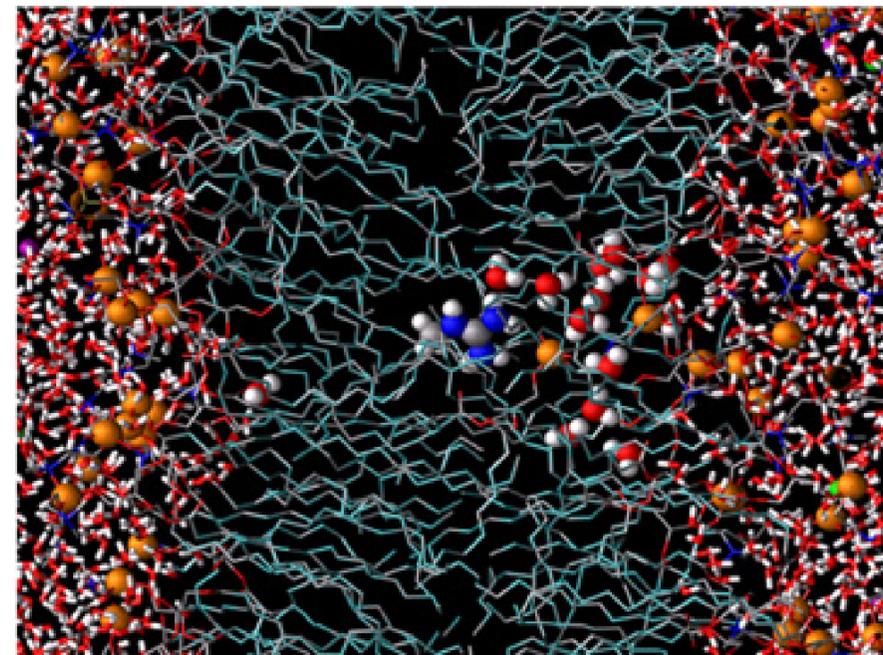
Ajith Bandara<sup>1</sup>, Sarah Fraser<sup>1</sup>, Paul J. Chambers<sup>2</sup> & Grant A. Stanley<sup>1</sup>

<sup>1</sup>School of Engineering and Science, Victoria University, Melbourne, Vic., Australia; and <sup>2</sup>The Australian Wine Research Institute, Glen Osmond, SA, Australia

# Ethanol disrupts membranes → environmental ions affect stability?

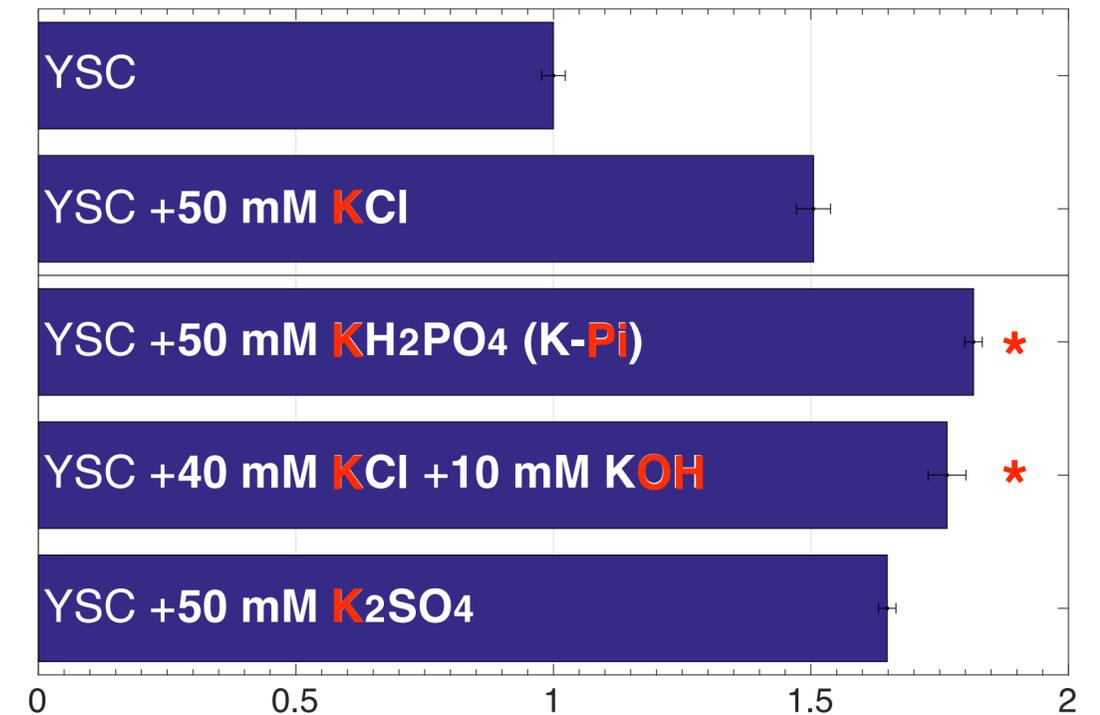
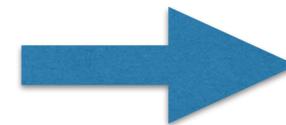
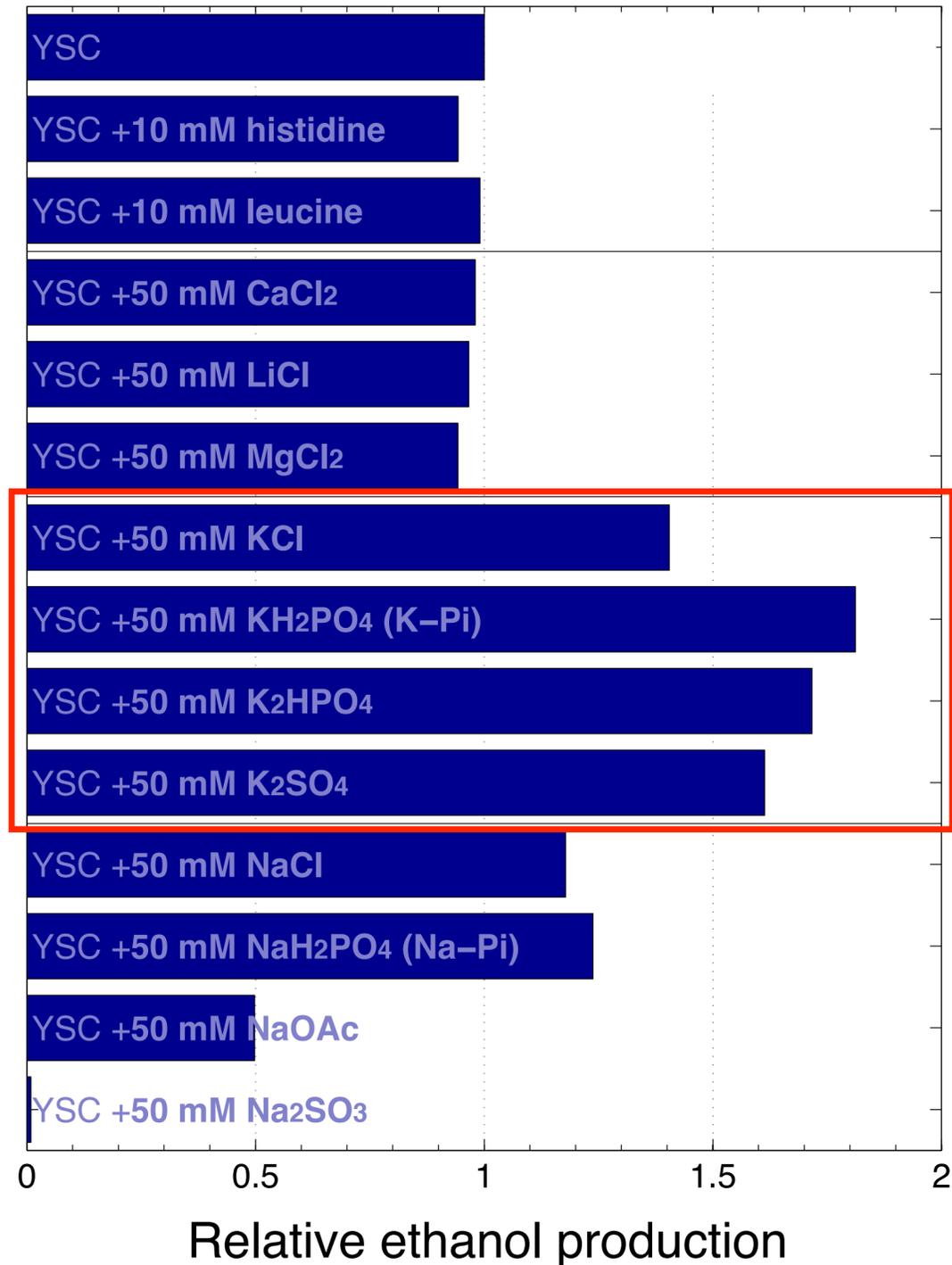


Hypothesis:  
**ethanol** interacts /  
disrupts lipids...



→ can  
**external ions**  
counteract?

# Only external **K<sup>+</sup>** and **pH** counteract → **boost ethanol** output

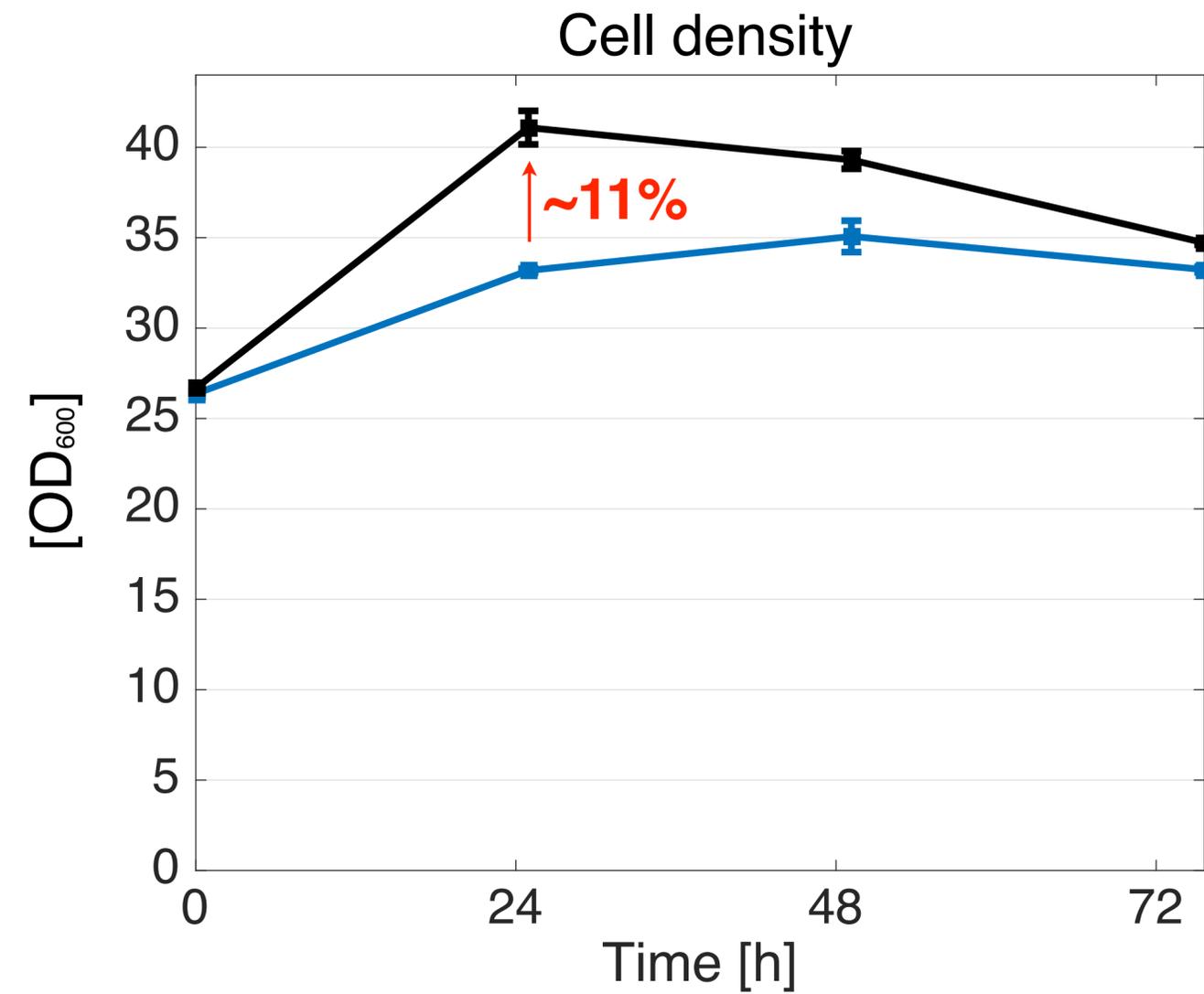
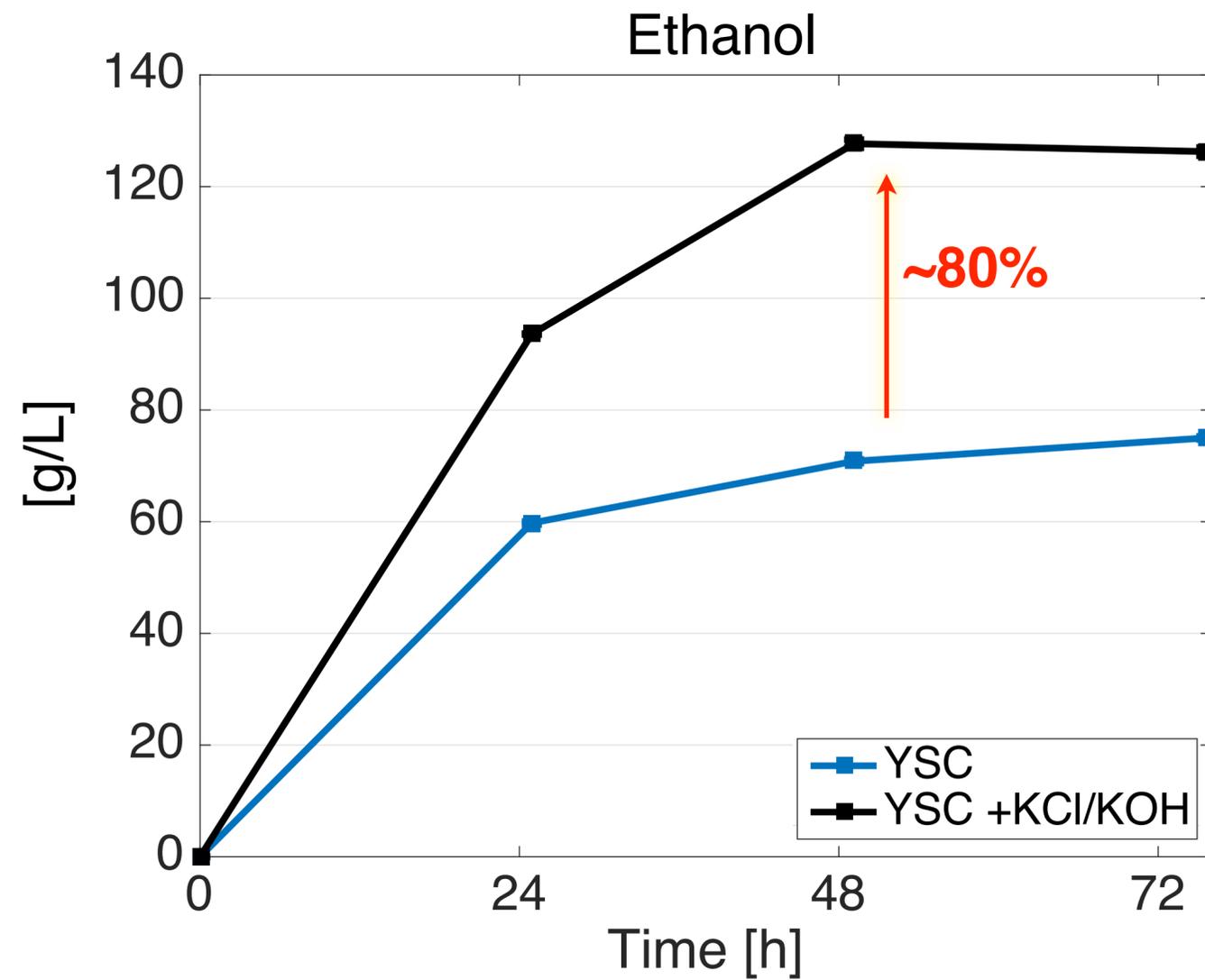


\* statistically identical

**...raising pH gives further boost**

**Largest increase from potassium (K<sup>+</sup>) salts...**

# Elevated $K^+$ /pH boost **cell tolerance** — NOT ethanol made **per-cell**



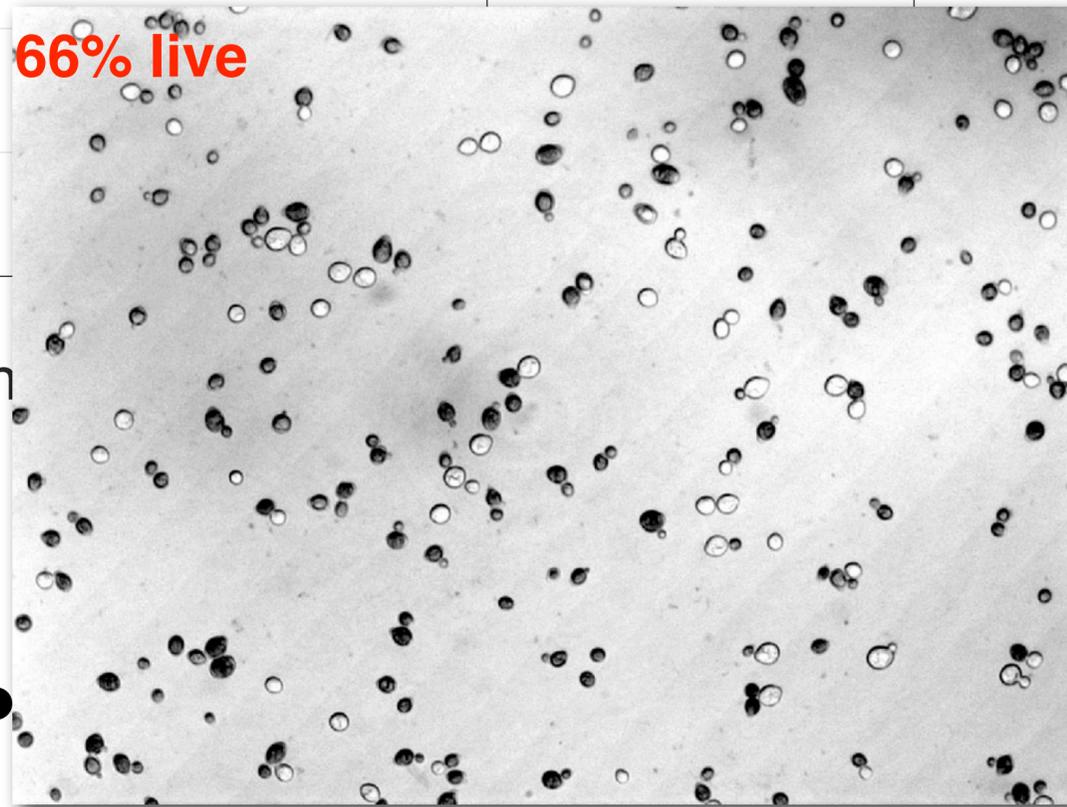
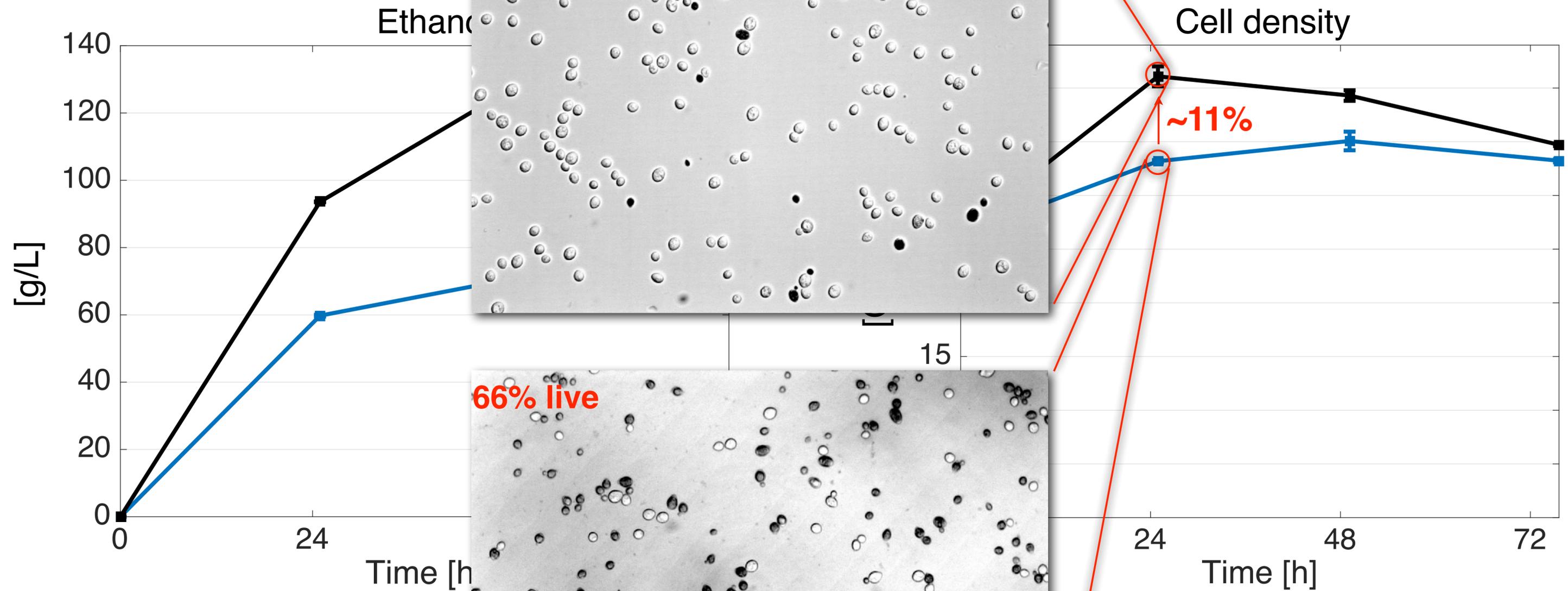
Despite higher ethanol,  $K^+$ /pH enhance:

- Cell growth (moderately)

Elevated  $K^+$ /pH

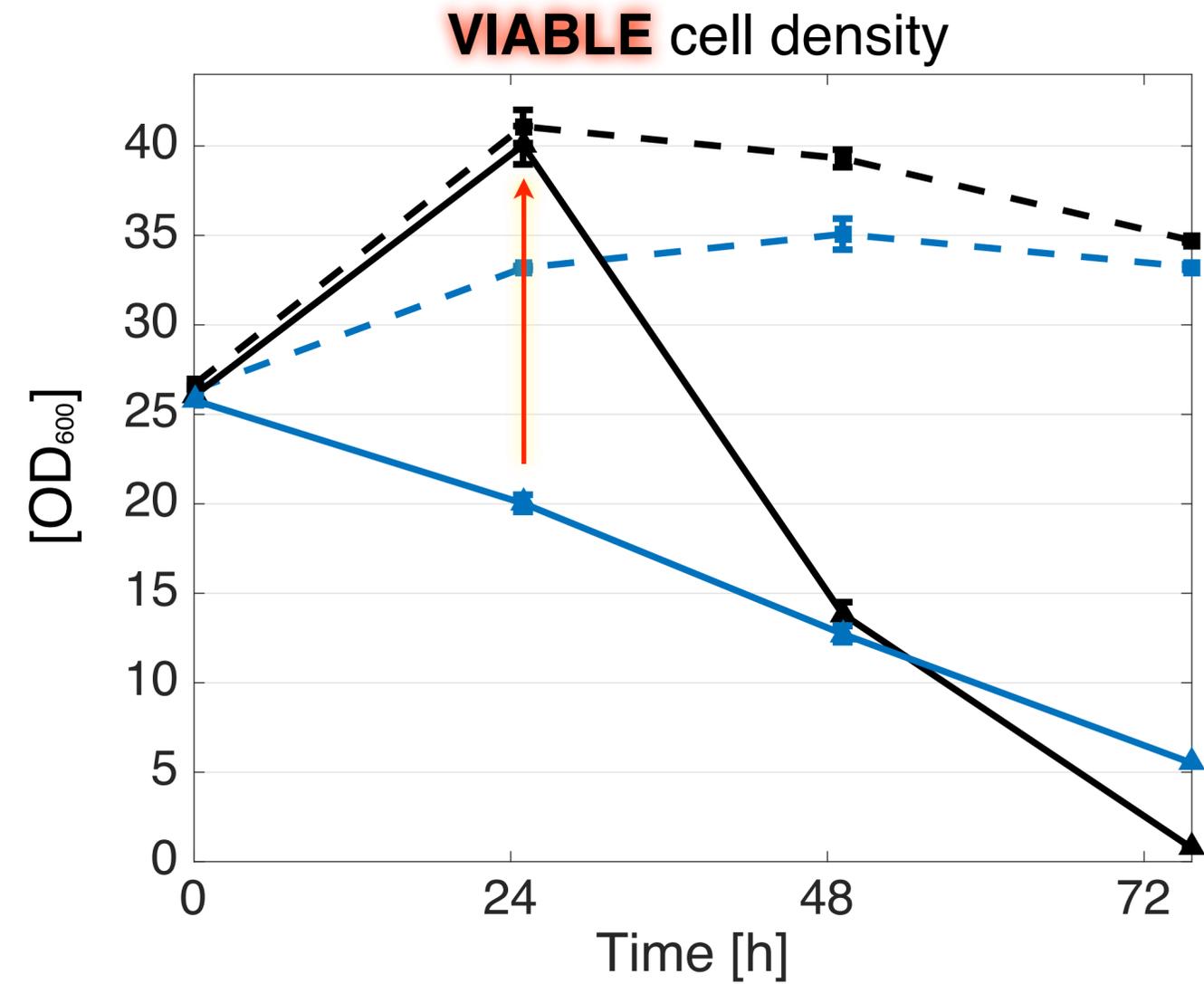
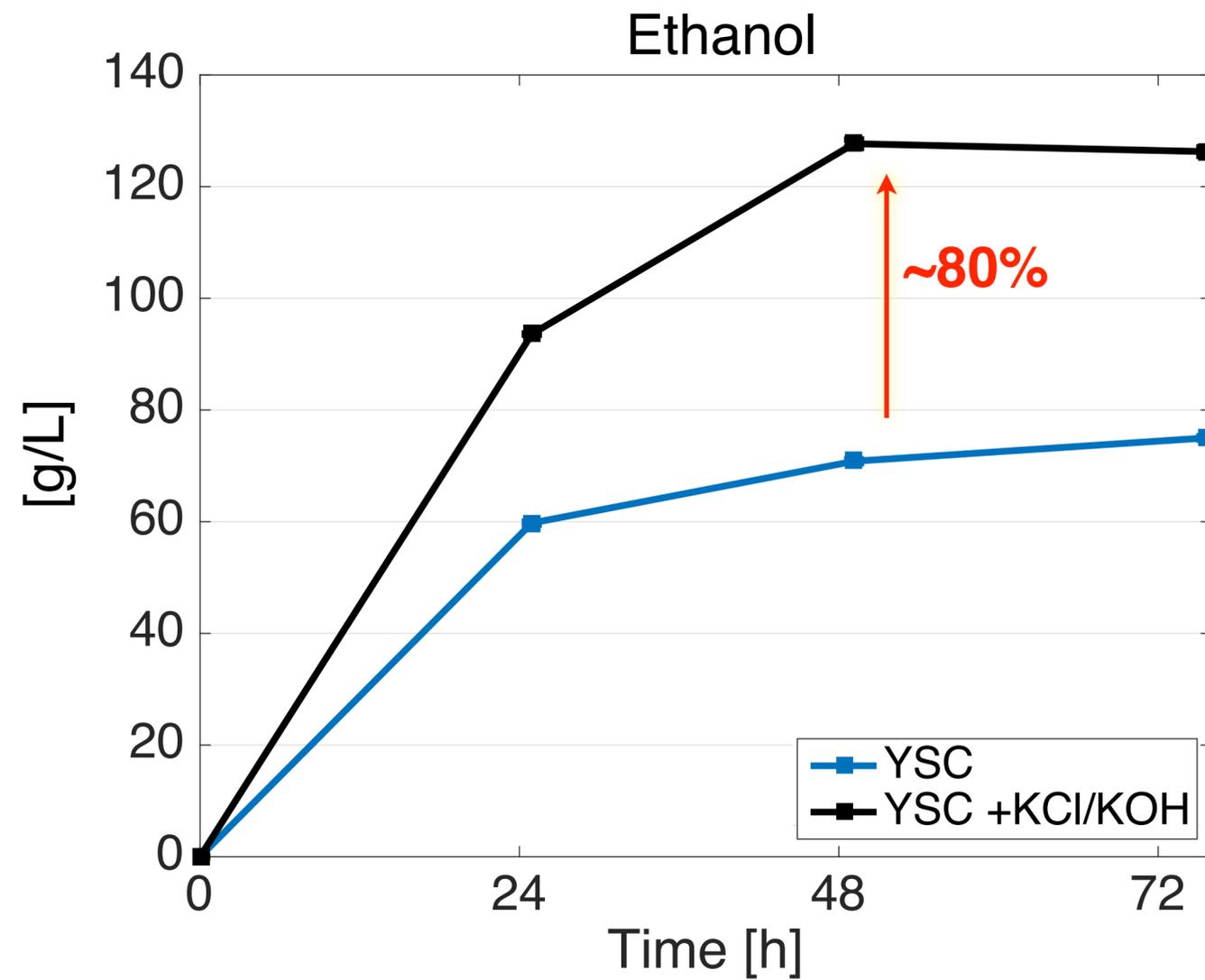
98% live

ethanol made per-cell



enhance:

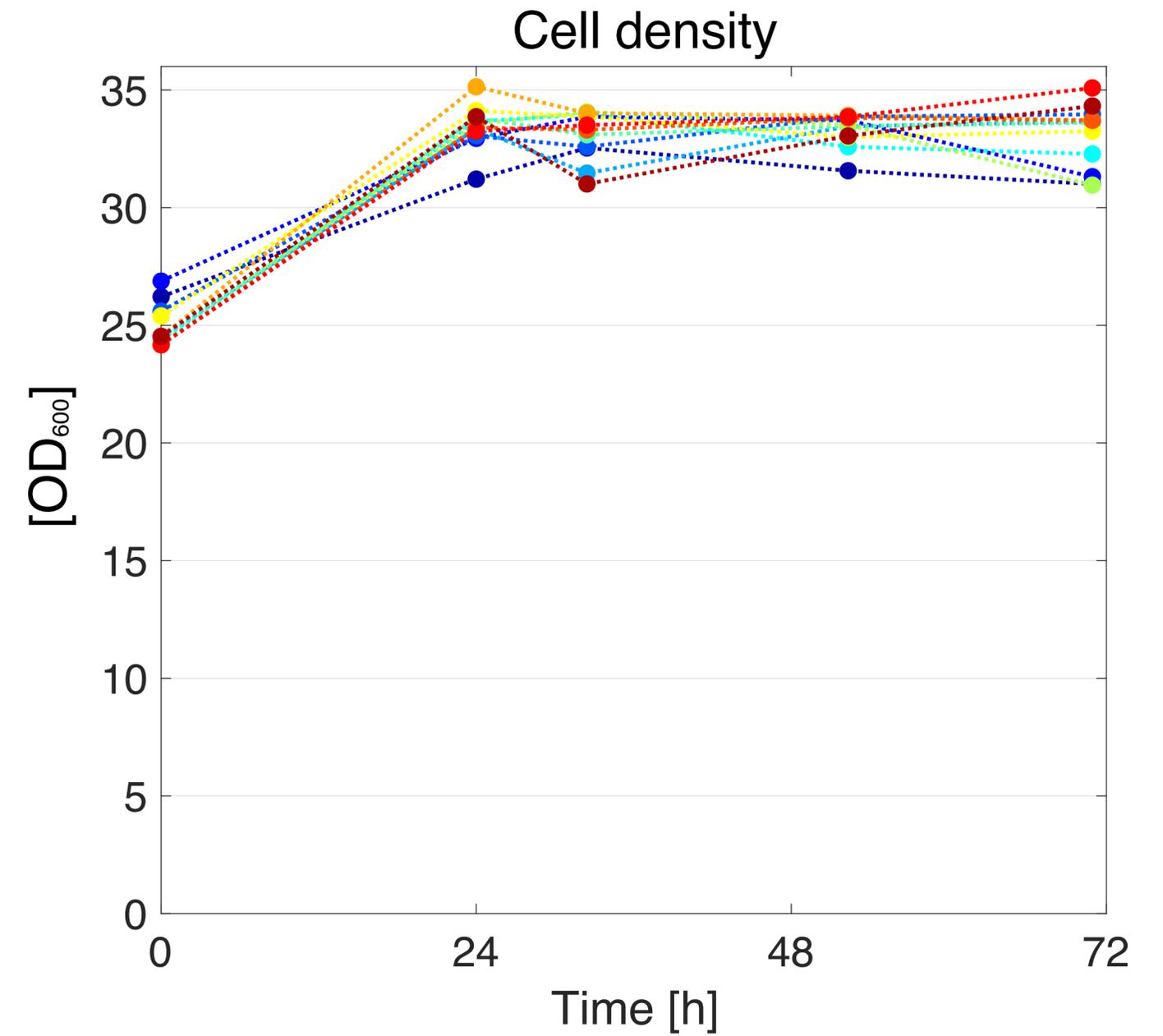
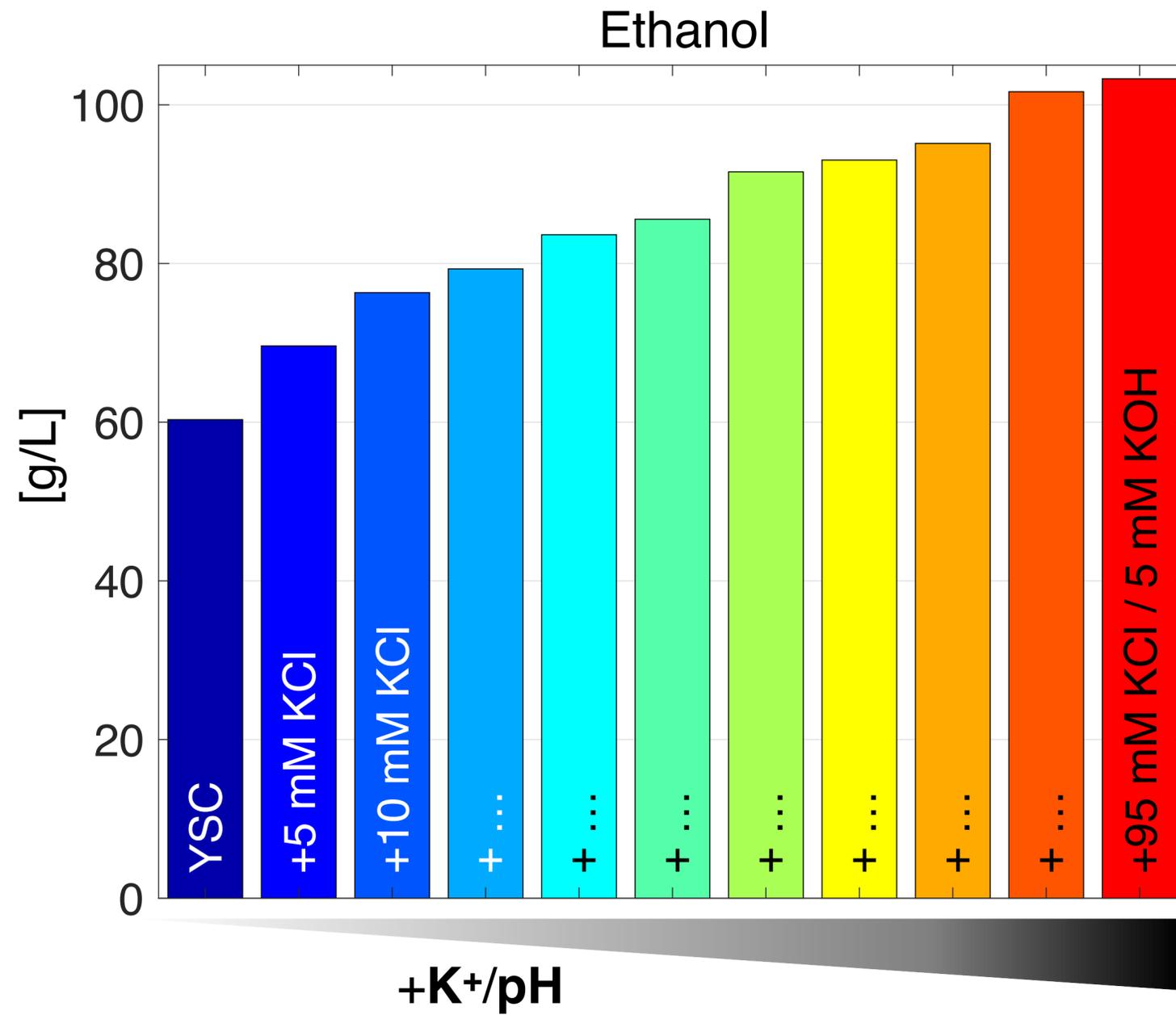
# Elevated $K^+$ /pH boost **cell tolerance** — NOT ethanol made **per-cell**



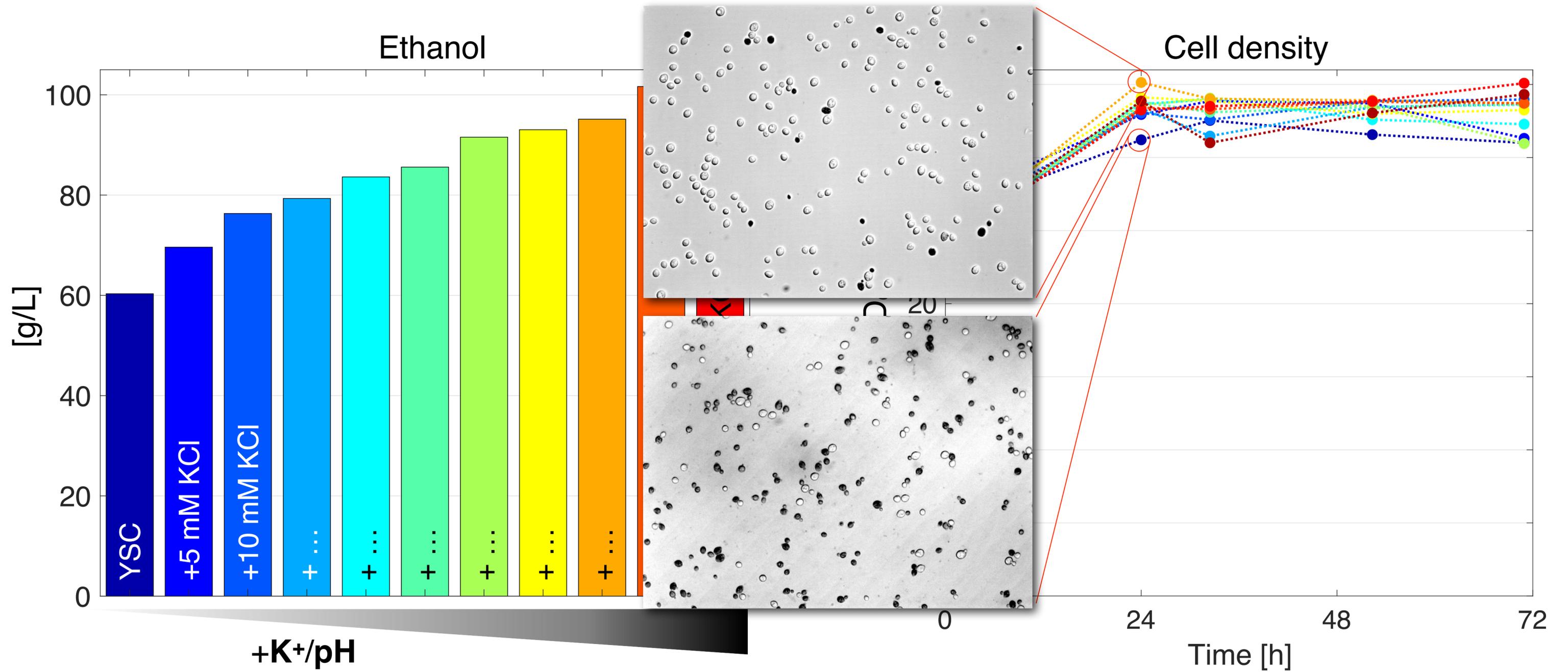
Despite higher ethanol,  $K^+$ /pH enhance:

- Cell growth (moderately)
- **Tolerance**

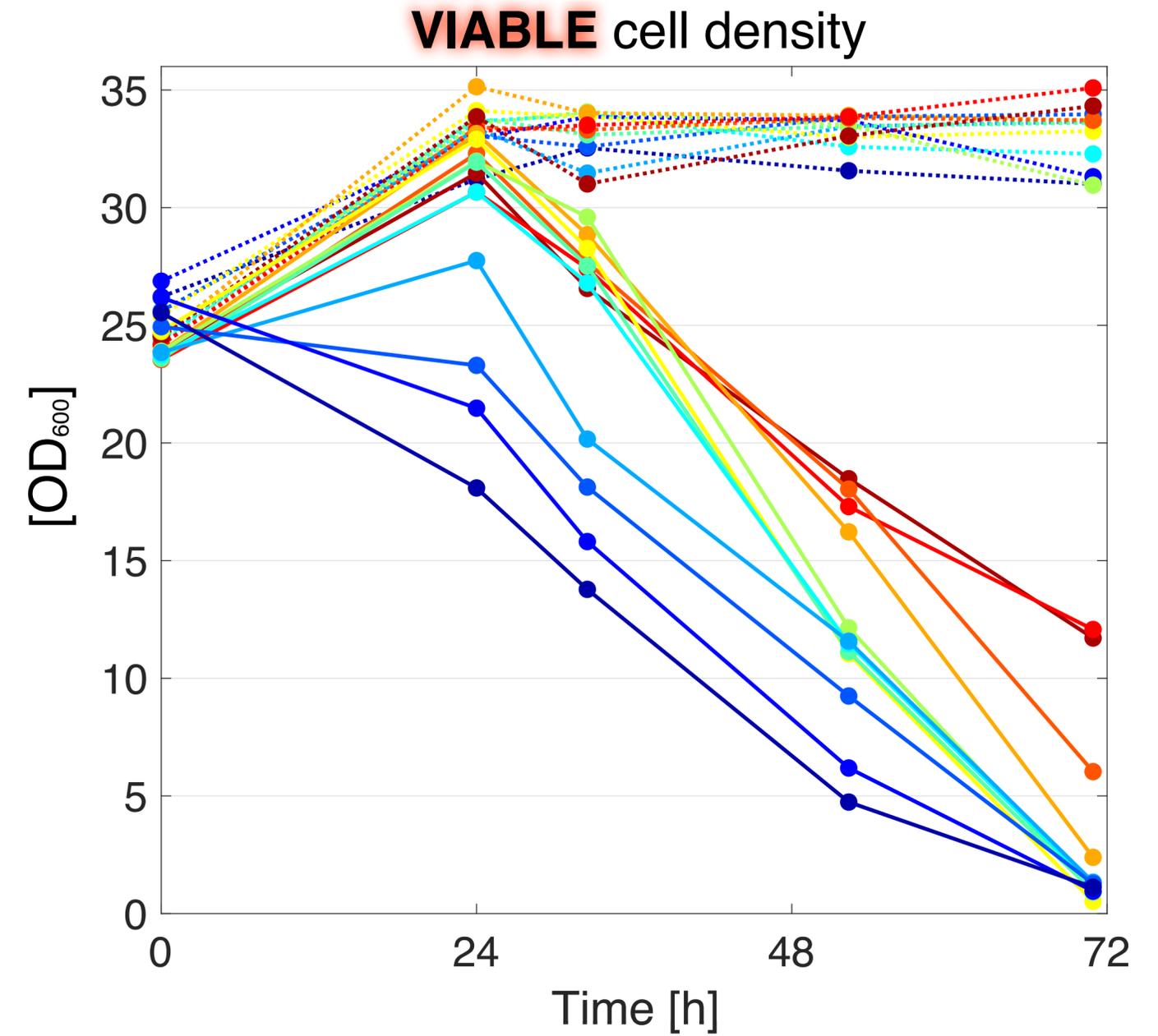
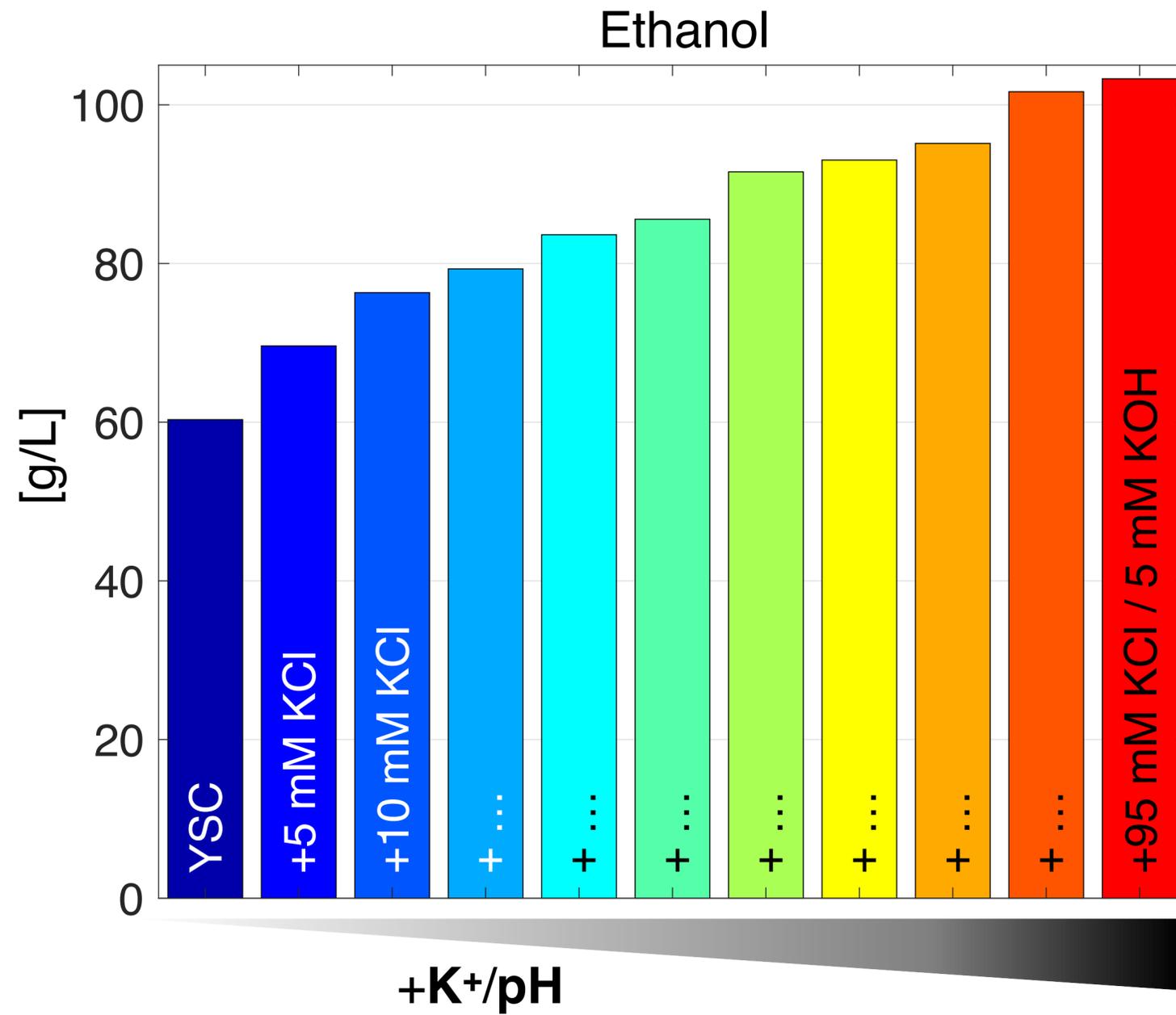
# Elevated $K^+$ /pH boost **cell tolerance** — NOT ethanol made **per-cell**



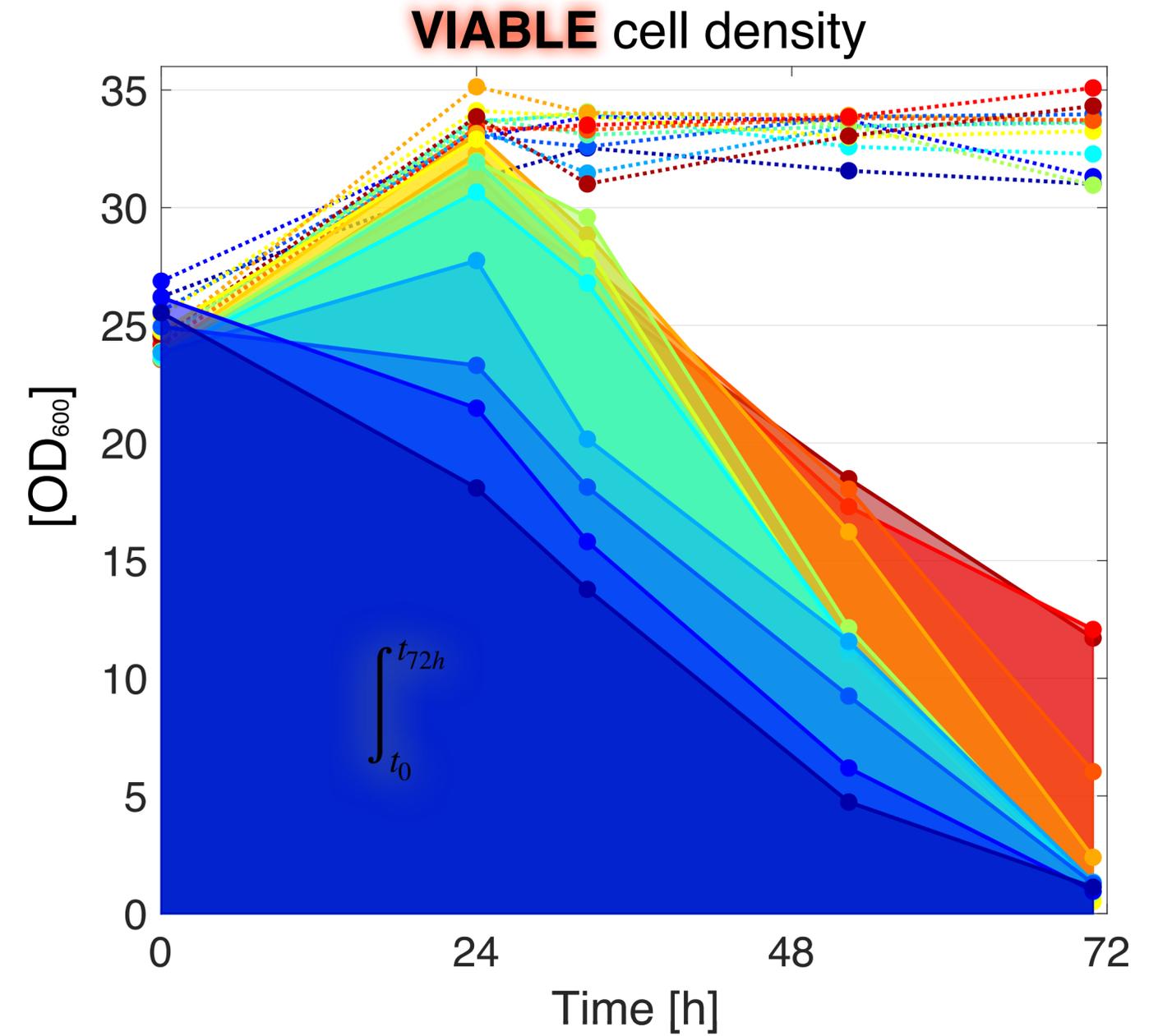
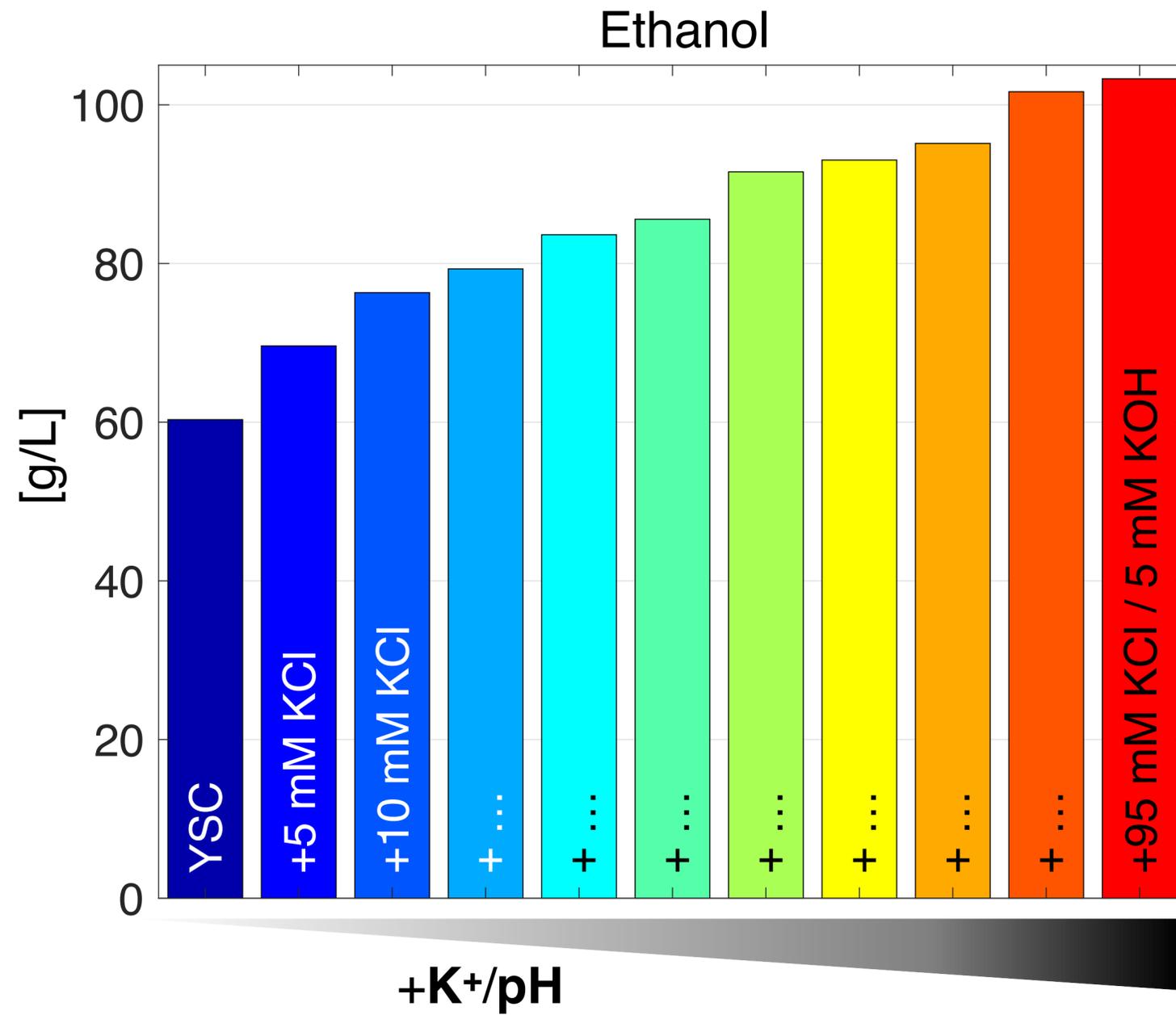
# Elevated $K^+$ /pH boost **cell tolerance** — NOT ethanol made **per-cell**



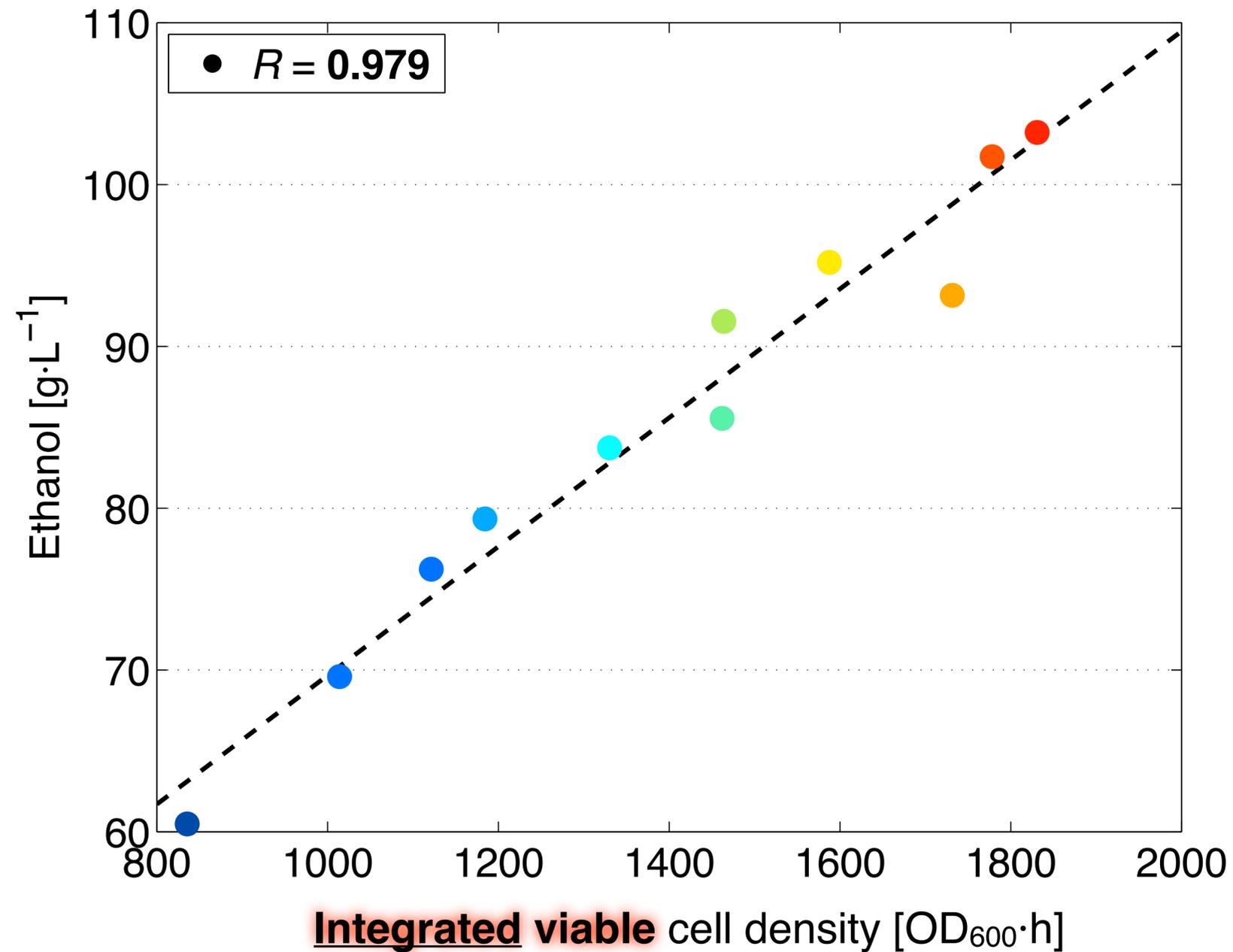
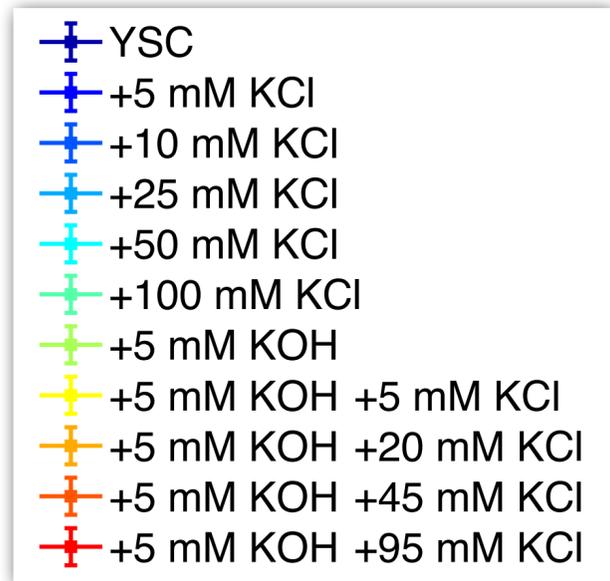
# Elevated $K^+$ /pH boost **cell tolerance** — NOT ethanol made **per-cell**



# Elevated $K^+$ /pH boost **cell tolerance** — NOT ethanol made **per-cell**



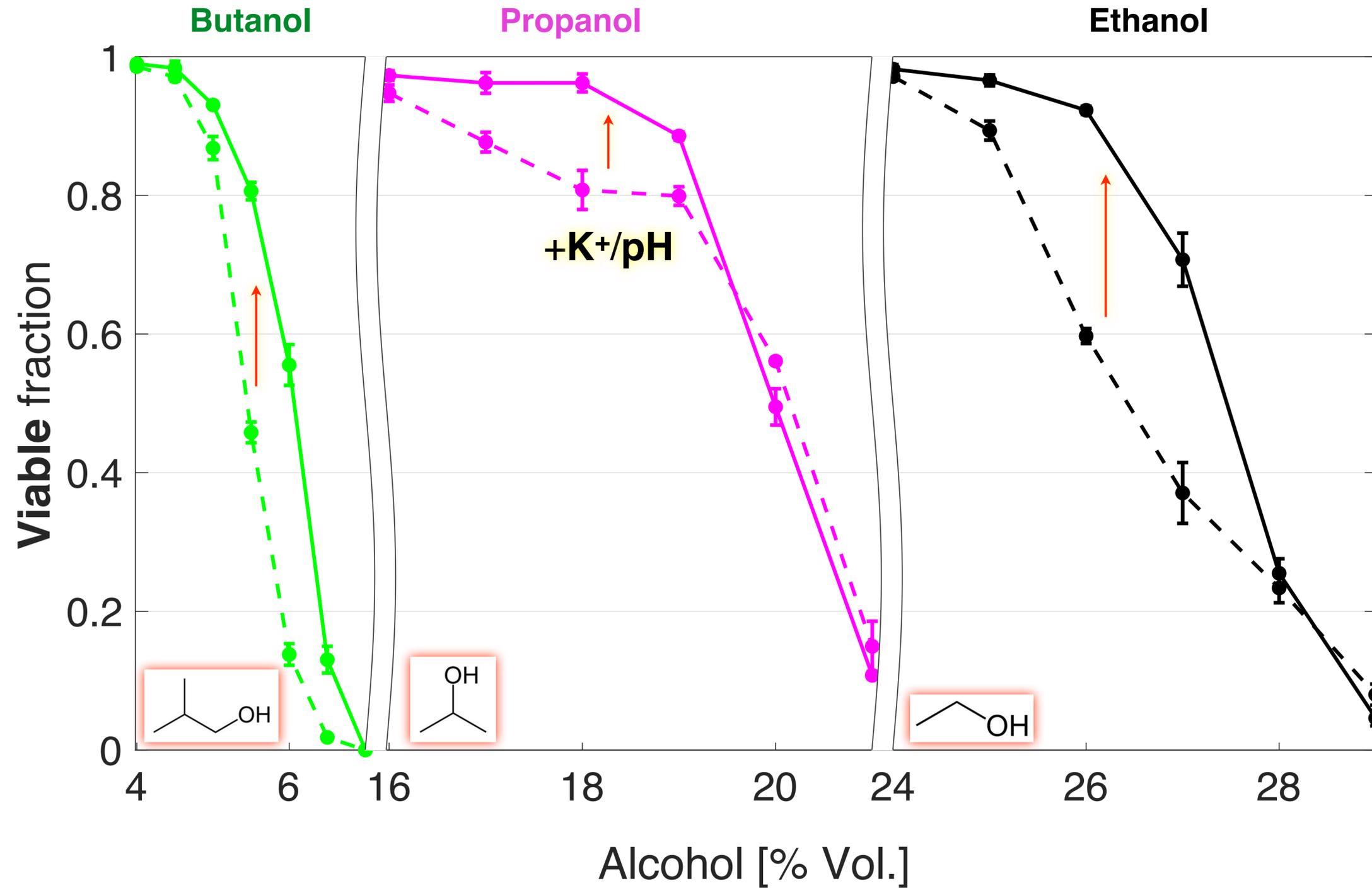
# External $K^+$ /pH directly control ethanol **tolerance** and **production**



## Straight line reveals:

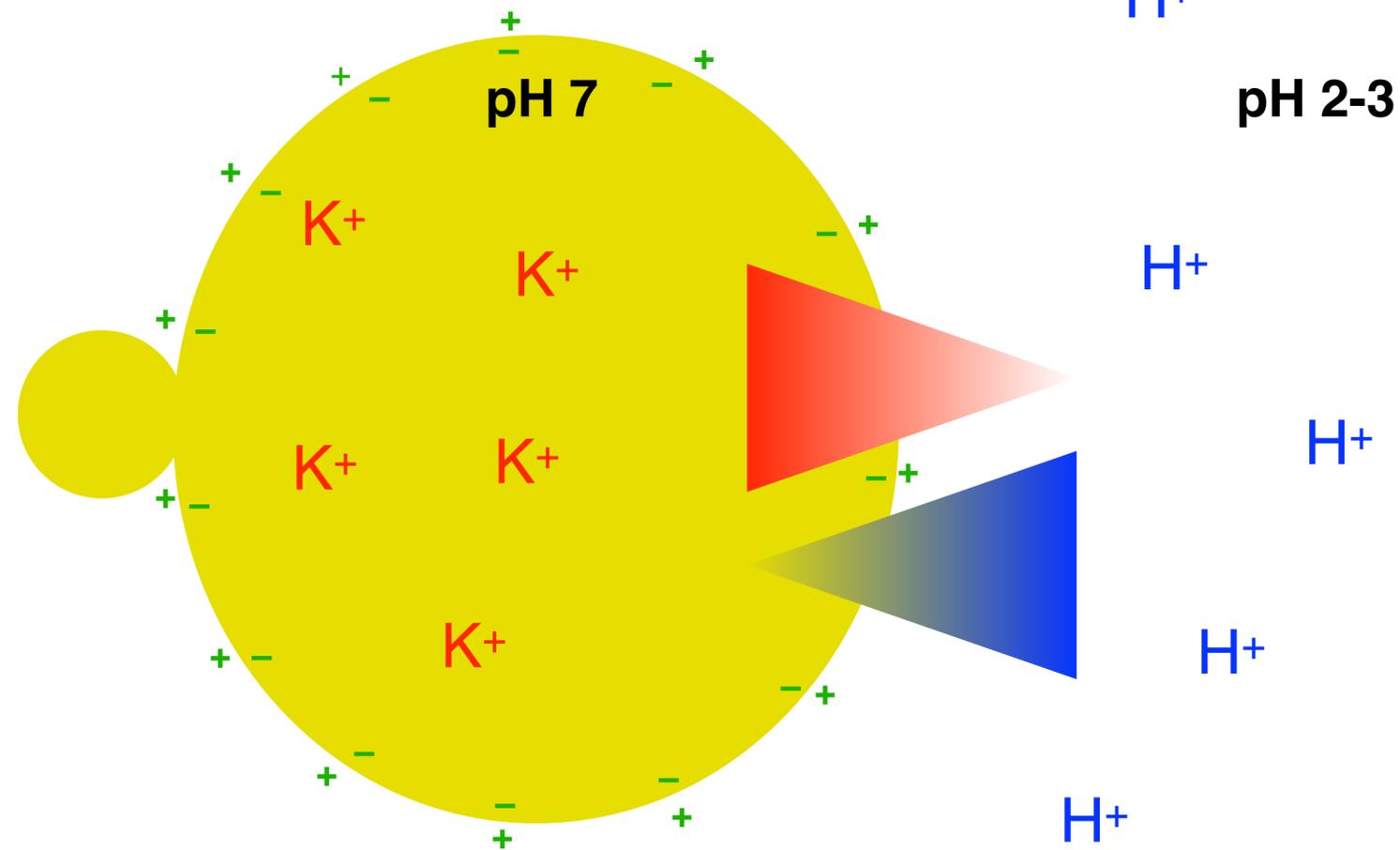
- Ethanol **production per-cell** remains **same**
- **Only** population **tolerance** / endurance **is varied**  
→ directly determines ethanol produced

# K<sup>+</sup>/pH control yeast **tolerance** to ethanol **many alcohols**



# Why $K^+$ /pH specifically? Form **gradients** of yeast **membrane potential**

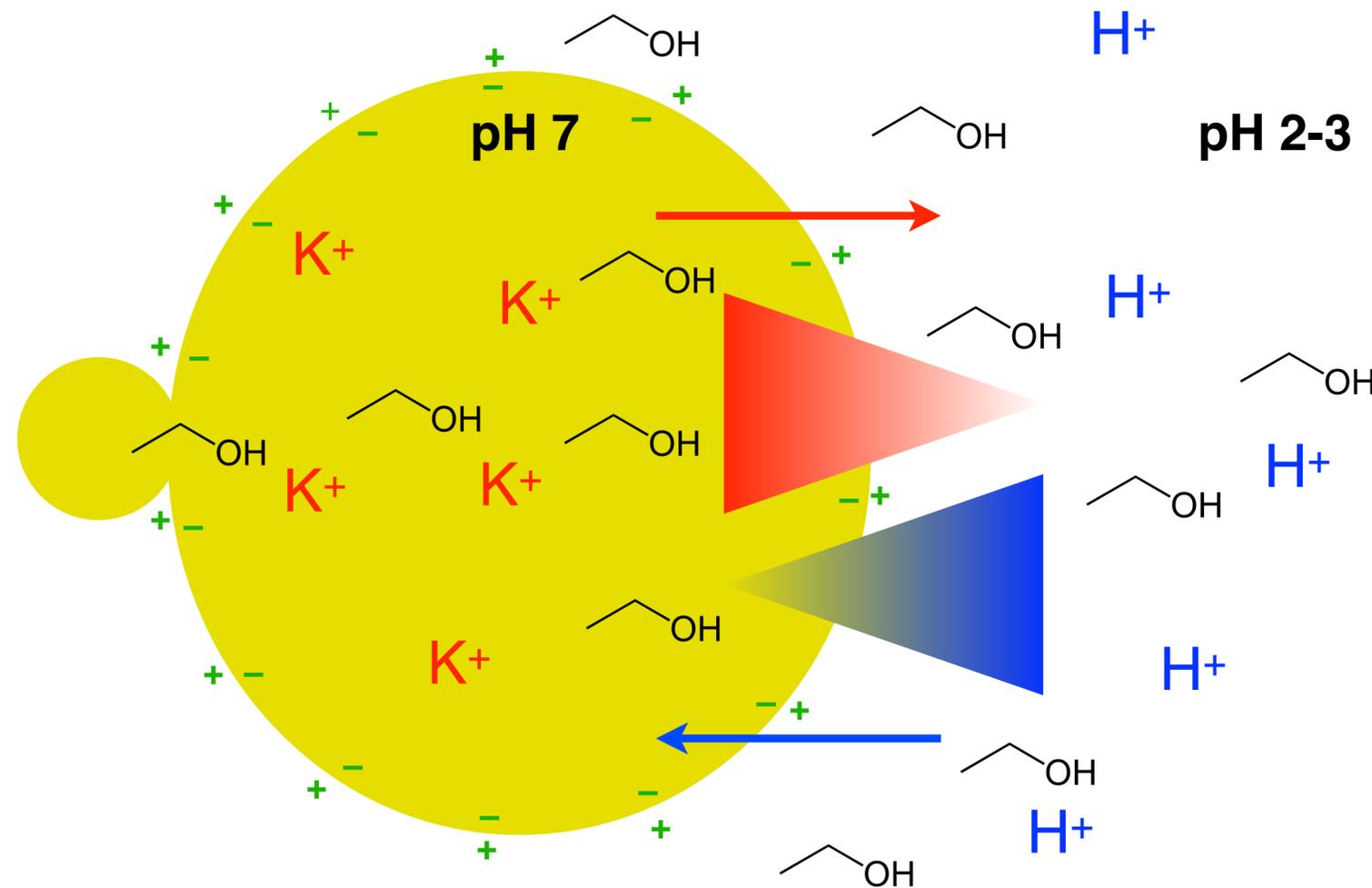
---



Normally:

- I.  $H^+$  and  $K^+$  membrane pumps maintain:  
Internal [pH 7 + high  $K^+$ ]  
External [pH 3 + low  $K^+$ ]  
→  $H^+$  /  $K^+$  gradients **charge membrane**

# Why $K^+$ / $pH$ specifically? Form **gradients** of yeast **membrane potential**



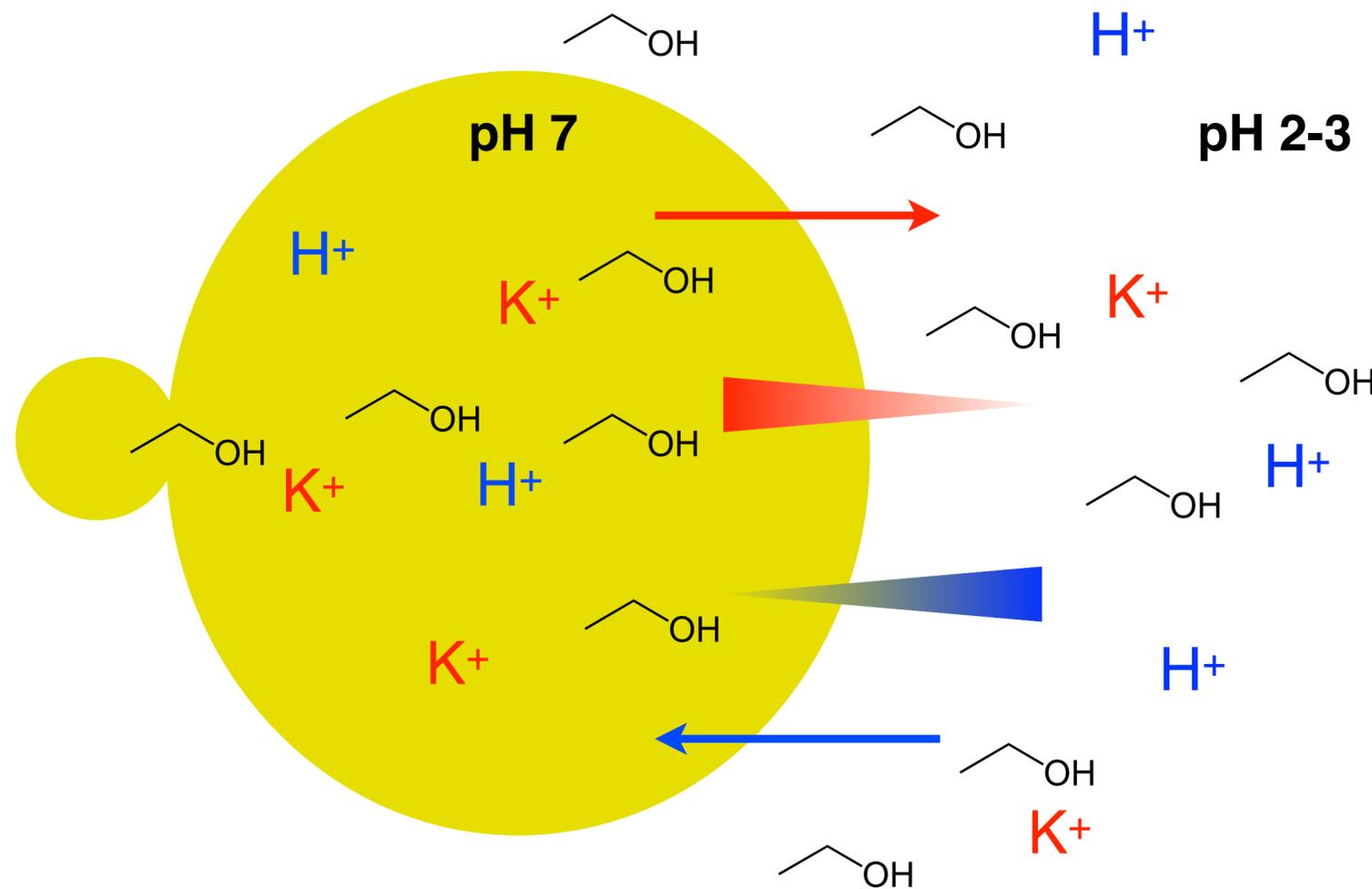
Normally:

1.  $H^+$  and  $K^+$  membrane pumps maintain:  
Internal [ $pH\ 7$  + high  $K^+$ ]  
External [ $pH\ 3$  + low  $K^+$ ]  
→  $H^+$  /  $K^+$  gradients **charge membrane**

When ethanol accumulates:

2. Membrane becomes permeable to  $H^+$  and  $K^+$  → ions leak

# Why $K^+$ / $pH$ specifically? Form **gradients** of yeast **membrane potential**



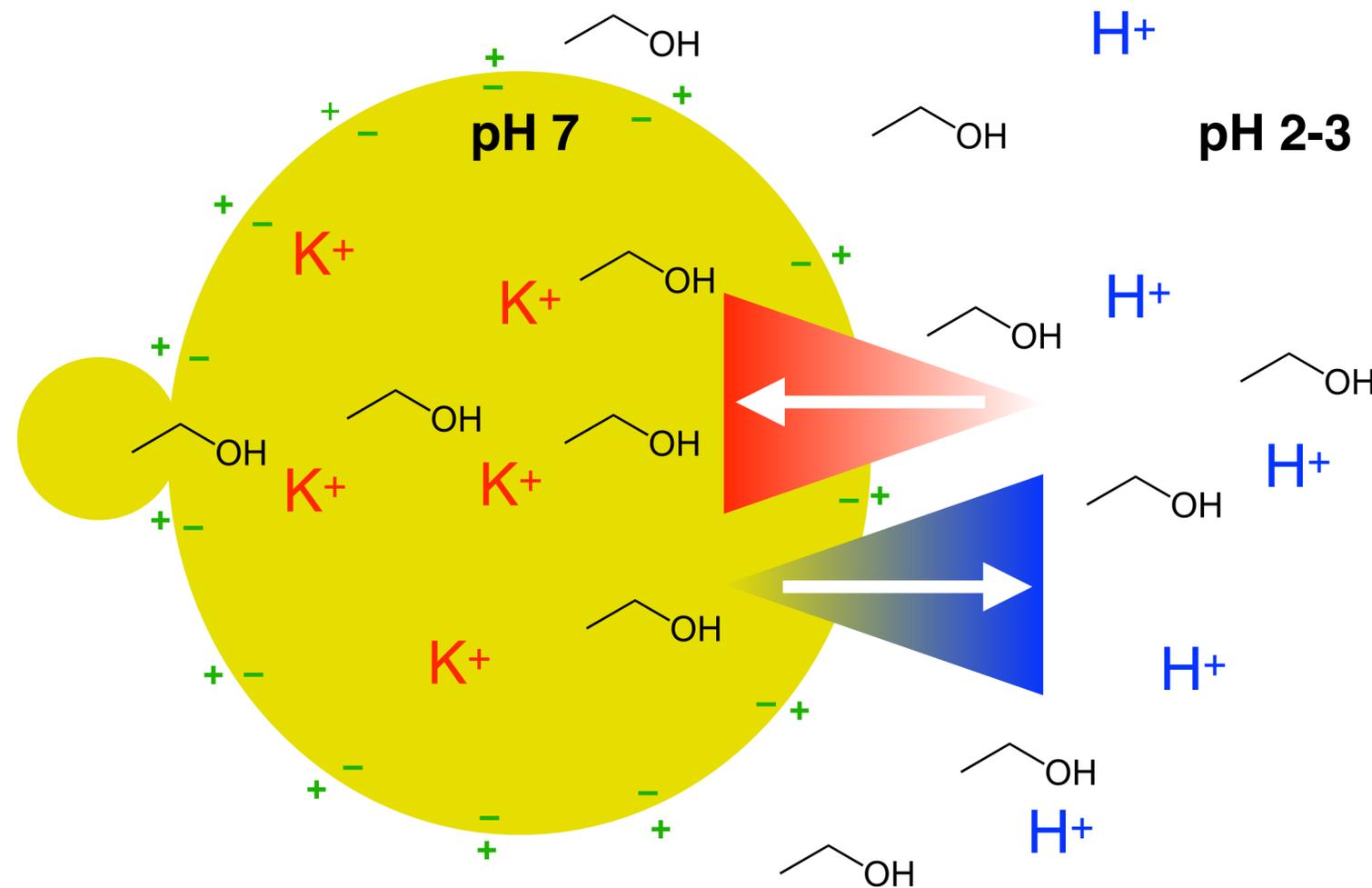
Normally:

1.  $H^+$  and  $K^+$  membrane pumps maintain:  
Internal [ $pH\ 7$  + high  $K^+$ ]  
External [ $pH\ 3$  + low  $K^+$ ]  
→  $H^+$ / $K^+$  gradients **charge membrane**

When ethanol accumulates:

2. Membrane becomes permeable to  $H^+$  and  $K^+$  → ions leak → **membrane gradients** dissipate → **cell death**

# Why $K^+$ / $pH$ specifically? Form **gradients** of yeast **membrane potential**



Normally:

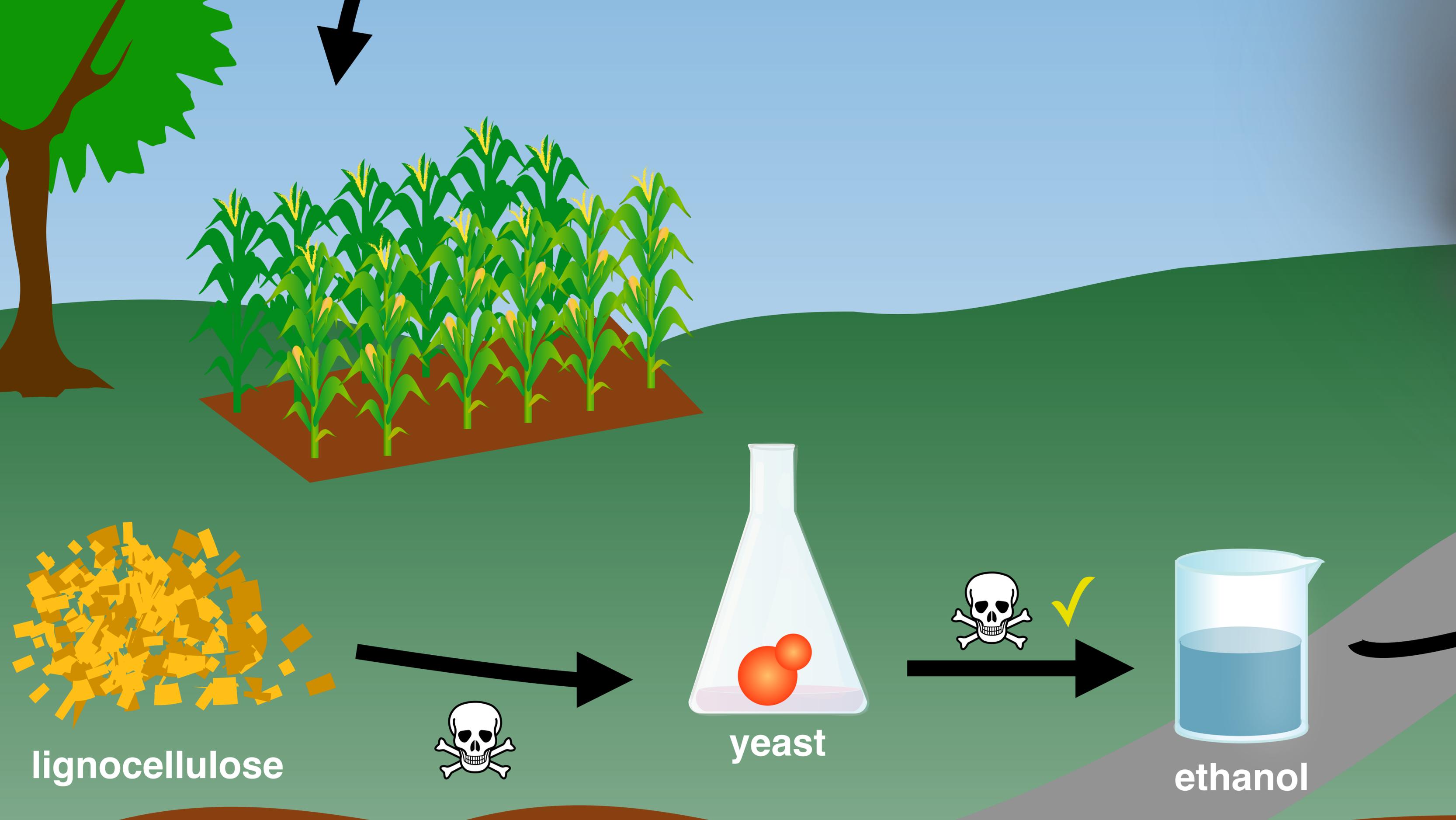
1.  $H^+$  and  $K^+$  membrane pumps maintain:  
Internal [ $pH\ 7$  + high  $K^+$ ]  
External [ $pH\ 3$  + low  $K^+$ ]  
→  $H^+$  /  $K^+$  gradients **charge membrane**

When ethanol accumulates:

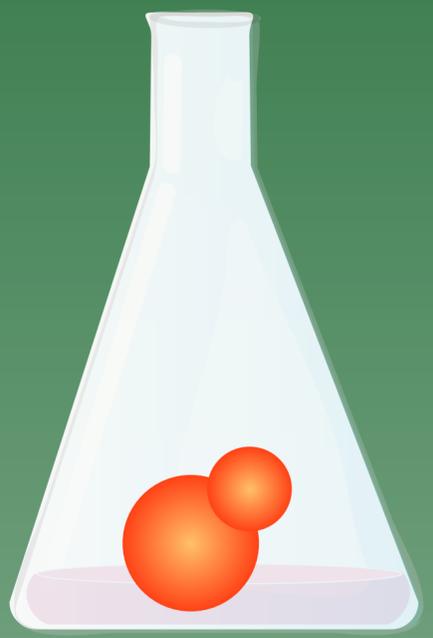
2. Membrane becomes permeable to  $H^+$  and  $K^+$  → ions leak → **membrane gradients** dissipate → **cell death**

Increasing external  $K^+$ / $pH$ :

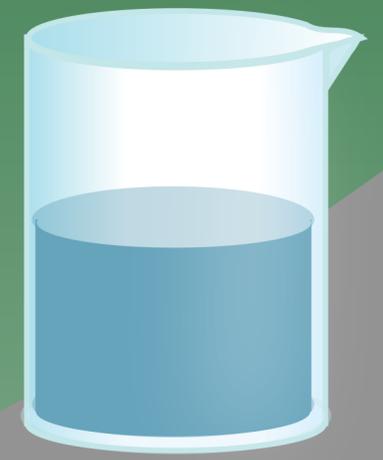
3. Assists membrane pumps with re-establishing gradients
4. **Membrane charge** is restored ✓



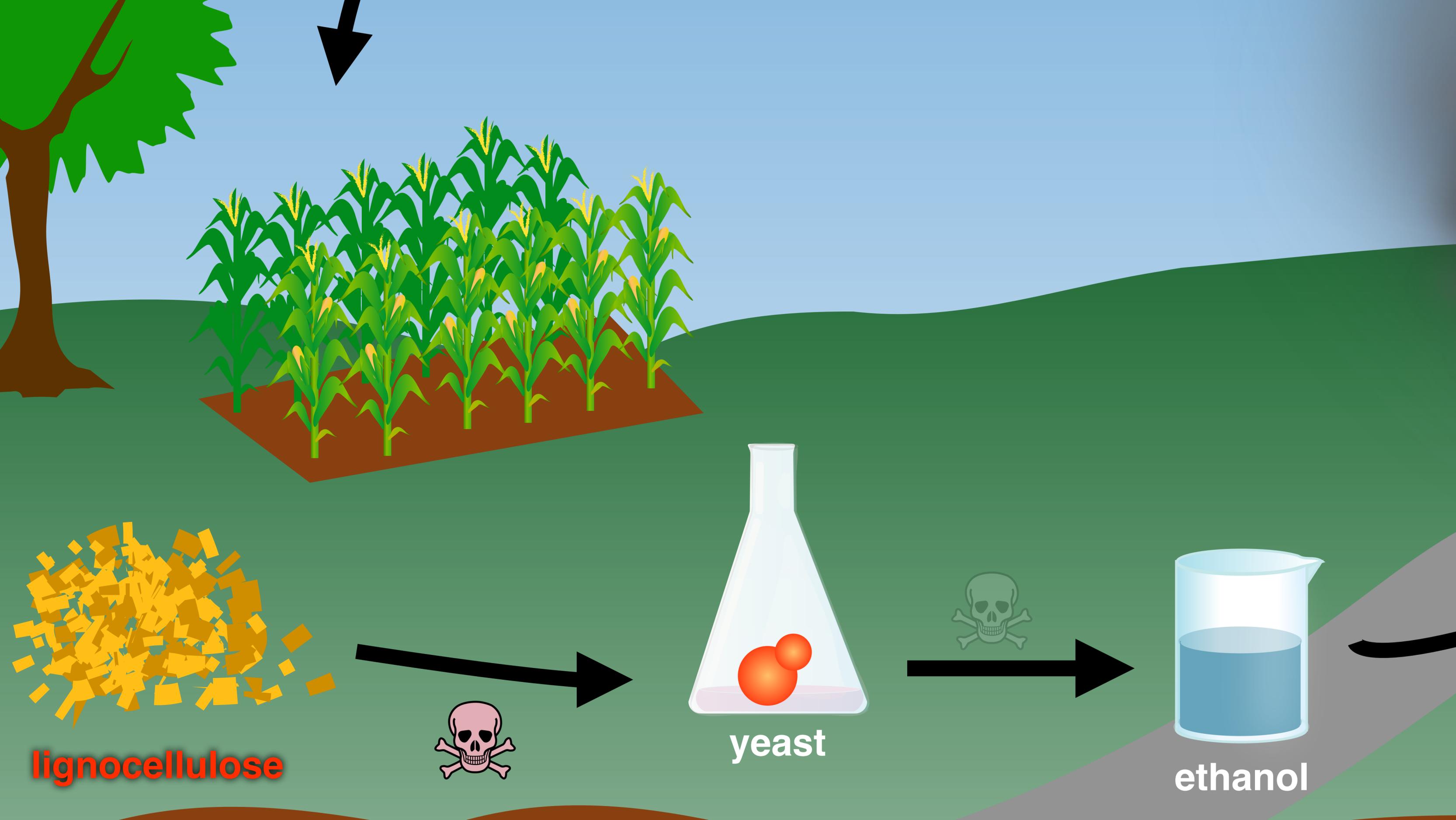
**lignocellulose**



**yeast**



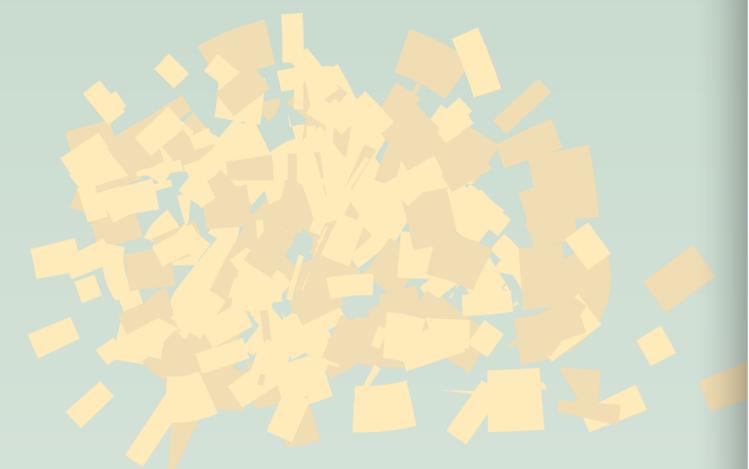
**ethanol**



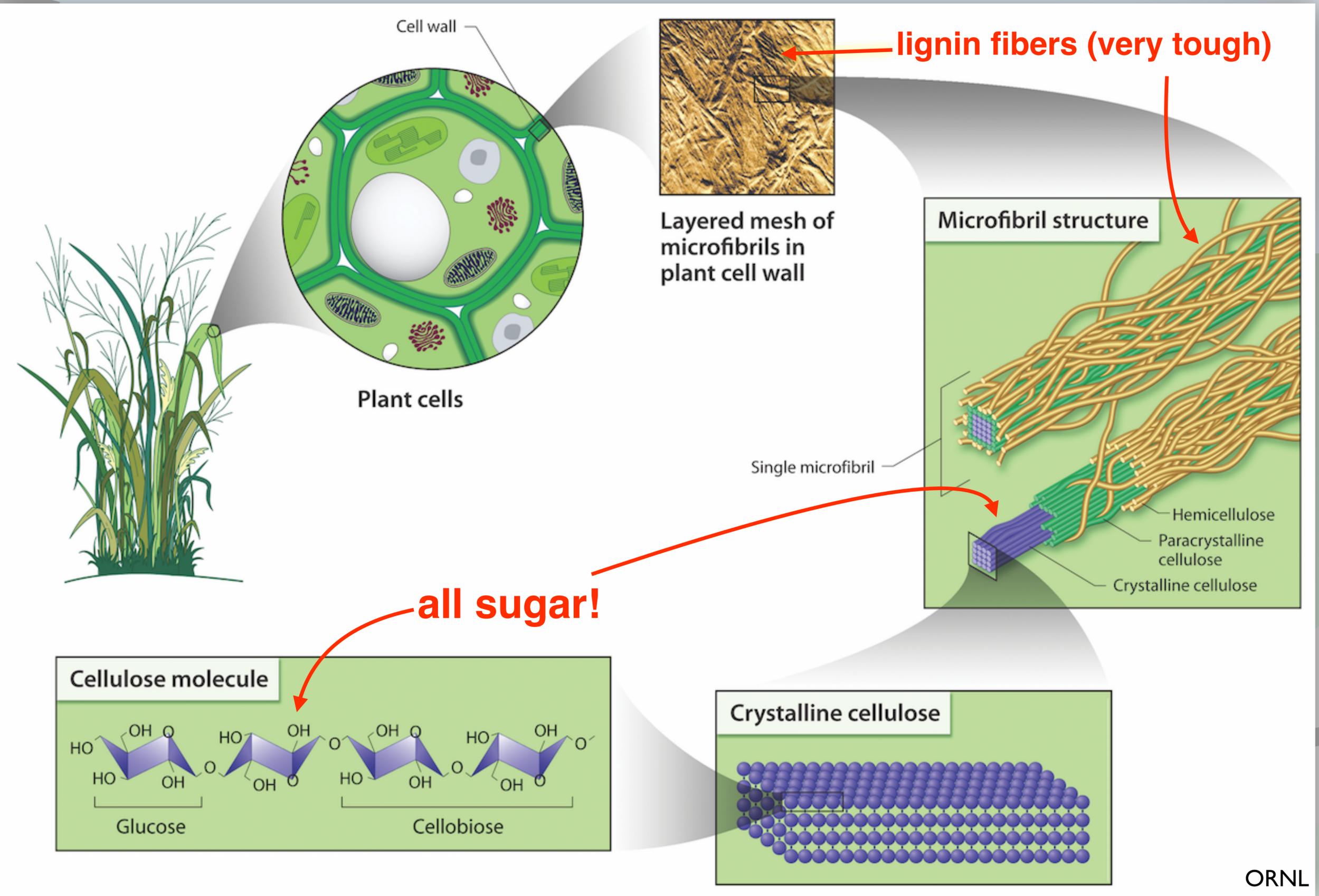
**lignocellulose**

**yeast**

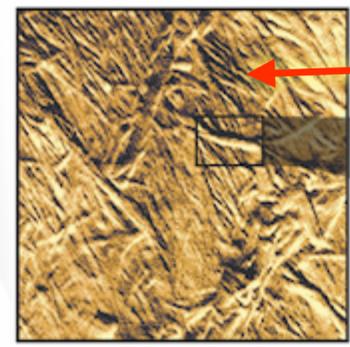
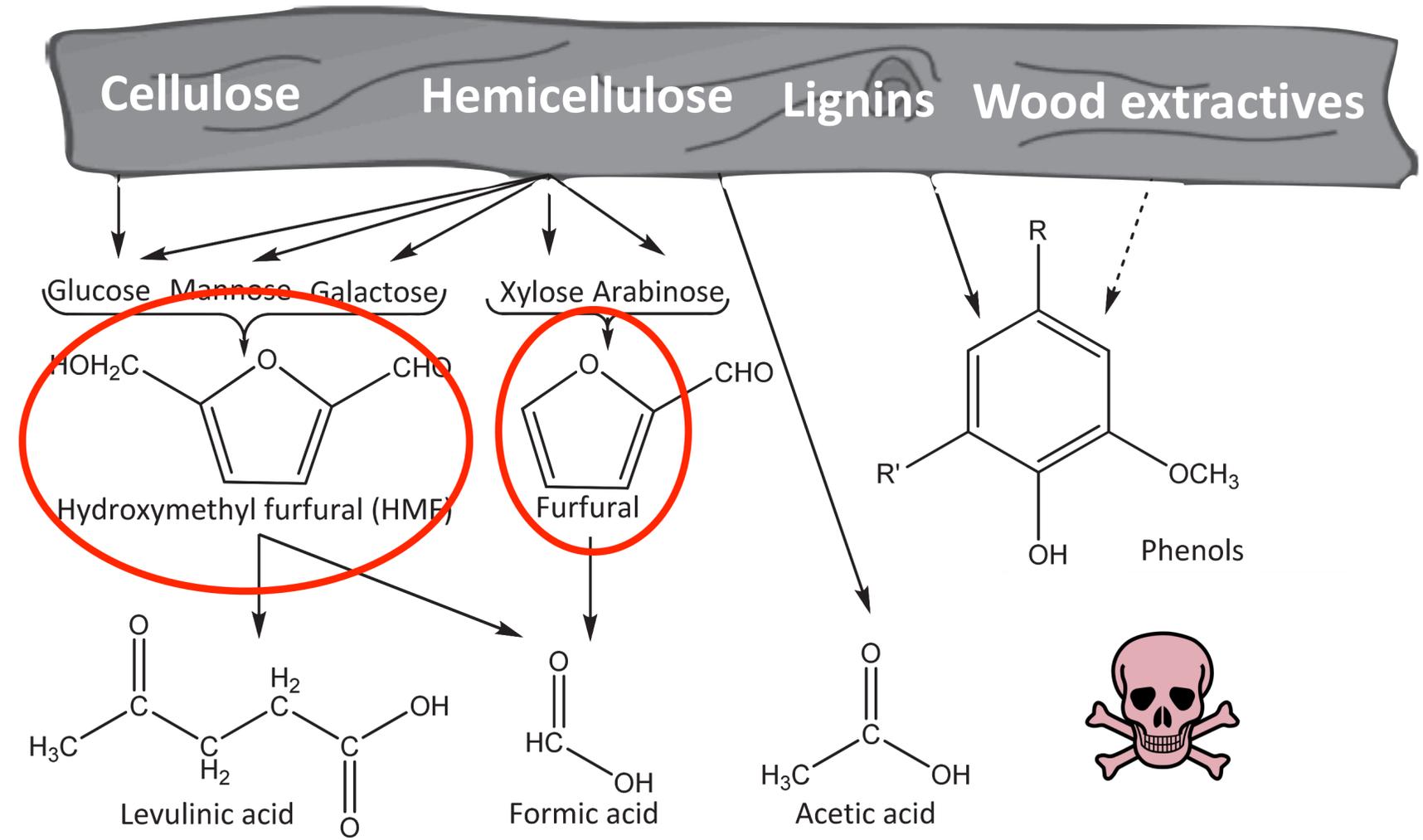
**ethanol**



**lignocellulose**

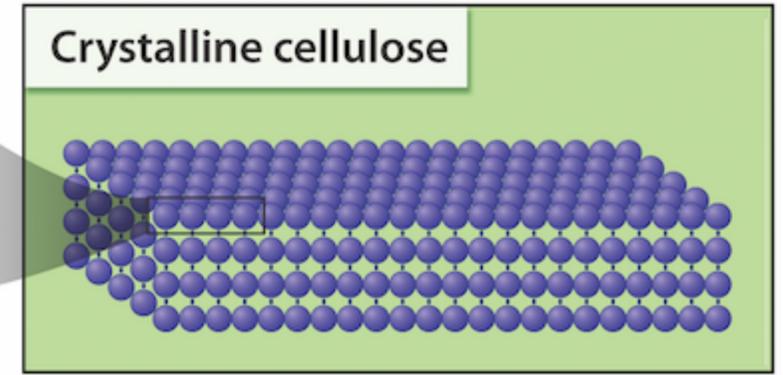
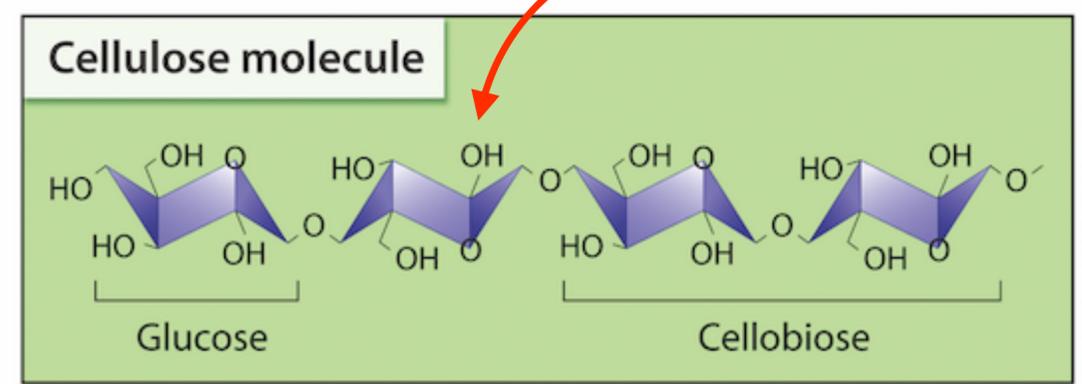
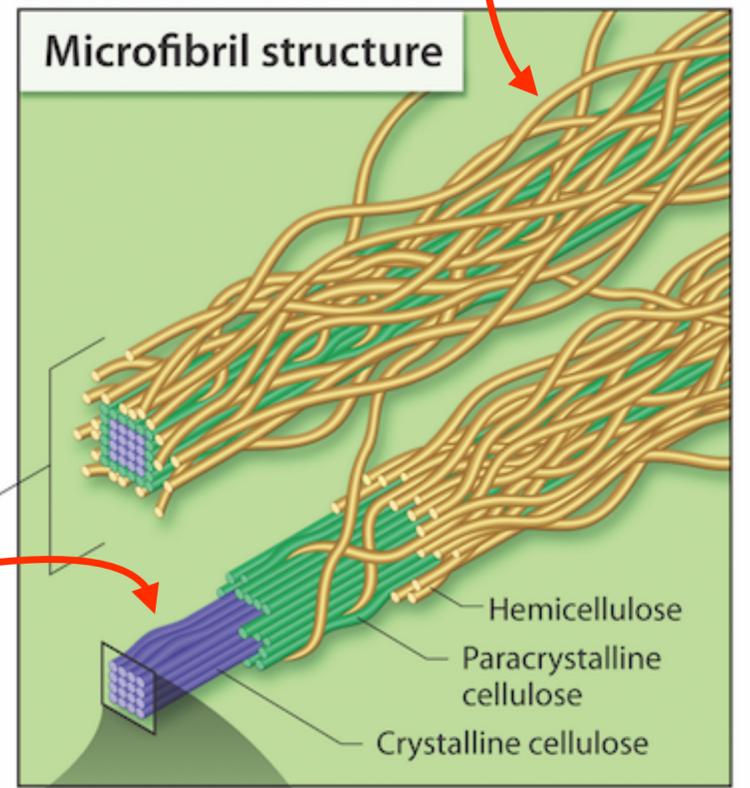


# Inhibitors universal to all lignocellulosic sources



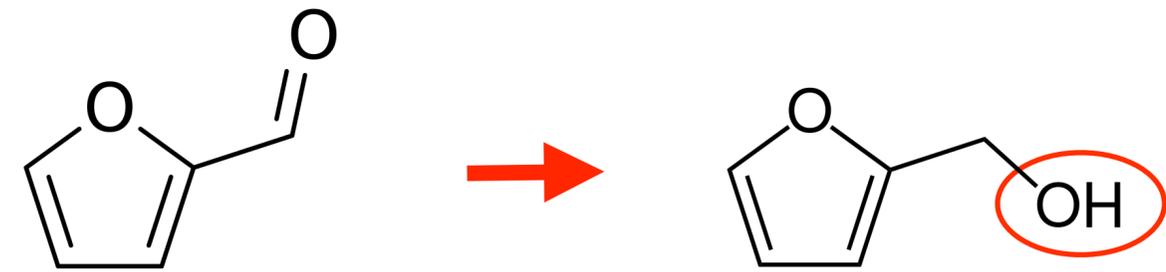
**lignin fibers (very tough)**

Layered mesh of microfibrils in plant cell wall



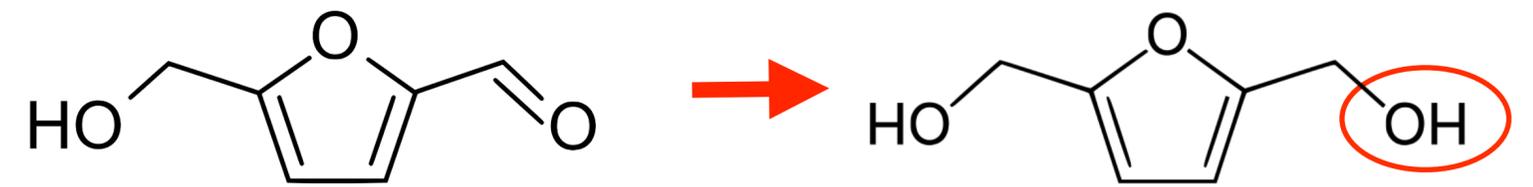
**lignocellulose**

# Alcohol forms of furfural, HMF less toxic → respond to $K^+$ /pH



Furfural

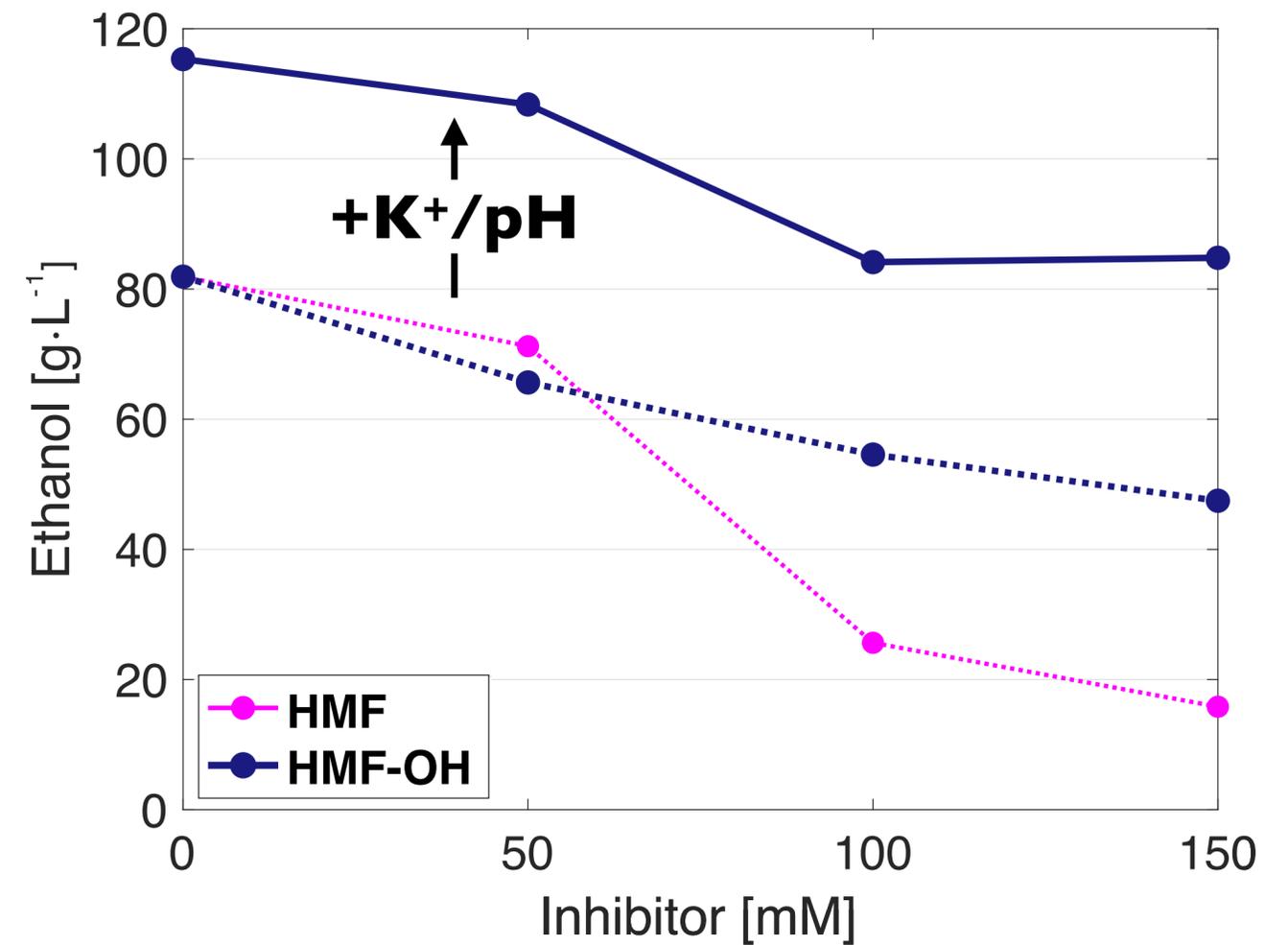
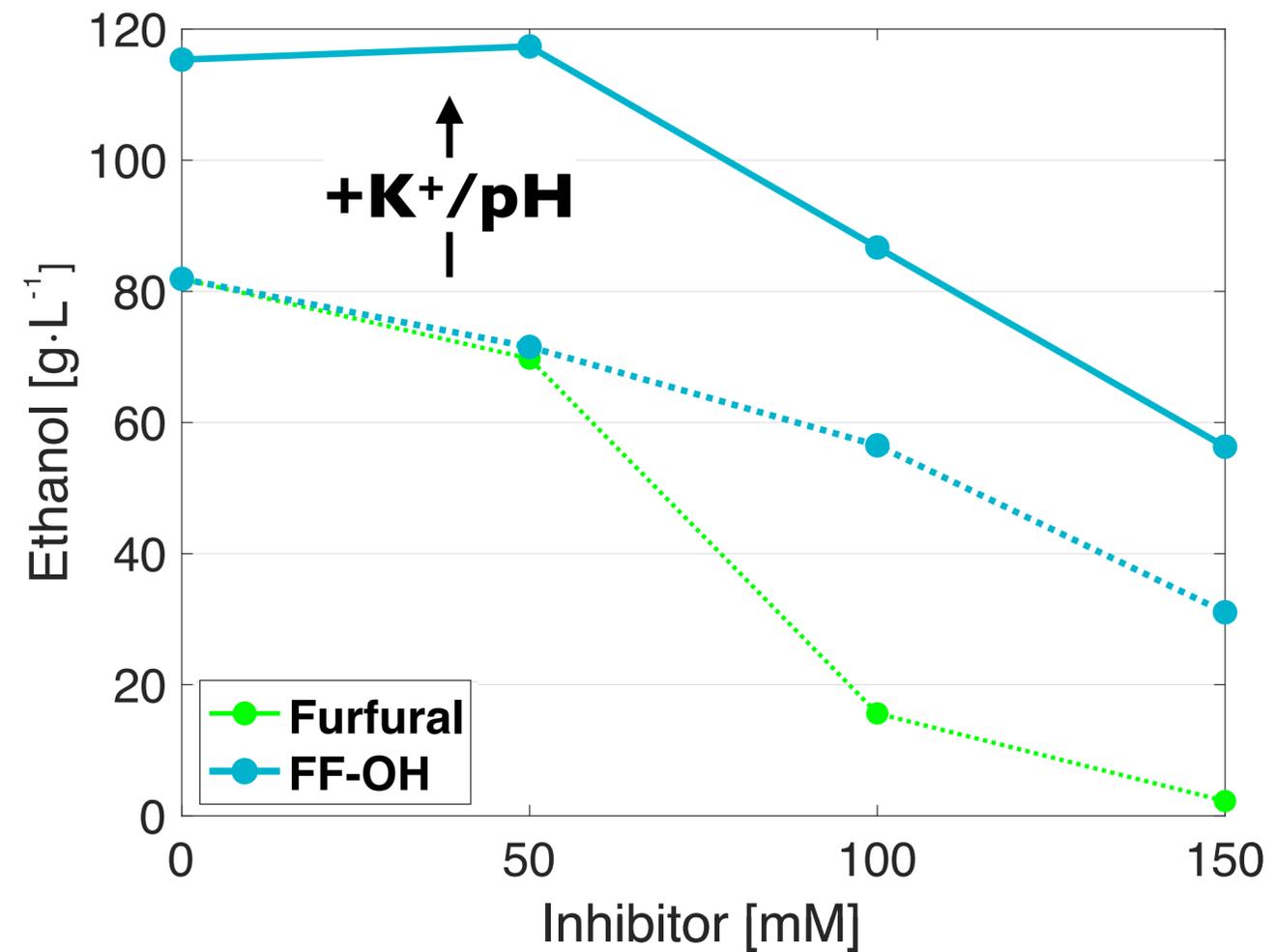
FF-OH



HMF

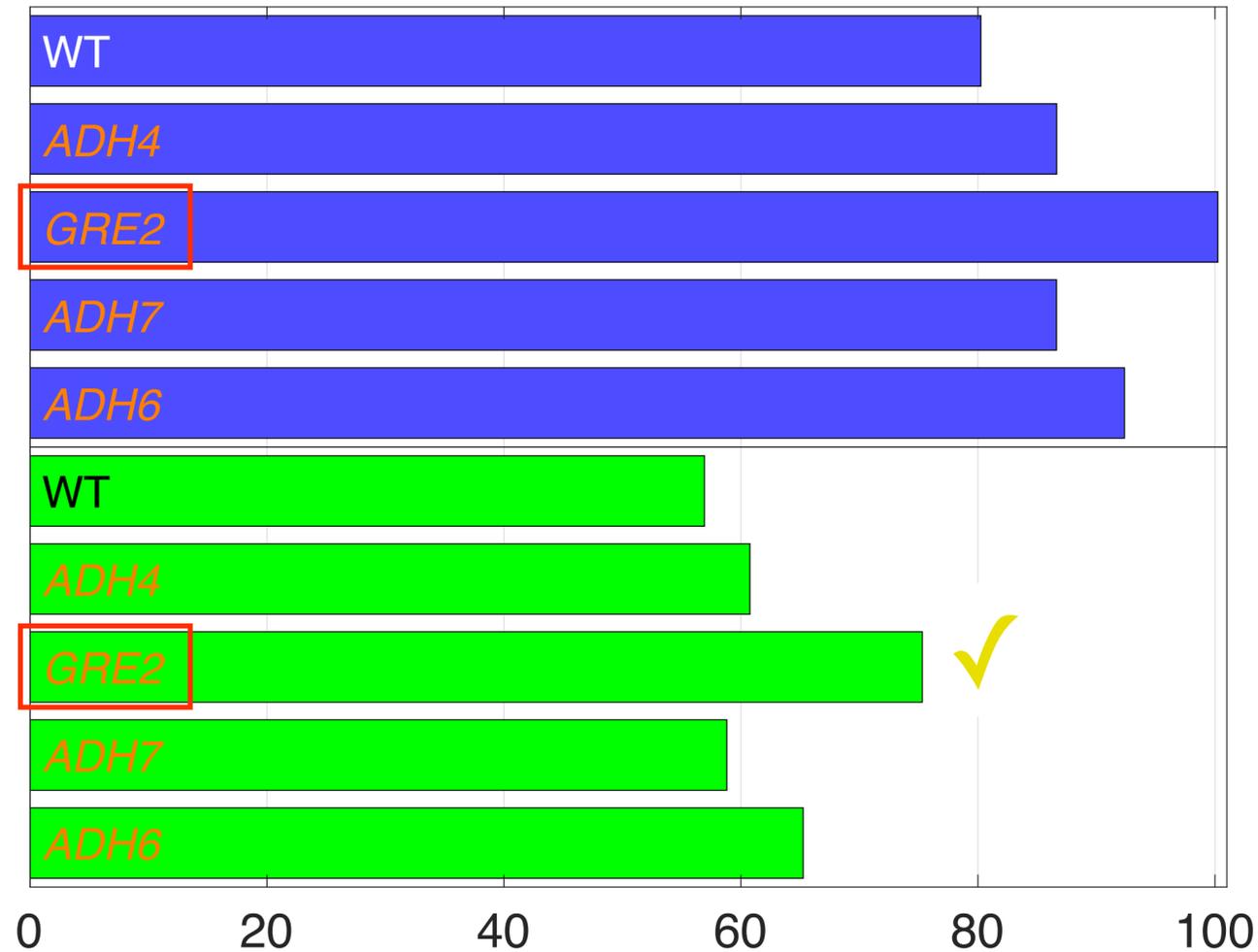
HMF-OH

Can we engineer this conversion within yeast?



# Reductase **GRE2** enhances **conversion** of **furfural, HMF**

6 g/L furfural +  
6 g/L HMF



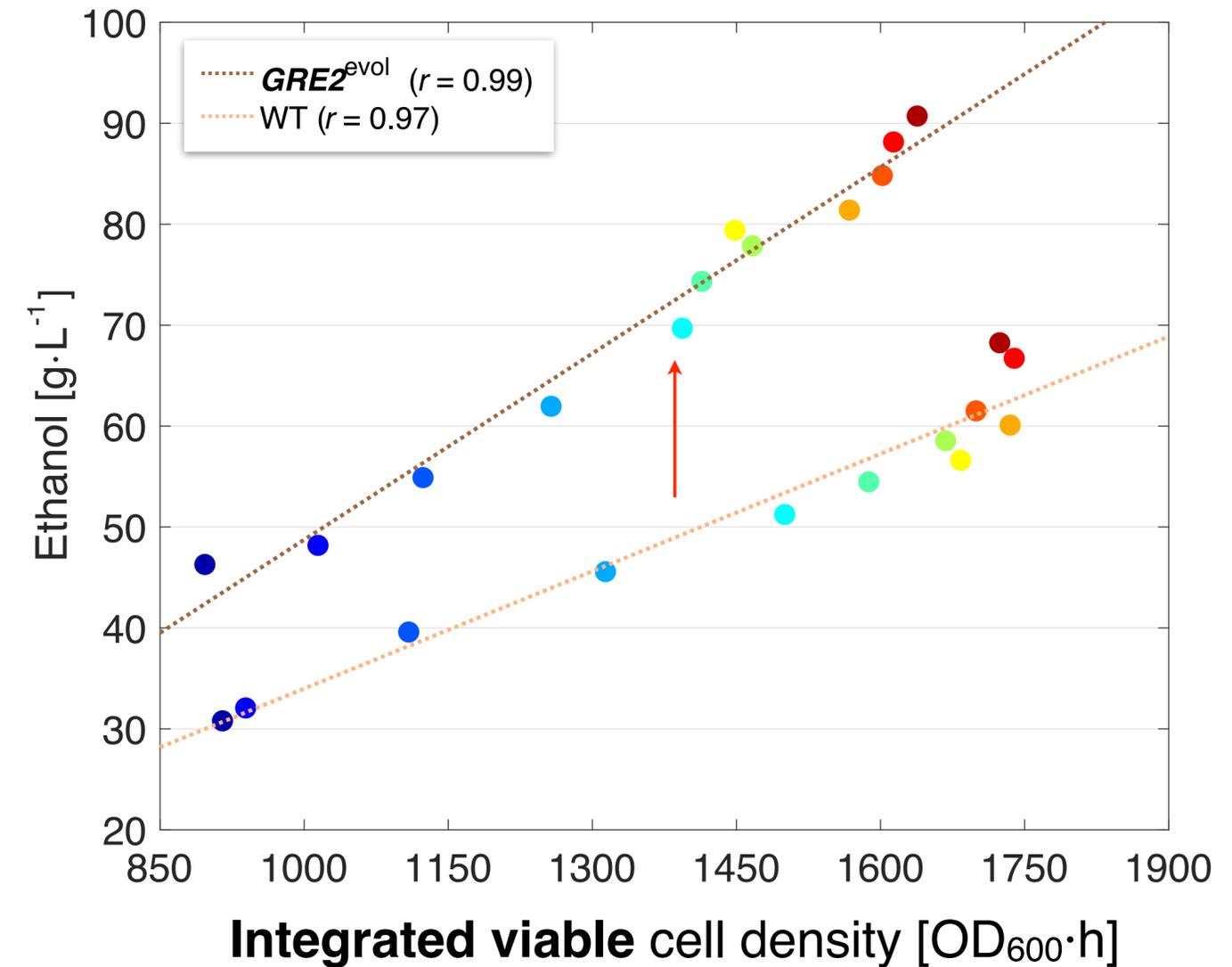
8 g/L furfural +  
8 g/L HMF

Ethanol [g·L<sup>-1</sup>]

8 g/L furfural +  
8 g/L HMF

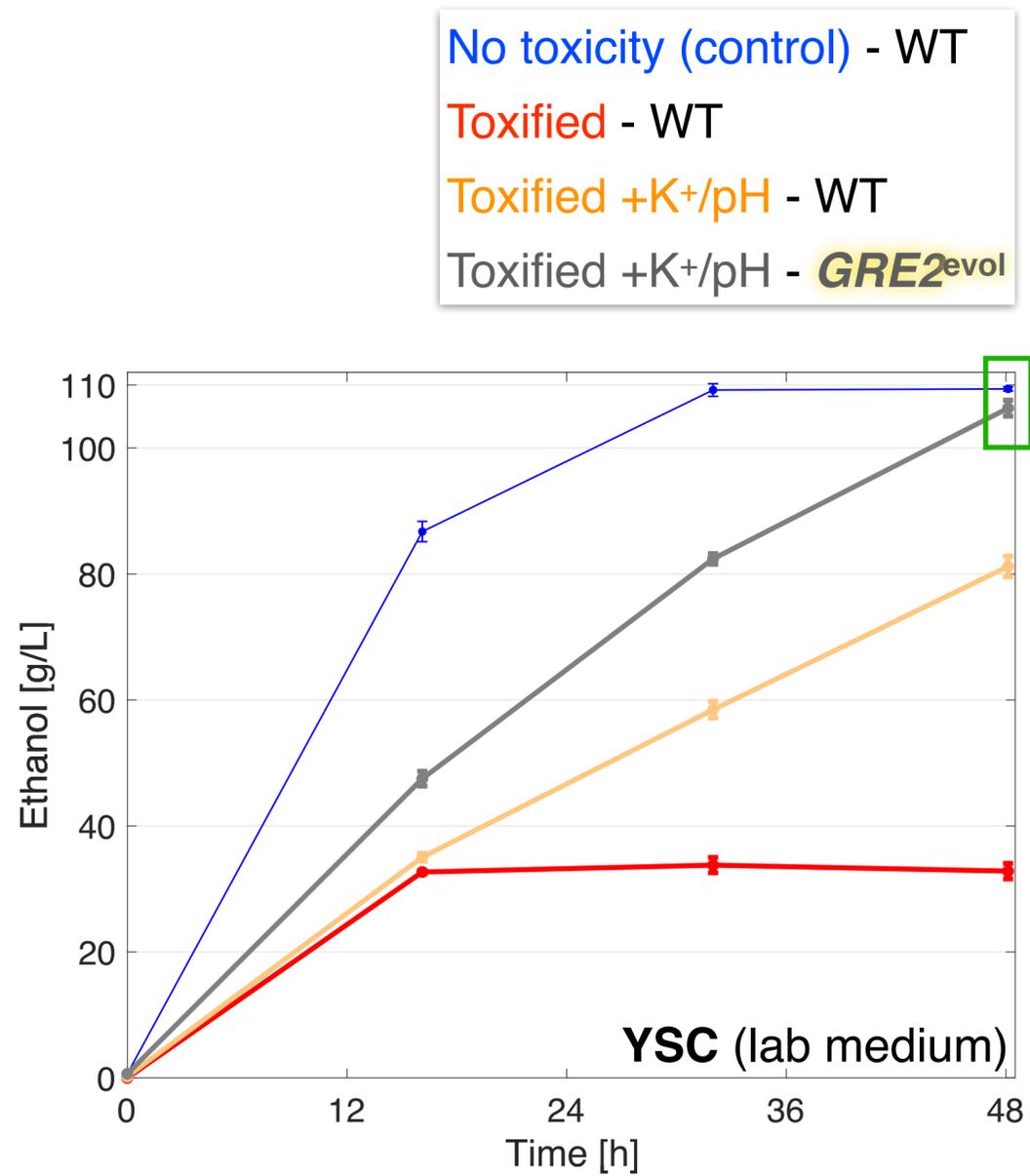


**Evolved  
GRE2**  
for even  
**higher**  
detoxification

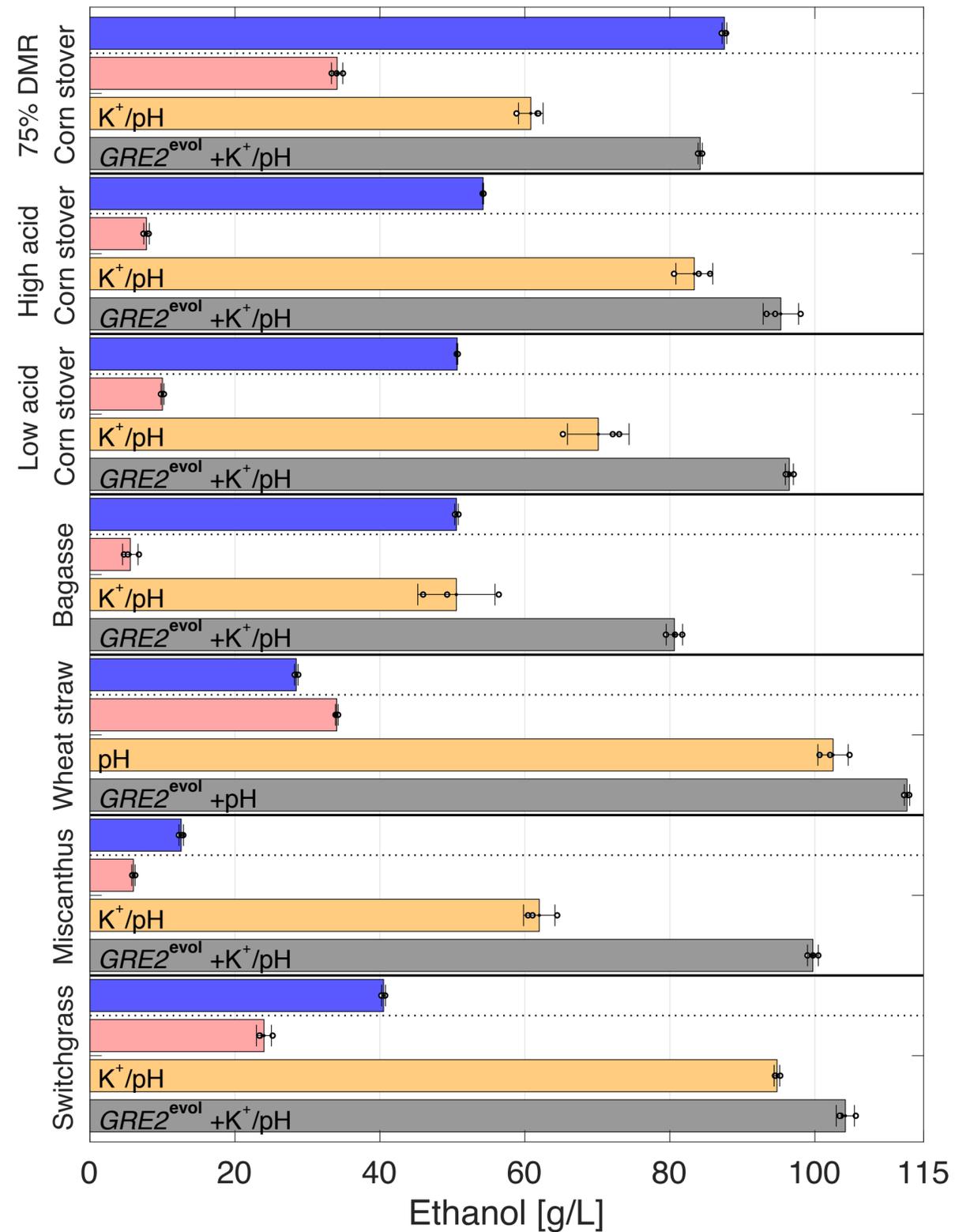


Increased **per-cell ethanol** production

# **GRE2<sup>evol</sup>** combined with **K<sup>+</sup>/pH** robustly ferments **genuine feedstocks**\*

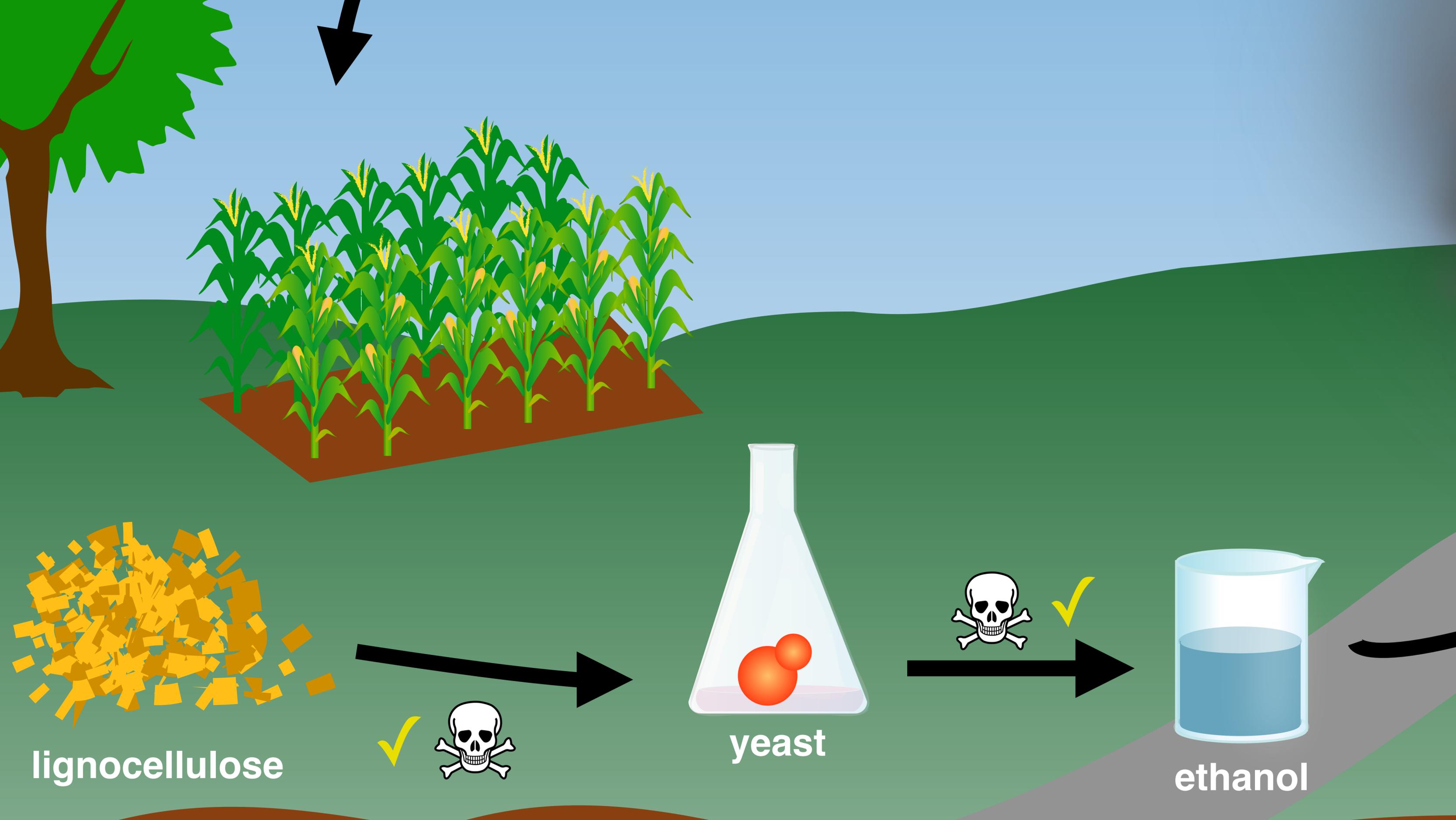


→ Matches performance in “clean sugar”



\* All genuine feedstocks low in toxicity → **toxification** with furfural, HMF (and sugar) required

→ **Single strain handles diversity of cellulosic feedstocks**



**lignocellulose**

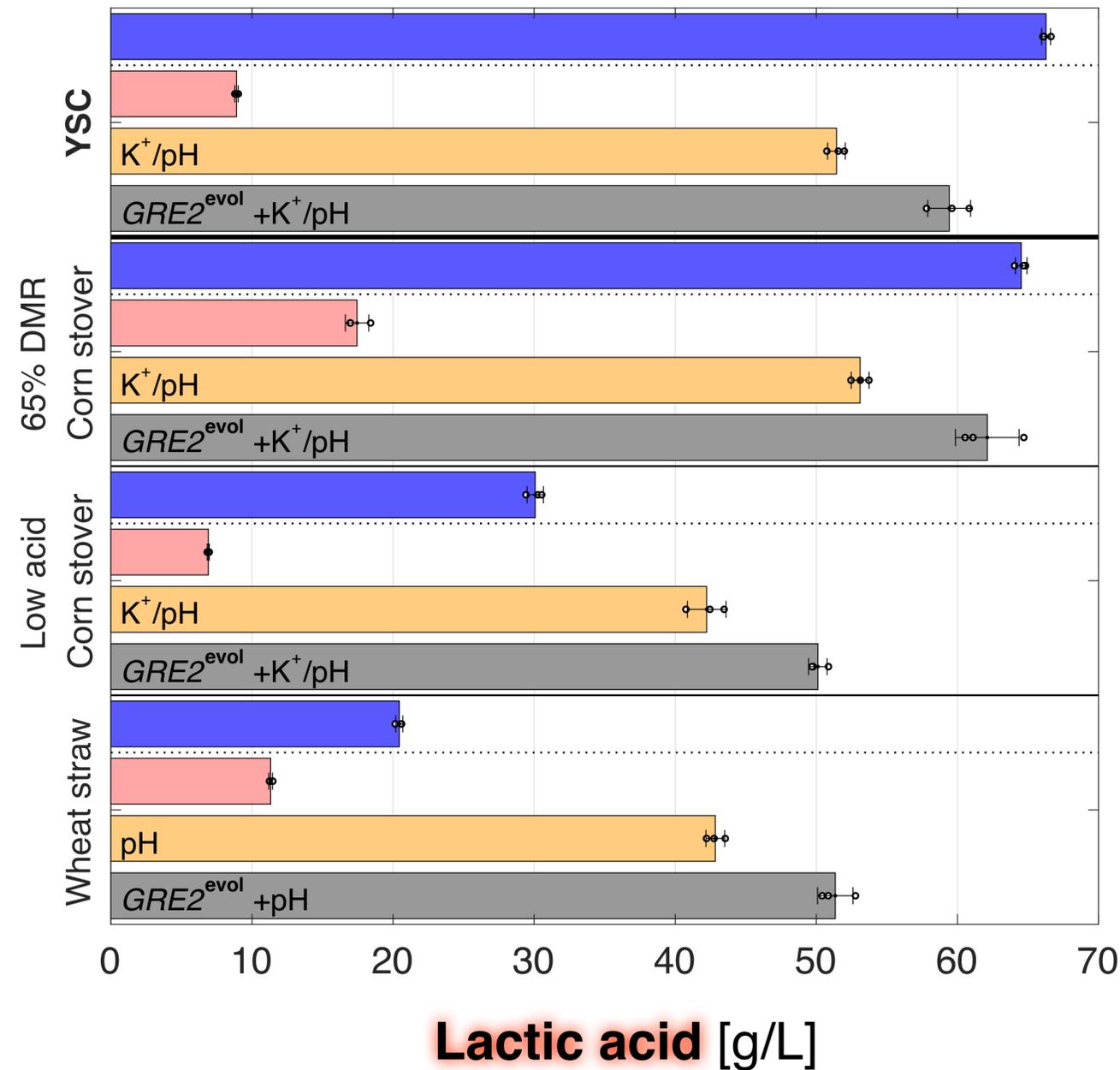
**yeast**

**ethanol**

# **GRE2<sup>evol</sup>** and **K<sup>+</sup>/pH** applied to production of **bio-plastic** precursor

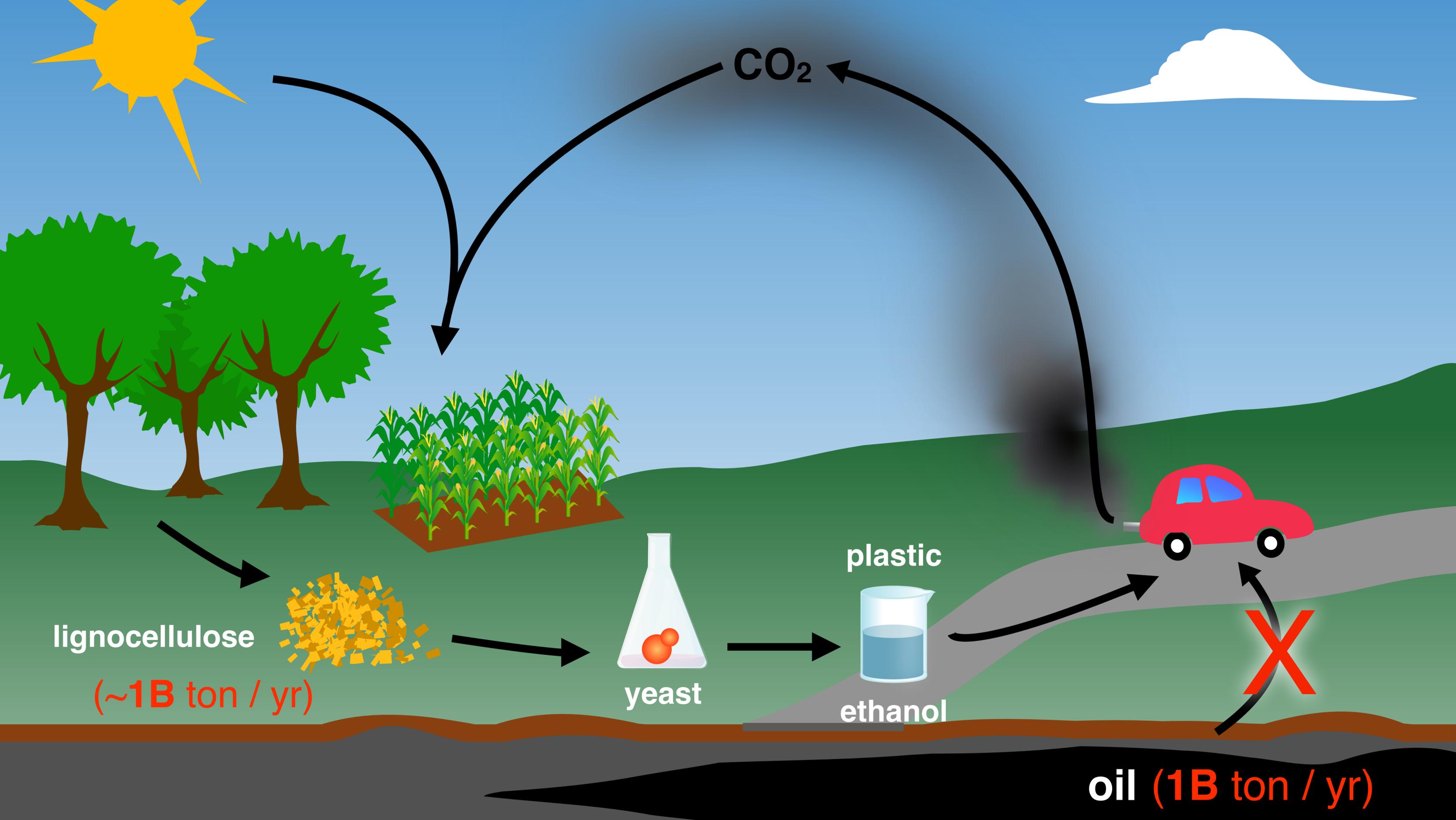


**PLA**  
polylactic  
acid



No toxicity (control) - WT  
Toxicified - WT  
Toxicified +K<sup>+</sup>/pH - WT  
Toxicified +K<sup>+</sup>/pH - *GRE2<sup>evol</sup>*

- Single strain
- Diversity of cellulosic feedstocks
- Matches “clean sugar” performance



CO<sub>2</sub>

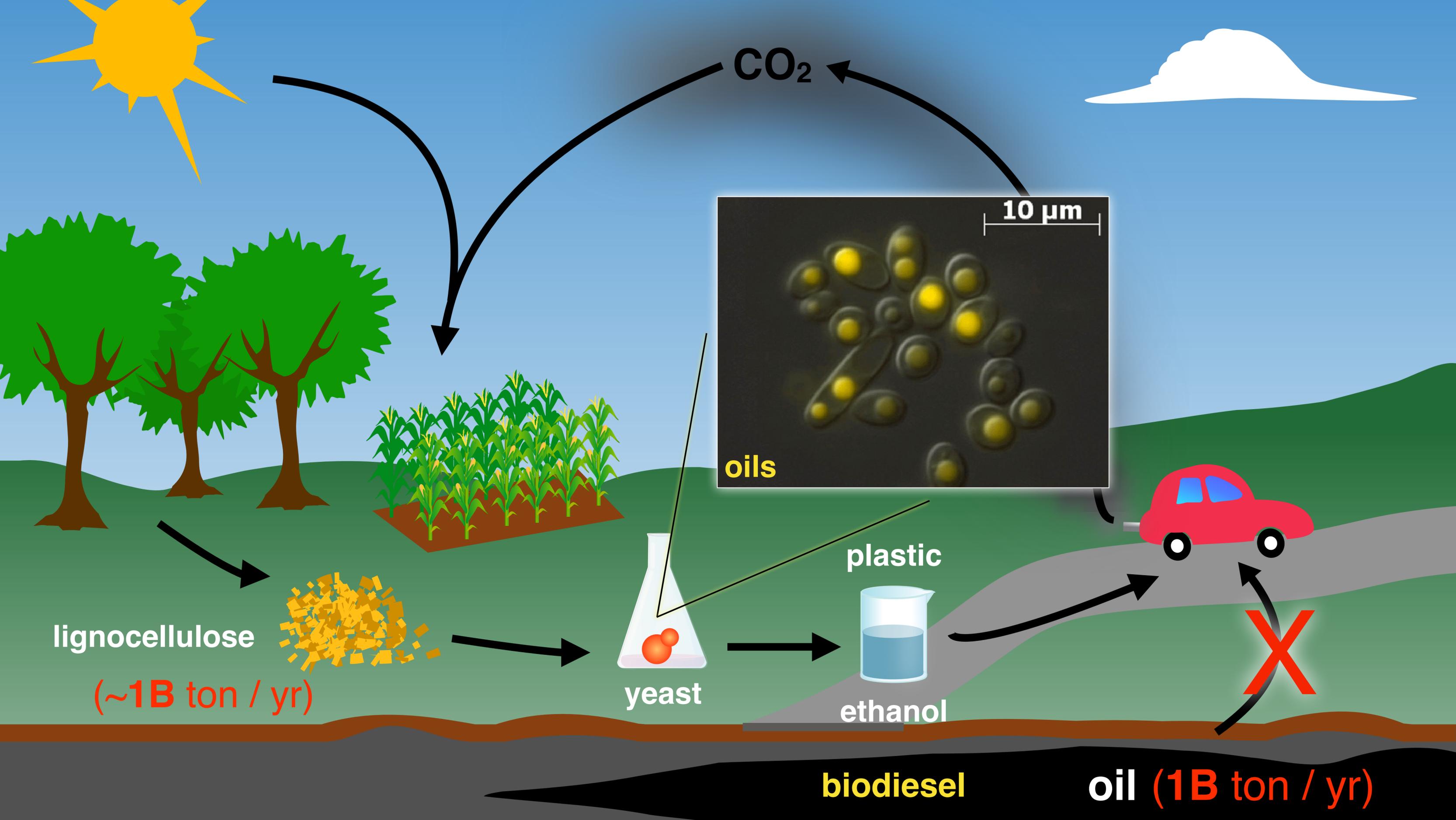
lignocellulose  
(~1B ton / yr)

yeast

plastic

ethanol

oil (1B ton / yr)



# Thanks!

Greg Stephanopoulos



Gerald Fink



## RESEARCH REPORTS



### Boosting biofuel production

Supplements help yeast survive

Left to right: Gregory Stephanopoulos of chemical engineering, Gerald Fink of biology and the Whitehead Institute, and Felix Lam of chemical engineering are developing new insights and techniques that could one day dramatically increase the amount of ethanol, butanol, and other biofuels that yeast can produce from raw materials such as corn and sugar cane.

*This research was supported in part by the MIT Energy Initiative Seed Fund Program. See page 8 for other sponsors and a publication resulting from this research.*

Photo: Stuart Darsch

Chemical engineers and biologists at MIT have found a simple way to make yeast produce more ethanol from sugars: Spike the mixture they're growing on with two common chemicals. Adding potassium and an acidity-reducing compound helps the yeast tolerate higher concentrations of the ethanol they're making without dying. Aided by those "supplements," traditionally underperforming laboratory yeast made more ethanol than did industrial strains genetically evolved for ethanol tolerance. The supplements also enabled lab yeast to tolerate higher doses of high-energy alcohols such as butanol, a direct gasoline substitute. In other "firsts," the researchers described the mechanism by which alcohols poison yeast; they defined two genes that control ethanol tolerance; and they modified those genes in lab yeast to make them out-produce the industrial strains—even without the supplements.



U.S. DEPARTMENT OF  
**ENERGY**

Energy Efficiency &  
Renewable Energy

BIOENERGY TECHNOLOGIES OFFICE